Technological and Nutritional Aspects of Incorporating Jamun (*Syzygium cumini* (L.) Skeels) Fruit Extract into Yoghurt

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Abstract

The study aimed to evaluate the technological suitability of incorporating the jamun extracts into Yoghurt. The light focused on the effect of the extraction method and rate of addition on the flavonols profiles, antioxidant activity and sensorial characteristics of the final Yoghurt product. Jamun fruit was subjected to either mechanical cold extraction or steam extraction and introduced to milk at rates of 5 and 10%. The results indicated that the extraction technique had no effect on the values of protein, fat, ash and titratable acidity. The steam extraction led to increase the total solids, pH, total hydrolysable tannins, antioxidant activity, color, flavor and overall sensorial acceptability of Yoghurt. While the cold mechanical extraction led to increase the total solids, protein, fat, appearance and thickness in a concentration depending way, as well as to increase all the detected flavonols, tannins and antioxitant power indicators. The 5% juice containing Yoghurt was distinguished with the highest scores of color, flavor, taste, smell and overall acceptability. Jamun fruit may be a promising source for fortifying Yoghurt with flavonols and enhancing its antioxidant power.

Keywords: blackberry, java plum, scavenging, tannins

1. Introduction

The consumer demand for functional food and dairy products is in a continuous increase due to the health benefits they reflect, leading to the research and development of different functional food products (Abdel-Hamid et al., 2020). Moreover, the concern with the immunity boosting foods has been increased many folds with the outbreak of covid-19. Jamun fruit or Syzygium cumini is one of the family Myrtaceae well-known plants. It is commonly known as Indian Blackberry, Jambul, Black Plum and Java Plum. It is a health stimulating fruit and contains many phytochemicals incorporated in different parts of the tree including leaf, seed, and pulp. Due to its richness of these phytochemicals, it is used to treat several diseases in tropical Aisha such as allergies, viral infection, gastric ulcer, cancer and diabetes mellitus (Chaudhary & Mukhopadhyay, 2012; Joshi, Paudel & Upreti, 2019). In the traditional medicine, jamun was known to be anti-stomach pain, anti-scorbutic, carminative, and diuretic. Moreover, black plum vinegar is good to reduce enlargement of spleen and diarrhea. Jamun fruit with its polyphenol compounds has attracted the attention of the food manufacturers, as a potential material for the fortification and development of several functional food products and meals. In Europe, jamun was known as a delicacy and sold in a dried form (Singh et al., 2016). It was used also for the preparation of bread, pulp, jelly and wine (Branco et al., 2016; Kapoor, Ranote & Sharma, 2015). On the other hand, Yoghurt is commonly consumed as a functional food due to its nutritional value besides its richness in the bioactive peptides with antioxidant power that are produced by the action of the lactic acid bacteria fermentation (Muniandy, Shori & Baba, 2016). Yoghurt is a very good source of protein, fat, phosphorus, thiamin, riboflavin, vitamin B12, niacin, calcium, magnesium, and zinc. Consumption of Yoghurt has considered to be beneficial to digestion and overall health (Morelli & Yogurt, 2014). It causes a decrease in the stomach pH, which in turn lowers the potential risk of pathogen transition and ameliorates the improper effects of the reduced gastric juice secretion and hence any acidic supplementation to Yoghurt is beneficial to enhance these effects. Yogurt is always flavored by application of natural substances or artificial flavoring preparations (Co ison et al., 2005). Fruit extracts: specially

strawberries, blueberries and raspberries are widely used to provide Yoghurt with natural colors, flavors and bioactive compounds, those berries are commonly considered as functional components in the food and dairy manufacture (Wallace & Giusti, 2008). The impact of different functional fruit extracts on the sensorial, physicochemical, antioxidant and nutritional properties of fruit yoghurt have been evaluated (Kim et al., 2009; Oh &Kang, 2015; Nguyen & Hwang, 2016), but jamun extracts have not been evaluated in that concern. For that in view, the present study was conducted to evaluate the technological suitability of incorporating the jamun cold and steam extracts into fruit Yoghurt with the light on the effect of the extraction method and rate of addition on the flavanols' profiles, antioxidant activity and sensorial characteristics of the final Yoghurt product.

2. Materials and Methods

2.1 Materials

Skimmed milk powder was obtained from Fonterra brands, Ltd – New Zealand (3.8 % moisture, 34% protein, 1.5% fat and 8.3% ash). Cow's fresh cream (56.5 % fat, 0.371 % ash and 1.71 % protein) was obtained from Shahencoland company for dairy products, Foah, Egypt. Concentrated lyophilized mixed yoghurt starter culture (YC-183) contains *Str. thermophilus* and *Lb. delbrueckii* ssp. *bulgaricus*, 1:1, were obtained from Chr. Hansen Lab., Denmark. Sucrose was obtained from Savola, Egypt. Frozen Brazilian Jamun (*Syzygium cumini* L. Skeels) fruit was provided with confirmed identity by Diamond Company for food industries and import, El-Obour, Egypt.

2.2 Experimental Procedure

2.2.1 Preparation of the Jamun Juices

The frozen Jamun fruits were allowed to thaw at the room temperature for about 2 h and washed using running water. The flesh with skins of fruits was separated manually from the seed. The prepared flesh fruit parts were divided to two portions and then the fruit juice was either mechanically extracted at 5 $^{\circ}$ C by blending in a processing blender at 1500 rpm or extracted by steam in a stainless steel proper steamer pan at 85±5 $^{\circ}$ C for 2 h. the extracted two forms of juice were filtered with clean cloth and packed in sterile plastic bags and stored at -18 $^{\circ}$ C until use.

2.2.2 Manufacturing of Jamun Fruit Yoghurt

The method of Kulshrestha & Jha, 2006 was applied with some modifications as the powdered Skim milk was reconstituted in water at 40 °C and standardized with the fresh cream to obtain full cream milk containing 14% total solids and 3.5% fat. The standardized yoghurt milk was heated to 65 °C, and then homogenized at 2500 and 500 psi for the first and second stage respectively. Sucrose was added at a rate of 6% and mixed well. The sweetened milk was heated at 90 °C for 8 min., cooled to 42°C, inoculated with 2% of freshly activated YC starter culture and divided to 5 equal portions; the cold extracted Jamun juice was added to the 3rd and 4th portions at rates of 5 and 10% respectively. The steam extracted Jamun juice. All the experimental samples were packaged in 150 ml poly stearin cups with closers. The filled cups were incubated at 40°C for 4 h, and then kept at 5 °C until use.

2.2.3 Preparation of Fruit Yoghurt Samples for the Determination of Flavonols and Hydrolysable Tannins Profiles

A portion of 100 g of each Yoghurt sample was subjected to three times of extraction to recover the polyphenols. The samples were dissolved in 50 ml of a solvent mixture of methanol, water, and formic acid (50:48.5:1.5, v/v resp.), subjected to an ultrasonic agitation for 10 min, and were then centrifuged at 2.500g and 5 % /10 min. The obtained supernatants were then combined into a single extract, roto-evaporated at 35 % and lyophilized. aliquots of 3 ml of Yoghurt samples extracts were subjected to solid-phase extraction on Bond Elut Plexa PCX cartridges (Agilent; 6 cm3, 500 mg of adsorbent) to remove interfering anthocyanins (Castillo-Mu ñoz et al., 2009). These anthocyanin-free extracts were filtered through 0.20-µm membrane and injected directly on the chromatographic column (20 µL). The total concentrations of gallotannins and ellagitannins were estimated after acidic hydrolysis of anthocyanin-free extracts as described by Peng, Scalbert & Monties, 1991. The hydrolyzed extracts were then homogenized, filtered, and injected directly on the chromatographic column (20 µL).

2.3 Analytical Methods

2.3.1 Determination of the Physicochemical Properties of the Fruit Yoghurt

Total nitrogen, dry matter, fat, ash, and titratable acidity contents were determined according to AOAC, 2007. The pH values were measured using a microprocessor pH meter (model 8417, Hanna Instruments, Singapore) at

20 °C.

2.3.2 Determination of Flavonols, and Hydrolysable Tannins in the Fruit Yoghurt using HPLC-DAD-ESIMSⁿ

The method of Tavares et al., 2016 was followed to analyze the phenolic extracts of fruit Yoghurt samples. HPLC separation, identification, and quantitation were performed using an Agilent 1100 Series HPLC system equipped with DAD (G1315B) and an LC/MSD Trap VL (G2445C VL) ESI-MSn detectors, coupled to an Agilent ChemStation (version B.01.03) data-processing station. The mass spectra data were processed using the Agilent LC/MS Trap software (version 5.3). A Zorbax Eclipse XDB reversed-phase-C18 column (2.1×150 mm, 3.5μ m particle) was used and maintained at 40 °C with a 0.19 mL/min flow rate. The identification was depended on spectroscopic data of UV–Vis and MS/MS for standards or data from previous research reports of Castillo-Mu ñoz et al., 2009 and Tavares et al., 2016. Quantitative determination of polyphenols was carried out by extraction of the DAD chromatograms at 360 nm for flavonols and by the external standard method. All calibration standard solutions were prepared by dilution of an appropriate aliquot of the stock solution of M3glc (from 5 to 100 mg/L) for flavonols. All calibration curves of components showed good linearity. The total concentrations of the flavonols were expressed as mg of M3glc equivalents, per 1 L or kg of sample.

The method of Peng, Scalbert & Monties, 1991 was applied for the determination of hydrolysable tannins using HPLC. The identification of hydrolysable tannins was based on spectroscopic data (UV–Vis and MS/MS) for standards. For quantitation, DAD chromatograms were extracted at 280 nm and calibration curves of gallic (from 5 to 100 mg/L) and ellagic acids (from 5 to 40 mg/L) were prepared, as described for flavonols. The total concentrations of gallotannins and ellagitannins were expressed in mg of gallic acid equivalents and in mg of ellagic acid equivalents, respectively, per 1 L of sample.

2.3.3 Determination of the Antioxidation Power Indicators of the Jamun Fruit Yoghurt

The DPPH radical scavenging activities of plain and fruit yogurt samples were determined according to the method of Cheung, Cheung & Ooi, 2003 with some modifications as follows: 192 µL of 50 µM DPPH was added with mixing to 48 μ L of diluted sample. The mixture was covered and kept in the dark at room temperature for 30 min. The control consisted of 48 µL distilled water in 192 µL of 50 µM DPPH for the ascorbic acid standard or 48 µL of 94% ethanol in 192 µL of 50 µM DPPH for the test samples. Decolorization of DPPH was measured at 517 nm with a micro-plate reader. The DPPH radical scavenging activity was calculated according to the following equation: Inhibition (%) = $I_{control}$ - $I_{sample}/I_{control} \times 100$, where $I_{control}$ is the absorbance of the control reaction which contains all reagents except the test compound, and Isample is the absorbance of the test compound. The ABTS radical scavenging activities of plain and fruit yogurt samples were determined using the method of Re et al., 1999 with slight modifications. ABTS was dissolved in distilled water to a concentration of 7 mM. ABTS radical cations were produced by reacting the ABTS stock solution with 2.45 mM K₂S₂O₈ (in a 2:1 ratio) in the dark, covered for 24 h before use. The ABTS reagent was diluted with 94% ethanol to the appropriate absorbance (0.17 \pm 0.03), which was measured at 734 nm. The ABTS reagent (950 μ L) was mixed with 50 μ L of the indicated concentrations of test samples. The mixture was covered and kept in the dark at room temperature for 10 min. The absorbance at 734 nm was measured with a micro-plate reader. Each sample was measured in triplicate, and percent inhibition was calculated using the same equation of calculating the DPPH. To assay the reducing power, the ferric reducing power of yogurt samples was determined using the method of Oyaizu, 1986 with slight modifications. Various concentrations of samples (0.25 mL) were mixed with 0.25 mL of 0.2 M phosphate buffer (pH 6.6) and 0.25 mL of 1% (w/v) K₃[Fe (CN)₆]. The mixture was incubated in a 50°C water bath for 20 min, and their action was stopped by adding 0.25 mL of 10% (w/v) trichloroacetic acid solution. The solution was then centrifuged at 3000 rpm for 10 min. The supernatant (0.5 mL) was mixed with 0.5 mL of distilled water and 0.1 mL of 0.1% (w/v) FeCl₃ for 10 min. The reducing power was determined by measuring the absorbance at 700 nm with a microplate reader. A standard curve of ascorbic acid was generated with concentrations ranging from 0 to 200 µg/mL.

2.3.4 Sensory Evaluation of the Jamun Fruit Yoghurt

The samples were evaluated for sensorial characteristics after 3 days from manufacturing by 5 panellists who are working in the dairy manufacture and familiar with the Yoghurt acceptability criteria; using nine-point hedonic scales as described by Obi, Henshaw & Atanda, 2010.

2.3.5 Statistical Analysis

The obtained data were statistically analyzed through the ANOVA procedure and Duncan's multiple range tests according to statistical analyses system user's guide; SAS/STAT[®] 9.1. SAS, Inst Inc Ed Cary NC USA (SAS, 2004). Design was completely random with replications.

3. Results and Discussion

3.1 Physicochemical Properties of the Jamun Fruit Yoghurt

Data shown in Table (1) revealed that, all studied factors had significant effect on the total solids of the jamun Yoghurt. The steam extracted juice addition to milk led to significantly increase the total solids of Yoghurt as compared to the addition of the juice extracted by the cold blending method. That could be due to the strong effect of stem in breaking the fruit cells leading to increasing the extraction efficiency (Chen & Chen, 2011). This phenomenon was supported by the findings of Astuti et al., 2018 who concluded that the total soluble solids increase by increasing the heat treatment degree. The total solids contents were decreased by increasing the percentage of jamun fruit juice addition, possible due to the lower total solids' contents of the extracted juices than that of the standardized yoghurt milk. The protein and fat contents of the yoghurt samples containing the cold extracted jamun juice were statistically like those having the steam extracted juice. Likewise, the increase in the rate of jamun juice addition to the milk led to decrease the protein and fat of the resultant Yoghurt significantly. That could be ascribed essentially to the extremely low protein and fat contents in the jamun fruit flesh (Ghosh et al., 2016). Regarding the ash contents, the statistical analysis indicated that the two studied factors had no significant impact on the ash contents of the experimental samples. The cold extracted jamun fruit Yoghurt samples obtained higher titratable acidity than the steam extracted fruit yoghurt ones, possibly due to the damage occurred to the organic acids during steam heating. The addition of jamun fruit juice to milk led to increase the titratable acidity of the final yoghurt regardless the method of juice extraction and the rate of addition as Yoghurt with 5% juice had statistically similar titratable acidity to that made with 10% juice. That could be attributed to that the applied method for acidity determination was customized to measure the lactic acid. Likely, the cold extracted jamun fruit Yoghurt samples were distinguished with lower pH values than those of the other method of juice extraction. The fruit yoghurt having 10% juice obtained statistically similar pH to that having 5% juice, and lower than the control Yoghurt. That could be ascribed essentially to the strong buffering capacity of milk proteins which in turn kept the pH of Yoghurt unchanged when the rate of juice addition was increased from 5 to 10%. The round figures of these results are in accordance with the finding of Rahman et al., 2020.

Table 1. Ls means values for method of Jamun fruit juice extraction, percent of juice addition and method of Jamun fruit juice extraction x percent of juice addition on the physicochemical properties of the Jamun fruit yoghurt

Source of variation	Total	Protein (%)	Fat (%)	Ash (%)	Titratable	pH		
	Solids (%)				acidity (%)			
Method of Jamun fruit juice extraction (M)								
Cold blending at 1500 rpm	18.55 ^b	3.50 ^a	3.27 ^a	0.566^{a}	0.86^{a}	4.20^{b}		
Stem at 85±5 °C for 2 h.	18.77^{a}	3.50 ^a	3.26 ^a	0.573 ^a	0.83 ^a	4.40^{a}		
P value	0.0019	1.0000	0.9094	0.9004	0.6026	0.0045		
Percentage of Jamun fruit ex	stract addition	n (P)						
0 %(control)	19.00 ^a	3.69 ^a	3.43 ^a	0.584^{a}	0.67^{b}	4.47^{a}		
5%	18.64 ^b	3.50 ^b	3.26 ^b	0.570^{a}	0.93 ^a	4.26 ^b		
10%	18.34 ^c	3.33 ^c	3.10 ^c	0.556^{a}	0.94 ^a	4.19 ^b		
P value	<.0001	0.0009	0.0021	0.9241	0.0031	0.0048		
M X P								
Control	19.00	3.69	3.43	0.584	0.67	4.47		
Cold blending x 5%	18.53	3.50	3.27	0.566	0.95	4.15		
Cold blending x 10%	18.12	3.33	3.11	0.549	0.97	4.00		
Steam extraction x 5%	18.76	3.50	3.26	0.574	0.91	4.37		
Steam extraction x 10%	18.57	3.33	3.10	0.563	0.92	4.38		
P value	0.0245	1.0000	0.9966	0.9950	0.9299	0.0564		

Values with different letters (a-c) within the same column are significantly different at P<0.05.

3.2 Flavonols Spectral Characterization of the Jamun Fruit Yoghurt

Data shown in Table (2) elucidated that the identified flavonols in the experimental jamun fruit Yoghurt samples were the Bring tri-substituted aglycones: laricitrin (L), myricetin (M) as well as syringetin (S) with their derivatives; 3-galactoside, 3-glucoside, 3-rhamnoside, 3-pentoside and 3-glucuronide. Moreover, free syringetin and myricetin were detected in all the Yoghurt samples, while free laricitrin was detected only in the steam

extracted jamun Yoghurt. The flavonols from quercetin, isorhamnetin or kaempferol were not detected in all samples. The cold extracted jamun fruit Yoghurt samples were distinguished with higher amounts of the flavonols: M3glcU, M3gal, M3glc, M3rhm, M3pent, L3gal, S3gal and free S than the steam extracted jamun fruit Yoghurt which gained the highest levels of L3glc, S3glc, Free M and Free L. The latter free laricitrin flavonol was considerably detected only when the steam extraction was applied to prepare the jamun fruitextract, possible due to role of heating in the degradation of the cell walls and the hydrolysis of flavonol glycosides leading to releasing aglycones. Similar observations were reported by Teleszko, Nowicka & Wojdyło, 2016 who studied the factors affecting the polyphenols content in strawberry cloudy juices. Nevertheless, the application of cold blending at 1500 rpm to jamun fruit led to attain the highest level in mg/L of Yoghurt total flavonols when compared to the steam extraction method. The statistical analysis confirmed that the flavonols contents of the fruit Yoghurt were increased by increasing the percentage of added juice, regardless the method of juice extraction. It's noteworthy to mention that the average total favonols content of the cold extracted jamun fruit yoghurt was about 5.5 % of the proportion of total falvonols in the jamun fruit itself.

Table 2. Ls means values for method of Jamun fruit juice extraction, percent of juice addition and method of Jamun fruit juice extraction x percent of juice addition on the flavonols content by HPLC-DAD-ESI-MS/MS (negative ionization mode) spectral characterization of the Jamun fruit yoghurt

Source	Flavonol*	M3glcU	M3gal	M3glc	M3rhm	M3pent	L3gal	L3glc	S3gal	S3glc	Free	Free	Free	Total
of	Molecular	493;317	479;317	479;317	463;317	449;317	493;331	493;331	507;345	507;345	M 317	L 331	S 345	flavonols
variation	ion; product													mg/L
	ions (m/z)													
Method of	Jamun fruit juice	extraction (!	(Iv											
Cold blend	ling	0.198a	0.192a	2.368a	0.350a	0.310a	0.093a	0.232b	0.116a	0.090b	0.979b	0.017b	0.038a	2.18a
at 1500 rpr	n													
Stem at 85	±5 ℃	0.127b	0.088b	2.193b	0.310b	0.232b	0.089b	0.255a	0.091b	0.095a	1.379a	0.046a	0.031b	1.72b
for 2 h.														
P value		<.0001	<.0001	<.0001	<.0001	<.0001	0.0004	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001
Percentage	of Jamun fruit ex	xtract additio	n (P)											
0 %(contro	ol)	0.00c	0.00c	0.000c	0.000c	0.00c	0.00c	0.000c						
5%		0.182b	0.142b	2.221b	0.337b	0.291b	0.102b	0.247b	0.098b	0.083b	1.169b	0.023b	0.026b	1.945b
10%		0.306a	0.279a	4.622a	0.653a	0.524a	0.171a	0.483a	0.213a	0.194a	2.370a	0.074a	0.079a	3.917a
P value		<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001
M X P														
Control		0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Cold blend	ing x 5%	0.230	0.206	2.382	0.364	0.331	0.113	0.252	0.098	0.072	0.961	0.000	0.020	2.163
Cold blend	ing x 10%	0.365	0.372	4.724	0.687	0.601	0.166	0.444	0.252	0.199	1.978	0.056	0.096	4.381
Steam extr	action x 5%	0.134	0.079	2.060	0.310	0.251	0.091	0.243	0.098	0.095	1.377	0.046	0.032	1.727
Steam extr	action x 10%	0.247	0.186	4.520	0.620	0.447	0.176	0.522	0.175	0.190	2.762	0.092	0.062	3.453
P value		<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001	<.0001

*M: myricetin; L: laricitrin; S: syringetin; glcU: glucuronide, gal: galactoside; glc: glucoside; rhm: rhamnoside; pent: pentoside.

Values with different letters (a-c) within the same column are significantly different at P<0.05.

3.3 Hydrolysable Tannins Spectral Characterization of the Jamun Fruit Yoghurt

The results of Table (3) indicated that the application of steam at 85 °C for 2 h in jamun juice extraction caused significant increments in the total gallotannins contents, which are referring to the total sum of gallic acid and its methyl ester in the jamun Yoghurt. That could be attributed to the covalent bonds' cleavage occurred through heating (Branco et al., 2016). Although the total ellagic acid contents took the same trends as the gallotannins, it was observed in relatively low concentrations. That could be due to that ellagitannins are very polar in case of the presence of many hydroxyl groups and hence very easy to be liberated (Nuengchamnong & Ingkaninan, 2009). The extraction of juice with steam was associated with increasing the total ellagic acid in jamun Yoghurt in comparison with the cold extraction technique, possibly due to the enhanced release of ellagic acid from ellagitannins by hydrolysis during heating. These phenomena are in accordance with those reported by Fracassetti et al., 2013. The depending on cold blending for jamun fruit juice extraction led to decrease the levels of total vanoleic acid and total ellagitannins in the final fruit Yoghurt. Nevertheless, the levels of all determined hydrolysable tannins were increased by increasing the rate of jamun juice addition to the Yoghurt standardized milk. The total gallotannins were within 3.45- 14.73 mg/L; the highest value was for the steam extracted jamun Yoghurt with 10% juice. The lowest level of the hydrolysable tannins was for total vanoleic acid which is referring to the total sum of valoneic acid dilactone methyl ester and its isomer; it was ranging from 0.25 to 6.20

mg/L.

Table 3. Ls means values for method of Jamun fruit juice extraction, percent of juice addition and method of Jamun fruit juice extraction x percent of juice addition on total hydrolysable tannins (as gallic acid for gallotannins; as ellagic acid for ellagitannins) of the Jamun fruit yoghurt

Source of variation	Total gallotannins	Total ellagic acid	Total vanoleic acid	Total ellagitannins
	(mg/L)	(mg/L)	(mg/L)	(mg/L)
Method of Jamun fruit juice	extraction (M)			
Cold blending at 1500 rpm	3.521 ^b	0.398 ^b	0.275 ^b	0.659 ^b
Stem at 85 ± 5 °C for 2 h.	7.474 ^a	2.959 ^a	3.186 ^a	5.947 ^a
P value	<.0001	<.0001	<.0001	<.0001
Percentage of Jamun fruit ex	stract addition (P)			
0 %(control)	0.000°	0.000°	0.000°	0.000°
5%	5.571 ^b	1.637 ^b	1.803 ^b	3.143 ^b
10%	10.922 ^a	3.399 ^a	3.390 ^a	6.767 ^a
P value	<.0001	<.0001	<.0001	<.0001
МХР				
Control	0.000	0.000	0.000	0.000
Cold blending x 5%	3.452	0.412	0.254	0.665
Cold blending x 10%	7.111	0.783	0.573	1.313
Steam extraction x 5%	7.690	2.863	3.352	5.620
Steam extraction x 10%	14.733	6.015	6.207	12.222
P value	<.0001	<.0001	<.0001	<.0001

Values with different letters (a-c) within the same column are significantly different at P < 0.05.

3.4 Antioxidant Activity of the Jamun Fruit Yoghurt

As shown in Table (4) the antioxidant activity indicators of the jamun-containing Yoghurt experimental samples were significantly higher than those of the control plain sample which was made without jamun juice. Concerning the DPPH radical scavenging activity, data revealed that the depending on steam for jamun fruit juice extraction for the application in fruit Yoghurt caused to increase the DPPH % of Yoghurt when compared to the cold extraction method. That could be ascribed essentially to the considerably high hydrolysable tannins contents of the steam extracted jamun Yoghurt samples, with its -oH groups of the aromatic rings which enable the tannins to form complexes with other macromolecules. These observations were supported by the previous findings of Amoako & Awika, 2016. Furthermore, the DPPH radical scavenging activity was increased in the jamun containing yoghurt in a dose-depending style, as the control sample obtained the lowest significant value, followed by the yoghurt with 5% juice then the samples with 10% juice. The inhibition degree of ABTS radical formation was increased in the fruit yoghurt samples when compared to the control plain one. The ABTS radical scavenging activities of yogurt reached its highest value in the 10% steam extracted juice containing sample, while the lowest value was for the control Yoghurt. Moreover, the control Yoghurt reflected a relatively high reducing power activity possibly due to the lactic acid bacteria metabolic activity; this high reducing power was increased significantly when the jamun juice was incorporated into Yoghurt milk. It is worth noting to mention that the increase in the reducing power of Yoghurt with the addition of jamun extracts could be ascribed essentially to the phytochemical content of the jamun pulp (Sarkar et al., 2015). Likewise, the highest reducing power activity was achieved when the steam extraction method was applied. This phenomenon was in accordance with the results of Gong et al., 2012 who studied the effect of treatment with steam on the antioxidant capacity of barley bran.

3.5 Sensory Evaluation of Jamun Fruit Yoghurt

As declared from Figures (1, 2 and 3), the application of the cold blending method for jamun fruit juice extraction was associated with the highest significant values of thickness and smell, while the other extraction method *i.e.*, the steam one was distinguished with the increased values of color, appearance, flavor, taste and overall acceptability. The increase in the Yoghurt taste and flavor scoring upon application of steam extraction could be ascribed to the effect of heating on the breakdown of carbohydrates into different soluble sugar compounds, leading to support the yoghurt taste and aroma (Pantastico, 1986). Regarding the effect of the percentage of jamun fruit extract addition, the control plain Yoghurt got the highest scores of appearance and thickness, as well as the lowest score of smell. Adding the fruit extract at a dose of 5% led to enhance the

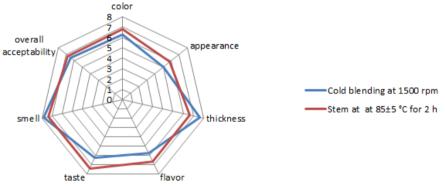
Yoghurt color, flavor, taste smell and total acceptability, as well as to reduce the scores of appearance and thickness. On the other hand, the Yoghurt of 10% fruit extract kept its smell score higher than the control and took the lowest scores of all other judging criteria. It is noteworthy to mention that all the experimental Yoghurt samples were totally accepted by the panellists as their overall acceptability scores were more than 5 out of 9 and the highest overall organoleptic acceptability was for the Yoghurt of 5% steam extracted jamun juice.

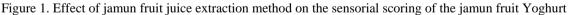
Table 4. Ls means values for method of Jamun fruit juice extraction, percent of juice addition and method of Jamun fruit juice extraction x percent of juice addition on the antioxidation power indicators of the Jamun fruit yoghurt

Source of variation	*DPPH (%)	ABTS (%)	Reducing power (%)					
Method of Jamun fruit juice extraction (M)								
Cold blending at 1500 rpm	72.23 ^b	62.44 ^b	32.16 ^b					
Stem at 85 ± 5 °C for 2 h.	75.63 ^a	65.84 ^a	34.44 ^a					
P value	<.0001	<.0001	<.0001					
Percentage of Jamun fruit ex	Percentage of Jamun fruit extract addition (P)							
0 %(control)	62.55 ^c	48.74 ^c	30.64 ^c					
5%	74.83 ^b	66.75 ^b	33.79 ^b					
10%	84.41 ^a	76.93 ^a	35.48 ^a					
P value	<.0001	<.0001	<.0001					
МХР								
Control	62.55	48.74	30.64					
Cold blending x 5%	72.28	64.20	32.74					
Cold blending x 10%	81.86	74.38	33.12					
Steam extraction x 5%	77.38	69.30	34.84					
Steam extraction x 10%	86.96	79.48	37.84					
P value	<.0001	<.0001	<.0001					

* DPPH: 2,2-diphenyl-1-picrylhydrazyl; ABTS: 2,2'-azino-bis-3-ethylbenzothiazoline- 6-sulfonic acid.

Values with different letters (a-c) within the same column are significantly different at P<0.05.





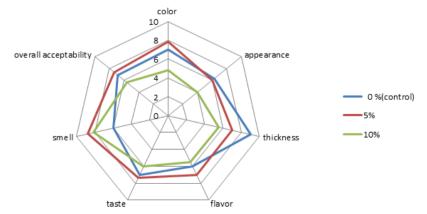


Figure 2. Effect of percent of jamun fruit juice addition to the Yoghurt standardized milk on the sensorial scoring of the jamun fruit Yoghurt

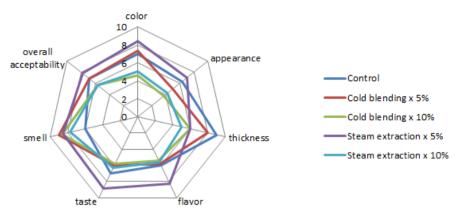


Figure 3. Effect of the interaction between the method of jamun juice extraction and percent of jamun fruit juice addition to the Yoghurt standardized milk and on the sensorial scoring of the jamun fruit Yoghurt

4. Conclusion

In this study, Jamun fruit was subjected to either mechanical cold extraction or steam extraction and added to the Yoghurt milk at rates of 5 and 10%, to evaluate its effects on the flavonols profiles, antioxidant activity and sensorial characteristics of the final Yoghurt product. The study findings led to conclude that, Jamun fruit may be a promising source for fortifying Yoghurt with flavonols and enhancing its antioxidant power and its overall sensorial acceptability.

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