# Ultraviolet-C Light Effect on Pitaya (Stenocereus griseus) Juice

Carlos Enrique Ochoa-Velasco Depto. Ingeniería Química Alimentos y Ambiental. Universidad de las Américas Puebla. Cholula Puebla 72820, México Tel: 52-222-229-2126 E-mail: carlos.ochoavo@udlap.mx

José Ángel Guerrero-Beltrán (Corresponding autor) Depto. Ingeniería Química Alimentos y Ambiental. Universidad de las Américas Puebla. Cholula Puebla 72820, México Tel: 52-222-229-2126 E-mail: angel.guerrero@udlap.mx

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# Abstract

Pitaya (*Stenocereus griseus*) juice, obtained from fresh pitayas, was processed using a continuous ultraviolet-C (UV-C) light (57  $\mu$ W/cm<sup>2</sup>) system. Juice was processed at five flow rates (0.46, 3.28, 6.57, 16.49 and 30.33 mL/s) and five treatment times (5, 10, 15, 20, and 25 min). Fresh juice was used as control. Some physicochemical (pH, total soluble solids, color, and betalains), antioxidant (total phenolic compounds and antioxidant activity), and microbiological (aerobic mesophylls bacteria and yeasts plus molds) characteristics were assessed in fresh and UV-C processed juices. It was observed that the UV-C treatments did not affect pH and total soluble solids in juice. The total change in color ( $\Delta E$ ) increased as treatment times increased; however,  $\Delta E$  values were reduced at high flow rates. The betalains and total phenolic compounds contents were reduced as flow rates and treatment times increased; consequently, the antioxidant activity lessened in juice. A maximum reduction of 2.11 and 1.14 log cycles was observed for mesophylls and yeasts plus molds, respectively, in the UV-C light treated pitaya juice.

Keywords: Antioxidant activity, Betalains, Phenolic compounds, Pitaya juice, UV-C light

# 1. Introduction

Emerging technologies such as high hydrostatic pressure, pulsed electric fields, ultrasound, and ultraviolet light have been used to obtain food products with characteristics similar to fresh products that consumers are demanding today. The short-wave ultraviolet-C light is a physical method that does not generate chemical residues in the food and today is used for water and surfaces disinfection (Quek & Hu, 2008). The UV-C light (254nm) is easy to use for disinfection purposes of liquid foods. It has lethal effects on micro organisms such as bacteria, viruses, protozoa, yeasts, and molds (Begum, Hocking-Ailsa & Miskelly, 2009). The germicidal effect of UV-C light on micro organisms at the DNA level. The absorption of UV-C light generates electronic changes that may cause breaking of the DNA bonds; therefore, microbial cells could be compromised. The photoproducts (pyrimidine nucleotide bases), generated by the application of UV-C light, block the DNA transcription and replication; even more, inhibits cell functions that may cause the cell death (Guerrero-Beltrán & Barbosa-Cánovas, 2004).

Fruit juices are usually pasteurized in order to inactivate micro organisms and enzymes responsible for undesirable changes. However, the sensory characteristics of the pasteurized food productcould be damaged by the high temperatures used for processing (Ibarz & Barbosa-Cánovas, 2002). In 2000, the Food and Drug Administration (FDA) approved the use of UV-C light as a method for "cold pasteurization"; however, it was

also advised that the reduction of resistant pathogens should be at least of 5 log cycles (FDA, 2001) to ensure the effectiveness of the process.

The UV-C lighth as recently been used for researching in the fruit juices processing area, but this has been mainly focused on the inactivation of microorganisms (Guerrero-Beltrán & Barbosa-Cánovas, 2005; Gabriel & Nakano, 2009; Lu *et al.*, 2010), enzymes (Barka*et al.*, 2000; Guerrero-Beltrán & Barbosa-Cánovas, 2006), and changes in color (Keyser *et al.*, 2008). Few researchers have comprehensively assessed the UV-C light effects (Pala & Tocluku, 2010; Falguera *et al.*, 2011) on other fruit juice components. The UV-C light treatment of liquid fruit products has to be performed carefully since some requirements should be accomplished. Among these requirements is to sustain aturbulent flow to warranty that the whole liquid is reached by the UV-C light to deliver a microbiologically safe food product (FDA, 2000). Despite of this and due to the presence of colored compounds, organic compounds, and suspended matter, characteristic that may affect the UV-C light efficiency could be reduced (Caminiti *et al.*, 2010).

Pitaya (*Stenocereus griseus*) is a red-peel fruit produced by a species of the *cactaceae* family. The fruit is surrounded of large prickles similar to small needles. It has an oval or spherical shape. Pulp fruit possesses weetened delicate flavor; it is also juicy and has small black seeds that crunches "pleasantly" when masticate them (Ayala *et al.*, 2007). The fresh fruits weight ranges from 85.9 to 398.5g. Pulp and peel make 76-84 and 16-24%, respectively (Luna, 2006). The total soluble solids content, pH, and titratable acidity (as citric acid) range10-11, 3.9-5.0, and 14.0-0.5 %, respectively (Luna, 2006). One of themain problems of fresh pitaya is the shortshelf-life (3 to 5 days); therefore, it is important to use the appropriate technologies to increase its shelf-lifein a fresh fashion or to obtain pitaya processed products such as juices and nectars.

The aim of this study was to evaluate the physicochemical, microbiological, and antioxidant characteristics of pitaya juice treated with ultraviolet-C light.

# 2. Materials and Methods

# 2.1 Pitaya juice

"Pitaya of May" (*Stenocereus griseus*) was obtained from the municipalities of Cuauhtémoc Huitziltepec and Tepeyahualco, Puebla, Mexico. Fruits were sorted and chosen free from physical and microbiological damages. Fruits were disinfected with a solution of sodium hypochlorite (150 ppm). Pitayas were peeled and homogenize dusing a Black and Dekker domestic food processor (Towson, Maryland, USA). Afterward, juice was sieved (0.297 mm) to remove seeds and some large particles suspended in pulp.

## 2.2 UV-C light equipment

Pitaya juice was processed using an ultraviolet light system, similar to a double-walled heat exchanger, assembled at the University of the Americas Puebla. The UV-C lamps, acquired from Light Sources, Inc. (Orange, Connecticut, USA) were 303 and 15 mm in length and diameter, respectively. Lamps were of 17 W in intensity to deliver a dose of  $57\mu$ W/cm<sup>2</sup>. The UV-C flowing system hosts a volume of 430 mL into the double-walled system. The system has an inner quartz tube with an outer diameter of 2.2 cm and a stainless still external tube with an inner diameter of 4.8 cm.

## 2.3 UV-C light treatment

Pitaya juice (600 mL) was placed in a double-walled vessel which was kept at 4° Cusinga Cole Parmer Cooling Polistat Circulator system (Vernon, Illinois, USA). The juice was pumped and recirculated in the UV-C system using a 75553-71 Master Flex peristaltic pump (Vernon, Illinois, USA) at 5 different flow rates (0.46, 3.28, 6.57, 16.49, and 30.33 mL/s). The processing time of juice, for each flow rate, was 5, 10, 15, 20, and 25 minutes corresponding to doses of 0.171, 0.342, 0.513, 0.684, and 0.86 kJ/m<sup>2</sup>, respectively (Guerrero-Beltrán & Barbosa-Cánovas, 2006). Untreated juice was used as control. The UV-Clight treatment was performed in duplicate.

## 2.4 Physicochemical characteristics

Total solublesolids and pH were evaluated according to the 932.12 and 981.12 AOAC (2000) methods, respectively.

#### 2.5 Color

Ten milliliters of pitaya juice were placed in a small petri dish (6 cm in diameter and 1.5 cm in height) to measure the L (luminosity, white-black), a (green-red), and b (yellow-blue) color parameters, in the Hunter scale,

using a Gardner Colorgard<sup>®</sup> System 05 (Geretsried, Germany) colorimeter in the transmittance mode. The total change in color ( $\Delta E$ ) was calculated using the next equation:

$$\Delta E = \sqrt{(L - L_0)^2 + (a - a_0)^2 + (b - b_0)^2}$$

Where  $L_0$ ,  $a_0$ , and  $b_0$  and L, a, and b are the color parameters before and after the UV-C light treatment, respectively. The hue angle (*H*) and the chroma (*C*, intensity) color parameters were calculated using the next equations:

$$H = Tan^{-1} \left(\frac{b}{a}\right) \qquad C = \sqrt{(a)^2 + (b)^2}$$

## 2.6 Phenolic compounds

Phenolic compounds in pitaya juice were determined using the Gao, Ohlander, Jeppsson, Bjork & Traljkovski (2000) method with modifications. Two mL of distilled water were placed in an amber glass tube; then, 200  $\mu$ L of the Folin and Ciocalteu's phenol reagent (Sigma-Aldrich, Toluca, Mexico) and 100  $\mu$ L of pitaya juice were added. Mixture was totally homogenized and then incubated for 3 minutes at room temperature (25 °C). Afterward, 1 mL of 20% (p/v) Na<sub>2</sub>CO<sub>3</sub> was added and thoroughly mixed. This blend was incubated for 1 hour at room temperature in a dark environment. The absorbance was measured at 765 nm using an UV-visible spectro photometer model 2800 H (UNICO, NJ, EUA). The calculation of the content of phenolic compounds was performed using a standard curve of Gallic acid:

$$GA = \left(\frac{A-b}{m}\right) * 100$$

Where GA is the Gallic acid content (mg Gallic acid/mL), A is the absorbance of the sample, b is the intercept (-0.01), and m is the slope (4.108 abs/mg GA/mL).

#### 2.7 Betalains

Betalains were assessed according to the Stintzing, Schieber & Carle (2002) method. Pitaya juice was diluted with Mcllvaine buffer (pH 6.5, citrate-phosphate) to obtain absorption values in the range 0.9-1.0. The betanin and indicaxanthin contents were measured at wave lengths of 538 and 480 nm, respectively. The betalains content is the sum of the betanin and indicaxanthin contents and was calculated according to the next equation:

$$CB = \frac{A * DF * M * 1000}{\varepsilon * l}$$

Where *CB* is the betalains content (mg/L), *A* is the absorbance, *DF* is the dilution factor, *l* is the quartz cell pathway (1 cm),  $\varepsilon$  is the molar extinction coefficient (for  $\varepsilon_{betanin}$  is 60,000 mole/L cm, and for  $\varepsilon_{indicaxanthin}$  is 48,000 mole/L cm), and *M* is the molecular weight (550 and 308 g/mole for betanin and indicaxathin, respectively).

## 2.8 Antioxidant activity

The antioxidant activity in juice was determined according to the Kuskoski, Asuero, Parrilla, Troncoso, & Fett (2004) methodology. The ABTS<sup>+</sup> radical was formed placing 5 mL of distilled water, 3.3 mg of potassium persulfate, and 19.4 mg of the ABTS reagent into an amber glass flask. Reagents were totally mixed and let stand for 16 hours in a dark environment. Afterward, absolute ethanol was mixed with the ABTS<sup>+</sup> radical (ABTS radical solution) until reaching an absorbance of  $0.70 \pm 0.02$  at 754 nm. Eighty  $\mu$ L of pitaya juice were mixed with 3,920 L of the ABTS radical solution, totally mixed and the initial absorbance measure ( $A_i$ ). Mixture was let react for 7 minute and the final absorbance measured ( $A_f$ ). The amount of the antioxidant activity was calculated using the trolox (T) standard curve as follow:

$$UI = \frac{A_i - A_f}{A_i} * 100$$

$$UT = \frac{UI - b}{m} * 100$$

Where UT is the amount of trolox (mg T/mL), UI is the percentage of inhibition, b is the intercept(3.52) and m is the slope (371.5 abs/mg T/mL).

#### 2.9 Total counts

Aerobic mesophyll bacteria (AMB) and molds plus yeasts (MY) were counted using the standard plate count agar and the acidified (10% tartaric acid) potato dextrose agar, respectively. Petri plates for the AMB were incubated in an oven at  $35\pm2$  °C and the number of colony forming units per mL (CFU/mL) were counted in a period of 24-48 hours, while the petri plates for the ML were incubated during 5 days at  $25\pm2$  °C.

## 2.10 Mathematical modeling

The decimal reduction time  $(D_{uv})$  values were calculated using the first-order kinetics model for the survivors in pitaya juice after the UV-C light treatment (Stermer, Lasater-Smith & Brasington, 1987) as follow:

$$Log\left(\frac{N_t}{N_o}\right) = -kIt = -kF = -kD$$
$$D = F = I * t$$
$$D_{UV} = -\frac{1}{k}$$

Where  $N_t$  is the survivors microbial load (CFU/mL) after UV-C light treatment, *No* is the initial microbial load (CFU/mL), *k* is the inactivation constant rate (min<sup>-1</sup> or m<sup>2</sup>/kJ), F = D is the dose or fluence (kJ/m<sup>2</sup>), *I* is the intensity of the UV-C lamp (W/m<sup>2</sup>), and *Duv* is the decimal reduction time required to inactivate 90% of the microorganisms at constant dose or fluence.

#### 2.11 Statistical analysis

All results were evaluated by analysis of variance (ANOVA) using the Minitab 14 program (Minitab Inc., PA, USA). A p value of 0.05 was used for deciding significant differences among averages according to the Turkey's test.

#### 3. Results and Discussion

#### 3.1 Juice characteristics

Table 1 presents the physicochemical and antioxidant characteristics of fresh pitaya juice. The total soluble solids (Bx) content was lower than that reported for other types of pitayas. Luna (2006) reported values of total soluble solids in the range 10-17.25% (w/w). pH and phenolic compounds content in pitaya juice were similar to those reported for cacti fruits (Nurliyana, Syed, Mustapha, Aisyah & Kamarul, 2010) other than pitaya. Ochoa & Guerrero (2012) reported a phenolic compounds content of 42.01±8.06 mg of GA/100 mL of red prickle pear juice. Nurliyana *et al.* (2010), on the other hand, reported values of 3.75-36.12 mg of GA/100 g of pitaya pulp. The content of betalains in pitaya juice was higher than the amount reported for other red-pigmented fruits (Repo de Carrasco & Encina 2008; Castellanos & Yahia, 2008). This may probably explain the high antioxidant activity found in pitaya juice.

It can also be observed that the L color parameter indicates that juice is dark in lightness. The a value is in the red side and the b value is in the blue side of the color space chart. Pitaya juice has a dark red-purple color; this was corroborated by thehuevalue which is in the red color side of the color space chart. This could be probably because pitaya juice contains a higher amount of betanin (red-purple color) than indicaxanthin (yellow-orange color). The value of chroma (intensity) indicates that pitaya juice has an intense dark red-purple color.

## 3.2 UV-C light effect on pitaya juice

## 3.2.1 Physicochemical characteristics

Neither pH nor total soluble solids of pitaya juice were significantly affected (P>0.05) by the UV-C light at the selected flow rates and treatment times. The average values of the total soluble solids and pH, after UV-C light

treatment, were  $6.79\pm0.04$  and  $5.93\pm0.07$  %, respectively. A number of researchers have reported that UV-C light did not have effect on pH, total soluble solids, and titratable acidityof fruit juices (Noci, Riener, Walkling, Cronin, Morgan & Lying, 2008; Caminiti *et al.*, 2010; Pala & Toklucu, 2011).

# 3.2.2 Color

Table 2 presents the *L*, *a*, *b*, and  $\Delta E$  color parameters of pitaya juice treated with UV-C light for 25 minutes. It is observed that the UV-C light significantly affected (P <0.05) color parameters at different flow rates. The higher the flow rate, the lower the change in color of pitaya juice. When comparing the lowest (0.46 mL/s) and highest (30.33 mL/s) flow rates, the highest flow rate make the least change in color of pitaya juice. Guerrero-Beltrán, Welti-Chanes, & Barbosa-Cánovas (2009) reported that increasing the flow rate, for treating grape juice with UV-C light, a lower contact between UV-C light and the liquid food product may occur; therefore, juice could be less affected in its color parameters. Moreover, although data are not presented here, no effect of UV-C light was observed on the color of pitaya juice when increasing treatment time. This could be probably due to the relationship between retention time and flow rate; therefore, further damage to pigments in juice at low flow rates may occur. However, this change in color is too small to make it visible to the naked eye.

# 3.2.3 Phenolic compounds

Figure 1 presents the phenolic compounds content in UV-C light treated pitaya juice. The phenolic compounds content significantly decreased (P < 0.05) as the UV-C light processing time increased. However, no effect (P > 0.05) was observed regarding the phenolic compounds content when increasing flow rates to treat pitaya juice in the UV-C light system. The reduction of phenolic compounds could be due to the UV-Clight effect on the structure of phenolics (Koutchma, 2009). Piga, Del Caro, Pinna, & Agabbio (2003) and Bakowska, Kucharska, & Oszmianski (2003) pointed out that phenolic compounds may protect pigments, ascorbic acid, and antioxidant activity in fruits and juices against environmental injuries. Pala & Toklucu (2011) treated pomegranate juice with UV-C light in a flow range of 12.5 to 62.4 J/mL. They reported no significant differences (P > 0.05) in the phenolic compounds content of phenolic compounds in apple juice was not affected by the UV-C light when juice was treated at 5.31 and 53.10 J/cm<sup>2</sup> during 30 and 300 s, respectively. Noci *et al.* (2008) reported that phenolic compounds decreased in apple juice treated with UV-C light during 30 minutes. However, their experiment consisted in exposing 800 mL of juice, in a Pyrex plate (25 mm in diameter), to the light of an UV-C mercury lamp (30 W) placed at a distance of 30 cm on top of the juice.

# 3.2.4 Betalains

Figure 2 presents the betalains content in pitaya juice processed with UV-C light. It is observed that increasing flow rates and treatment times significantly decreased (P < 0.05) the betalains content; the betalains reduction ranged 3.89-20.21%. This betalains reduction could be probably due to the photons producedby UV-C light. Photons could be absorbed by organic molecules such as betalains which possess conjugated bonds and aromatic rings responsible for color (Woo, Ngou, Ngo, Soong & Tang, 2011; Koutchma, 2009). Guerrero-Beltrán *et al.* (2009) reported that long periods of UV-C light to treat fruit products may undergo discoloration reactions of the pigments. Bakowska *et al.* (2003) reported that phenolic compounds may act as inhibitors of the degradation of anthocyanins during the exposition of fruit and, or vegetable products to the UV-C light. Likely, phenolic compounds might protect pitaya betalains for short periods of time. Pala & Tocluku (2011) reported that the anthocyanins content of pomegranate juice decreased gradually with increasing the UV-C light dose; they reported a decrease in the anthocyanin content of 1.8, 3.9, and 8.4% at UV-C exposure doses of 12.5, 34.4, and 62.4 J/mL, respectively.

# 3.2.5 Antioxidant activity

Figure 3 presents antioxidant activity in pitaya juice treated with UV-C light. Substantial antioxidant activity was expected since phenolic compounds and betalains may function as antioxidants. The antioxidant activity was reduced by the UV-C light at both treatment times and flow rates. The antioxidant activity creaseds ignificantly (P < 0.05) in juice as processing time increased. However, no significant differences (P > 0.05) were observed in the antioxidant activity content within flow rates. Caminiti *et al.* (2010) pointed out that the antioxidant activity decreased as the UV-C dose was increased for processing apple juice; however, Pala & Tocluku (2011) reported no significant difference in the antioxidant activity content in fresh and UV-C light treated pomegranate juice using different doses.

It has been reported that UV-C light may affect compounds with high antioxidant activity. For example, Sabliov, Fronczek, Astete, Khachaturyan, Khachatryan & Leonardi (2009) reported that the initial content of  $\alpha$ -tocopherol,

dissolved in hexane or methanol, was significantly reduced when increasing the UV-C light treatment. Cvetkovic, Markovic, Cvetkovic & Radovanovic (2011), on the other hand, reported that UV-A, UV-B, and UV-C may reduce the antioxidant activity of phenolic compounds such as rutin and quercentin; the damage of these compounds was increased as the UV-C exposure time was increased. They pointed out that the reduction of compounds with antioxidant activity could be due to the combination of UV-C light and oxygen.

## 3.2.6 Microbial inactivation

Figures 4 and 5 present the aerobic mesophyll bacteria and molds plus yeasts inthe UV-C light treated pitaya juice, respectively. The number of colony forming units per milliliter, in both types of microorganisms, decreased as the flow rate and treatment time increased. Pitaya juice possesses a dark red-purple color and this could avoid the penetration of light; however, turbulence regime, formed by increasing the speed, is enough for making all the liquid be in contact with the UV-C light and obtain a greater microbial inactivation (Li, Deng & Nyung, 2010). Caminiti *et al.* (2010) and Guerrero-Beltrán & Barbosa-Cánovas (2005) pointed out that the transparency and soluble and insoluble solids of the liquid food product, or medium, are critical factors in the microbial inactivation with UV-C light; both color and turbidity may block the pathway of light and prevent to microorganisms to be reached by the UV-C light. Li *et al.* (2010) reported that the penetration of UV-C light in the product is of utmost importance since better results, regarding microbial inactivation, are obtained on the surface of the food. Recent investigations have been performed to explore the effect of UV-C light in the inactivation of microorganisms inoculated in different fruit juices. Guerrero-Beltrán & Barbosa-Cánovas (2005) and Guerrero *et al.* (2009) reported a log reduction of 1.34  $\pm$  0.35 and 1.3 for *S. cerevisiae* inoculated in apple and grape juice, respectively. These results agree with those obtained in this research for yeasts plus molds in pitaya juice.

Table 3 presents the microbial counts and the log reductions for AMB and MY for the selected flow rates after 25 minutes of UV-C light treatment of pitaya juice. It can be observed that the AMB load is higher than the MY load; however, both types of microorganisms were reduced by the UV-C light. Li et al. (2010) pointed out that the efficiency of the UV-C light effect on microorganisms is a function of the initial microbial load. It is also observed that the higher the flow rate, the higher the inactivation effect for both types of microorganisms. The greater inactivation for AMB (2.11 log cycles reductions) and MY (1.14 log cycles reductions) was reached after 25 min of UV-C light treatment. This could be because bacterial cells are smaller than molds and yeasts; therefore, bacteria could be more easily reached and compromised by the UV-C light (Montgometry, 1985). In addition, bacterial cells are constituted by large levels of pyrimidine in the DNA; this may increase the probability of generating more cross-linkages between thymine and cytosine (Torkamani & Niakousari, 2011) by the UV-C light. Oteiza, Giannuzzi & Zaritzky (2010) reported that orange juice, inoculated with S. cerevisiae, decreased the ability of UV-C light for inactivating five E. coli O157: H7 strains. They concluded that yeasts, due to their size, increased the coefficient of absorption of the UV-C light. Therefore, high doses might be requiredto obtainthe same inactivation effect on the E. coli trains. López-Malo, Guerrero, Santiesteban & Alzamora (2005) processed apple juice, inoculated with L. monocytogenes and S. cerevisiae, with UV-C light. They reported that bacteria are more sensitive (> 5 log reduction) than yeasts (4 log reductions) to UV-C light.

Table 4 presents the first-order kinetics parameters for the AMB and MY inactivation in pitaya juice treated with UV-C light. The first order kinetics representation provided good fittings of the inactivation ( $R^2 > 0.95$ ) of AMB. This representation is not entirely appropriate for MY.  $D_{uv}$  values decreased as flow rates increased, this means that the inactivation of the microbial load increased as flow rate increased.Guerrero-Beltrán & Barbosa-Cánovas (2005) reported  $D_{uv}$  values of 5.9, 7.0, and 22.4 minutes for *E. coli*, *L. innocua*, and *S. cerevisiae* inactivation in apple juice treated with UV-C light at a flow rate of 9.13 mL/s. Torkamani & Niakosari (2011), on the other hand, reported  $D_{uv}$  values of 0.82 and 1.05 kJ/m<sup>2</sup> for total counts and yeasts plus molds, respectively, in not inoculated orange juice. Both sets of values are similar to those found in this study for same types of microorganisms.

# 4. Conclusions

The use of UV-C light to treat pitaya juice is a feasible alternative to deliver a microbiologically safe juice to consumers. Despite of not accomplish the 5 log reductions required by the FDA, pitaya juice maintains its quality because other physicochemical attributes remained barely unchanged. The decline in the quality attributes of pitaya juice is mainly a function of time or dose of treatment with the UV-C light. Maximum log cycle reductions of 2.11 and 1.14 for AMB and MY, respectively, were obtained after 25 minutes of UV-C light treatment at a flow rate of 30.33 mL/s. Under these conditions of treatment, changes in color were practically non-existent compared to fresh juice. The phenolic compounds, antioxidant activity, and betalains content

decreased by 13.27, 3.55, and 26.02%, respectively, in comparison to the content in fresh juice. It is important to take into account that the actual effect of the UV-C light on microbial inactivation in pitaya juice may depend on the UV-C dose used to treat juice as well as the type and amount of microorganisms.

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Table 1. Physicochemical and antiox	idant characteristics of fre	esh pitaya ( <i>Stenocer</i>	eus griseus) juice
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Characteristic	Quantity		
Total soluble solids (%)		6.75±0.08	
pH		5.91±0.08	
	L	30.52±0.10	
	a	61.91±0.65	
Color	b	18.78±0.06	
	Hue	16.87±0.08	
	Chroma	64.13±0.13	
Phenolic compounds (mg GA/100 mL)		39.00±0.09	
Datalaina	Betanin (mg/L)	61.05±2.22	
Detalalits	Indicaxanthin (mg/L)	52.48±0.18	
Antioxidant activity (mg de T/100 mL)		100.60±1.17	

Table 2. Color parameters of pitaya juice after 25 min of UV-C light treatement<sup>1</sup>

	Color parameters					
Flow rate (mL/s)	L	а	Ь	ΔΕ		
0.00	$30.52{\pm}0.10_{a}$	$61.91 \pm 0.65_{a}$	18.78±0.06 <sub>a</sub>	0.00 <sub>a</sub>		
0.46	$28.26 \pm 0.14_{b}$	$58.16 \pm 0.14_{b}$	$17.50\pm0.07_{b}$	3.38 <sub>b</sub>		
3.28	$29.27 \pm 0.08_{c}$	$59.88 \pm 0.16_{c}$	$18.21 \pm 0.06_{c}$	1.63 <sub>c</sub>		
6.57	29.47±0.10c	59.67±0.12c	$18.31 \pm 0.08_{c}$	1.70 <sub>c</sub>		
16.49	$28.57{\pm}0.08_d$	$58.68 {\pm} 0.09_d$	$17.63 \pm 0.08_{d}$	2.88 <sub>d</sub>		
30.33	$30.43 \pm 0.04_{a}$	$62.22{\pm}0.14_a$	$18.67 \pm 0.04_{a}$	0.87 <sub>a</sub>		

<sup>1</sup>: Same litters within columns indicate no significant differences (P > 0.05).

Table 3. Log cycles reduction for aerobic mesophyll bacteria and yeasts plus molds in UV-C light treated pitaya juice for 25 minutes

	Aerobic meso	ophyll bacteria	Yeasts plus molds		
Flow rate (mL/s)	(CFU/mL) (x10 <sup>-3</sup> )	log (N/No)	(CFU/mL)	log ( <i>N</i> / <i>No</i> )	
0.00	38.0±0.079	0	850±21	0	
0.46	18.0±0.079	-0.33	630±57	-0.12	
3.28	12.0±0.078	-0.52	460±35	-0.27	
6.57	9.6±0.020	-0.6	430±14	-0.29	
16.49	$1.0\pm0.014$	-1.58	165±21	-0.71	
30.33	$0.5 \pm 0.028$	-2.11	62±11	-1.14	

Aerobic mesophyll bacteria				Yeast	Yeasts plus molds					
Flow rate	m	h	$\mathbf{P}^2$	$D_{uv}$	$D_{uv}$	m	h	$\mathbf{P}^2$	$D_{uv}$	$D_{uv}$
(mL/s)	(1/min)	U	К	(min) $(kJ/m^2)$ (1/min)	ĸ	(min)	$(kJ/m^2)$			
0										
0.46	-0.013	0.005	0.97	80.1	2.74	-0.003	-0.042	0.50	375	12.8
3.28	-0.022	0.022	0.98	46.7	1.59	-0.012	0.019	0.93	87.1	2.98
6.57	-0.025	0.031	0.99	40.8	1.39	-0.011	0.049	0.75	101	3.45
16.49	-0.068	0.175	0.98	14.9	0.51	-0.029	0.014	0.89	35.9	1.23
30.33	-0.073	-0.130	0.96	13.9	0.47	-0.04	-0.26	0.91	23.5	0.80

Table 4. First order kinetics data from the aerobic mesophyll bacteria and yeasts plus molds inactivation in UV-C light treated pitaya juice



Figure 1. Phenolic compounds in UV-Clight treated pitaya juice



Figure 2. Betalains content in UV-C light treated pitaya juice



Figure 3. Antioxidant activity in UV-C light treated pitaya juice



Figure 4. Aerobic mesophyll bacteria in UV-C light treated pitaya juice



Figure 5. Molds plus yeasts in UV-C light treated pitaya juice