GC-MS and HPLC-ESI-MS-MS Characterization of *Sanchezia* oblonga (Acanthaceae) Extracts

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Abstract

The genus Sanchezia (Acanthaceae) comprises neotropical herbs and shrubs with showy flowers. Sanchezia oblonga (syn. S. nobilis) is a shrub of the rainforests of central and south America. The ethanolic extracts of leaves and stems from S. oblonga were analyzed by GC-EI-MS and RPHPLC-DAD-ESI-MS/MS. Fatty acids (free and esterified) and phytosterols were detected by the former method. Benzyl alcohol glycosides (21 and 25), sinapic acid glycoside esters (29 and 31), ethyl rosmarinate (24), sinapic acid-O-glucoside (28), dihydrosinapic acid-O-glucoside (26), catechin-O-arabinoside (36), in addition to flavonols glycosides (23, 32, 33 and 35) and rosmarinic acid-3'-O-glucoside (34) were detected by RPHPLC-DAD-ESI-MS/MS. Three new compounds, detected only in leaves, were tentatively identified as phenylpropane glyceride derivatives 1-O-coumaroyl-2-hydroxy propanal (20) and 1-O-coumaroyl-2-O-glycosyl propanal (22, 30). Compounds 20, 22 and 30 from S. oblonga are similar with phenylpropane glycerides present in red sorghum (Sorghum bicolor L. (Moench) and Lilium longiflorum Thunb. It is noteworthy that S. oblonga could be used in cooking as a complement after more detailed studies. Sorghum grain foods exhibit potential health benefits against chronic diseases related to over-nutrition. Lilium longiflorum possess flower buds and bulbs that are used for both culinary and medicinal purposes in many parts of the world. Studies on chemical composition and biological activity of the genus Sanchezia are scarce. The presence of phytosterols and flavonol glycosides were recently reported in leaves from this species. However, the chemical profile of the extracts analyzed in this work differs from that previously reported for aerial parts of S. nobilis (sin. S. oblonga). Further studies, including statistical methods, such as principal component analysis and hierarchical cluster analysis will be needed to evaluate chemical markers for this species.

Keywords: Sanchezia oblonga, Acanthaceae, sinapic acid ester glycosides, phenylpropane glycerides, phytosterols

1. Introduction

Species of *Sanchezia* Ruiz & Pav. (Acanthaceae) are neotropical herbs and shrubs. Several species of *Sanchezia*, such as *S. parvibracteata*, *S. nobilis* and *S. speciosa*, are cultivated as ornamentals in tropical areas and botanical gardens due to their showy leaves and conspicuous bright and colorful flowers. Recently, this genus was revised by Leonard and Smith (1964); among 58 species, over half were newly described (Tripp and Koenemann, 2015). There are few reports on chemical constituents and pharmacological activity of *Sanchezia* species. Therefore, there is no exhaustive identification of the basic chemical constituents and comprehensive quality control of this genus. *S. speciosa* contains cardiac and flavonoid glycosides and their extracts exhibited antioxidant and anti-inflammatory activities (Bui Thanh et al., 2017). Extracts of the same species exerted cytotoxicity in human epithelial cervical cancer cell lines (Shaheen et al., 2017, Parvin et al., 2015).

Sanchezia oblonga Ruiz & Pav. (syn. Sanchezia nobilis Hook. f.) is a perennial evergreen shrub from the rainforests of Central and South America (Ellah et al. 2013, 2014). An earlier study reported that *S. nobilis* extracts contain syringin, flavone glycosides, cinnamyl and benzyl alcohol glycosides, neolignan glucoside, besides matsutake alcohol glycosides (Ellah et al. 2013, 2014). The presence of the phytosterols daucosterol

and stigmasterol, flavonol glycosides, scopoletin and 3'-O-methyl-3,4- methylenedioxy ellagic acid in ethylacetate extract of the leaves of *S. nobilis* was reported recently (Xuan et al., 2019a, Loi et al., 2019a). On the other hand, mangiferin, β -sitosterol, in addition to margaric, ursolic and oleanolic acids were isolated from *n*-hexane extracts (Xuan et al., 2019b), while 4',5,7-trihydroxy-3',5'-dimethoxyflavone and kaempferol glycosides were isolated from aqueous extracts of leaves (Loi et al., 2019b).

In this study, the ethanolic extracts of *S. oblonga* leaves and stems were analyzed by GC-EI-MS and RPHPLC-ESI-MS / MS. Phenylpropane glycerides similar with thoses found in red sorghum *(Sorghum bicolor L. (Moench)* and and *Lilium longiflorum* Thunb. were detected only in ethanolic extracts of leaves, while flavonol glycosides and phytosterols were detected only in ethanolic extracts of stems.

2. Material and Methods

2.1 Plants Material

S. oblonga leaves and stems were collected from plants growing on the campus I of Santo Amaro University (UNISA), S $\tilde{a}o$ Paulo, State of S $\tilde{a}o$ Paulo (southeast Brazil). A voucher sample was deposited at the Santo Amaro University Herbarium (UNISA 4121). The leaves and stems were washed and cleaned prior to air drying at room temperature for one week. The dried and powdered leaf and stem material (2 g each) was macerated separately for 30 days in a 500 ml conical flask containing 200 mL of high quality ethanol (Sigma Chemical Co). The solvent was replaced weekly during extraction at room temperature and protected from light. The extracts were filtered first through cotton and then through Whatman paper # 1 and then concentrated in rotary evaporator (B üchi, Switzerland). Leaves and stems provided 130 mg and 90 mg, respectively of dried extracts, which were stored in amber flasks at 5 °C.

2.2 Analyses of the Leac Extract by Fourier Transform Infrared Spectrometry (FT-IR)

Only the ethanolic extract from leaves was analyzed by FT-IR. Several drops of the extract from leaves were placed onto a KBr aperture plate and sandwiched under another aperture plate such that no gas bubbles were trapped. The FT-IR spectrum was obtained using a Frontier PerkinElmer spectrometer equipped with a KBr beam splitter. The spectra were recorded at room temperature from an average of 32 scans in the range 4000-400 cm⁻¹, with 5 cm⁻¹ spectrum resolution, normalization at 1030 cm⁻¹ and aperture of 5.0 mm. The influences of H₂O and CO₂ were subtracted automatically by the software operating the spectrometer.

2.3 GC/EIMS Analyses of Ethanolic Extracts from Leaves and Stems

Crude extracts from leaves and stems (5 mg) were dissolved in 1 mL of high grade ethanol supplied (Sigma Chemical Co). An aliquot of 1 μ L of each extract was analyzed using a Shimadzu GCMS-QP505A gas chromatograph equipped with a ZB-5ms fused silica capillary column BPX5 (5% phenyl arylene/95% dimethylpolysiloxane, 30 m x 0.25 mm internal diameter x 0.25 μ m film thickness) coupled to an ion-trap mass detector. Mass spectra were acquired in the electron-impact (EI) mode with an ionization voltage of 70 eV. The GC conditions were set as follows: the oven was programmed with an initial temperature of 100 °C, maintained for 5 min and then increased to 320 °C at a rate of 6 °C/min. The final temperature was maintained for 10 min. The air and hydrogen flow rates were 400 mL/min and 29.5 mL/min, respectively. Helium was used as the carrier gas at a flow rate of 2.1 mL/min, linear velocity of 53.8 cm/sec, column pressure of 150.0 KPa and total flow of 29.5 mL/min. The MS conditions were set as follows: filament current 0.3 mA; detector voltage -0.7 kV, ion source temperature 300 °C; interface temperature, 300 °C; split ratio of 11 and scan speed of 2 scans s⁻¹. The mass range was 120-700 Da over 52 min (full scan mode) (Negri et al., 2018).

Fragmentation patterns of the EI mass spectra of compounds were compared via spectrum matching with reference mass spectra of libraries NIST 21, Wiley 275 and Wiley 229 (Hewlett Packard, Wiley/NBS) (Koo et al. 2013). The identification of constituents was based on the best mass spectrum (MS) matching score (NIST Mass Spectrum number was added in Table 1) and MS data published in literature. Relative content was estimated by integration of the areas under the corresponding chromatogram peaks.

2.4 RPHPLC-SPD-ESI-MS/MS Analyses of Ethanolic Extracts from Leaves and Stems

HPLC grade methanol was purchased from Merck (Darmstadt, Germany). HPLC grade water was prepared with distilled water using a Milli-Q system (Millipore, Waters, Milford, MA, USA). The ethanolic extracts of leaves and stems (5 mg dissolved in 2 ml of methanol – 2.5 mg/mL) were filtered through 0.45 μ m polytetrafluoroethylene (PTFE) membrane filters, before a 30 μ L aliquot was injected into the HPLC system (Coelho et al., 2018). The RPHPLC-SPD-ESI-MS/MS analysis were conducted using an SPD-M10AVP Shimadzu system equipped with a photodiode array detector coupled to an Esquire 3000 Plus Mass Spectrometry system (Bruke Daltonics, Billerica, MA, USA), coupled with two LC-20AD pumps, SPD-20A diode array

detector, CTO-20A column oven and SIL 20AC autosampler (Shimadzu Corporation Kyoto, Japan). All the operations, acquisitions and data analyses were controlled by the Shimadzu CBM-20A software. Separations were carried out using a C18 RP Luna Phenomenex reverse phase column (4.6 x 250 mm i.d., 5 µm of particle size) protected with a security guard cartridge (Gemini C18, $4.0 \times 2.0 \text{ mm i.d.}$). The mass spectrometer was an ion trap with atmospheric pressure ionization method through electrospray ionization interface (ESI) operating in the scan MS mode from m/z 100 to 1500. The mobile phase was composed of eluent A (0.1% aq. formic acid) and eluent B (methanol) at the constant flow rate 1.0 mL min⁻¹ and constant temperature of the column oven at 40 °C. The following elution program, based on concentrations of the B solvent, was used: 0 min, 20%; 10 min, 40%; 20 min, 60%; 30 min, 80%; 40 min, 100%; 50 min, 20%. The time for the whole run was 60 min, 10 minutes were needed for equilibration of the column. Helium was used as the collision and nitrogen as the nebulizing gas, respectively. Nebulization was aided with a coaxial nitrogen sheath gas provided at pressure of 27 psi. Mass spectra were acquired in negative mode with ion spray voltage at 3.0 kV, capillary temperature at 300°C, capillary voltage at 45 V and drying gas flow 6 L/min. Collision induced dissociation spectra were obtained in the ion trap using helium as the collision gas, with voltage ramping cycles from 0.5 to 1.3 V. Data were analyzed by an HP Chemistation System and Bruker Daltonics Data Analysis.

The analytes were characterized by their respective ultraviolet and mass spectra, which were compared with literature data and the mass spectra database Phenol-Explorer (www.phenol-explorer.eu), ChemSpider (http://www.chemspider.com), Metlin (http://metlin.scripps.edu) and HMDB (www.hmdb.ca).

3. Results

3.1 Analyses of the Leaf Extract by Fourier Transform Infrared Spectrometry (FT-IR)

Fourier transform infrared spectrometry is used for analysis, because it represents a non-invasive analytical tool allowing a fast and simultaneous qualitative and quantitative characterization of natural products and their constituents. The infrared (IR) spectrum of the ethanolic extract of leaves exhibited bands at 3369, 3012, 2925, 2854, 1737, 1640, 1454, 1409, 1378, 1346, 1240, 1162 and 1073 cm⁻¹. The maximum absorptions were attributed to the following functional groups: at 3369 - 3200 cm⁻¹ (stretching of free OH, as sharp peak), 2925 - 2854 cm⁻¹ (C-H aliphatic stretching), 1737 cm⁻¹ (C=O ester bond) and 1640 cm⁻¹ (C=C aromatic). The C–O stretching characteristic from esters appeared in the region 1346 – 1240 cm⁻¹. The peak at 1737 cm⁻¹ assigned to the C=O stretching vibration means that some carbonyl compounds exist in the leaves of *S. Oblonga*. Bands from 1240 to 1162 were attributed to C-O stretching of phenolics and asymmetric C-C-O stretching of esters. The bands between 1454- 1409 cm⁻¹ are attributed to the asymmetric in-plane bending of – CH₃, which is also the same spectral region reflecting to the phenyl (C=C bonds). The absorption bands that were attributed to C=C stretching vibration at 1640 cm-1 and the band at 1378 cm-1 referred to C-O stretching vibration of phenyl groups were observed indicating the presence of phenolic compounds. FT-IR analysis results revealed the presence of fatty acids and its respective ethyl esters and phenolic compounds, such as phenylpropane glycerides.

3.2 Compounds Detected In Leaves and Stems by GC-EI-MS Analysis

The extraction of phenolic compounds depends mainly on the nature of the sample matrix and its chemical properties including molecular structure, polarity of solvent, concentration, number of aromatic rings and hydroxyl groups. Identification of compounds 1-18 (Table 1) was performed by mass spectra through molecular ions M^{+} ⁻⁾ and fragmentation patterns, which were compared with NIST 21 and Wiley-275 libraries. For compounds 1-18, the similarity percentages obtained are equal to or greater than 95% with the NIST Match Factor - 850.6. Mass spectral data from the literature has also been used to identify fatty acids (4, 7, 10 and 11) (Hong and Kim, 2013) and their ethyl esters (5, 9 and 12) (Mon ção et al., 2015, Sudha et al., 2013), phytosterols (14-18) (Chen et al., 2015, Suttiarporn et al., 2015), squalene (13), butanoic acid butyl ester (1) (Oliveira-Garcia et al., 2015), methyl salicylate (2) (Jayasekara et al., 2002), 5-methyl-4-hepten-3-one (3), 9,12-octadecadienal (8) (Hong et al., 2013) and phytol (6) (Sudha et al., 2013). High content of 5-methyl-4-hepten-3-one (3) was detected in both extracts. Phytosterols 14-18 were not detected in the ethanolic extract of the leaves. Figure 1 shows the GC-EI-MS mass spectrum of ethyl palmitate (5), stigmasterol (15) and sitosterol (16).

Cp.N*	Rt**(m	EIMS[M ⁺] m/z	EIMS – Fragments <i>m/z</i> (%)	Identification and	Stems	Leaves
	in)	(%)		(NIST Mass Spectrum number)	(%)	(%)
1	4.6	144 (34)	101 (40), 89 (70), 71 (100)	butanoic acid butyl ester (228835)	2.8	1.5
2	5.6	152 (50)	121 (25), 120(100), 92 (60)	methyl salicylate (291552)	1.1	-
3	6.8	126 (50)	97 (100), 69 (50)	5-methyl-4-hepten-3-one (63310)	31.1	21.4
4	24.3	256 (11)	213 (13), 157 (14), 129 (32),	palmitic acid (151973)	5.4	7.3
			115 (14), 83 (40), 73 (100)			
5	24.8	284 (5)	241 (7), 157 (15), 101 (60), 88 (100)	ethyl palmitate (233204)	3.7	4.7
6	27.0	296 (1)	123 (40), 111 (40), 95 (50), 81 (60), 71 (100)	phytol (375015)	1.9	41.5
7	27.4	280 (4)	124 (10), 110 (17), 95 (57), 81 (75), 67 (100)	linoleic acid (229327)	8.2	1.5
8	27.8	264 (4)	165 (2), 108 (30), 95 (50), 79 (60), 67 (70)	9,12-octadecadienal (35820)	7.0	-
9	28.0	308 (3)	262 (6), 153 (15), 135 (14), 109 (23), 107 (10), 95 (59), 81 (81), 67 (100)	ethyl linoleate (155747)	5.5	2.7
10	28.4	282 (1)	264 (7), 222 (20), 180 (30), 135 (12), 121 (16), 108 (24), 101 (27), 95 (53), 69 (100)	oleic acid (134027)	4.5	8.5
11	28.7	284 (10)	241 (30), 185 (40), 129 (60), 73 (100).	stearic acid (290961)	1.0	-
12	30.2	312 (8)	157 (13), 101 (50), 88 (100)	ethyl octadecanoate (23294)	0.6	-
13	38.7	410 (1)	149 (30), 136 (30), 121 (30), 109 (30), 95 (30), 81 (63), 69 (100)	squalene (227620)	0.5	3.1
14	44.3	400 (21)	382 (30), 367 (10), 315 (22), 289 (22), 255 (15), 213 (57), 145 (50), 107 (60), 95 (70)	campesterol (151556)	1.3	-
15	44.6	412 (24)	397 (3), 394 (3), 369 (5), 351 (20), 300 (30), 271 (19), 255 (20), 159 (40), 145 (30), 133 (60)	stigmasterol (352610)	5.8	-
16	45.4	414 (35)	399 (18), 396 (19), 381 (21), 329 (33), 303 (17), 273 (13), 255 (22), 213 (39), 207 (41), 173 (21),161 (36), 147 (28), 133 (32)	sitosterol (251915)	4.8	-
17	46.2	410 (12)	367 (33), 298 (23), 271 (20), 245 (20), 191 (17), 147 (23), 133 (27), 119 (15)	stigmasta-4,22-dien-3-one (255385)	0.6	-
18	47.0	412 (30)	398 (20), 370 (30), 289 (50), 229 (40), 149 (40), 135 (25), 124 (100), 107 (22)	stigmast-4-en-3-one (17165)	0.9	-

Table 1. Compounds 1- 18 detected in the analysis of ethanol extract of stems and leaves of *Sanchezia oblonga* by GC-IE-MS, identified through NIST Mass Spectra and MS data from literature (see text)

*Cp.N: compound number. **Rt: retention time.



Figure 1. GC-EI-MS mass spectra of compounds 5, 15, 16

3.3 Compounds Detected in Leaves and Stems by RPHPLC-ESI-MS/MS Analyses

The identification of compounds 19-36 (Table 2) was based on the MS data in negative ionization mode conjugated with the UV-DAD, which were compared with literature data. In negative ion mode, high abundances of quasi-molecular ions [M-H]⁻ and [M-H + HCOOH]⁻ were exhibited by compounds 19-36. (Fal ção et al., 2013). The glycosyl and alkyl residues are weak chromophores, so these substituents did not change significantly, the

maximum absorption coefficients and absorption wavelengths of compounds 19, 21-23 and 25-36 (Table 2), indicating that the loss of 132, 146, 162 and 176 Da, in these compounds corresponds to arabinose, rhamnose, glucose and glucuronide, respectively.

3.4 Compounds Detected by RPHPLC-ESI-MS/MS Analyses of Leaf Ethanolic Extract

Compounds 20, 22, 29, 30, 31, 35 and 36 were detected only in ethanolic extract from leaves (Table 2). Compounds 20, 22 and 30 are new and their identification was based on UV and mass spectra, which were compared with that phenylpropane glycerides isolated from sorghum grain (Kang et al., 2016, Kadam et al., 2018; Nguyen et al., 2015, and Svensson et al., 2010) and Lilium longiflorum Thunb. (Munafo Jr and Gianfagna, 2015). Sorghum is the fifth most-produced cereal in the world and the major producers are USA, Mexico, Nigeria, Sudan and India. Sorghum grain foods is used in many parts of Africa, Asia and the semi-arid tropics world-wide and exhibited potential health benefits against chronic diseases related to over-nutrition. Compound 20 (Rt - 3.6 min) exhibited UV maximum absorption at 310 nm, characteristic of coumaric acid derivatives. Their mass spectra exhibited [M - H]- ion at m/z 235, base peak at m/z 217 and fragment ion at m/z 199, both resulting from water loss (18 Da). The fragment ion at m/z 155 was formed by the loss of carbon dioxide (44 Da) from fragment ion at m/z 199. Compound 20 was tentatively identified as 1-O-coumaroyl-2-hydroxy propanal. Compounds 22 (Rt - 8.3 min) and 30 (Rt - 13.3 min) displayed the same fragmentation pattern with a base peak at m/z 235 and were characterized as 1-O-coumaroyl-2-hydroxy propanal derivatives. Compound 22 exhibited [M - H + HCOOH] adduct ion at m/z 413, [M - H] ion at m/z 367 and the base peak at m/z 235 [M - H - H]132]-, was produced by loss of arabinose (132 Da). Based on literature data (Kang et al., 2016, Kadam et al., 2018; Nguyen et al., 2015, Munafo Jr and Gianfagna, 2015; and Svensson et al., 2010), compound 22 was tentatively identified as 1-O-coumaroyl-2-O-arabinosyl propanal. Compound 30 exhibited [M - H + HCOOH]adduct ion at m/z 427, [M - H]- at m/z 381 and the base peak at m/z 235 ([M - H - 146]-, was produced by loss of rhamnose (146 Da). Based on literature data, (Kang et al., 2016; Kadam et al., 2018; Nguyen et al., 2015; Munafo Jr and Gianfagna, 2015; and Svensson et al., 2010), compound 30 was tentatively identified as 1-O-coumaroyl-2-O-rhamnosyl propanal.

p-Coumaroylglycolic acid was reported for *Lepidium sativum* (Kadam et al., 2018). Caffeoylglycolic acid methyl ester was isolated from the grains of *Sorghum bicolor* (L.) Moench var. *hwanggeumchal* Kang et al., 2016; Nguyen et al., 2015). 2-*O*-coumaroylglycerol and 1-*O*-coumaroyl-2-*O*-glucosylglycerol were detected in hydromethanolic extracts (50:50) of brown, red and white sorghum whole grains (Kang et al., 2016; and Svensson et al., 2010). Feruloyl-caffeoylglycerol was found in *Ananas comosus* L. leaves and sorghum grain (Wu et al., 2016). Sorghum grains in the diet promoted cardiovascular health and exhibited beneficial effects for weight control (Kang et al., 2016). Caffeoylglycolic acid methyl ester and 1-*O*-caffeoylglycerol isolated from *Sorghum bicolor*, showed inhibitory potential on nitric oxide production (Salazar-López et al., 2018). *Lilium longiflorum*, an attractive ornamental plant, possesses flower buds and bulbs that are used for both culinary and medicinal purposes in many parts of the world. This species contains significant amounts of phenylpropanoid glycerol glucosides, which may contribute to plant pathogen defense, ultraviolet/high-intensity visible light (UV/high light) protection, and use in traditional medicine (Munafo Jr and Gianfagna, 2015).

Compound 29 (Rt - 12.8 min) exhibited UV maximum absorption at 318 nm, characteristic of cinnamic acid derivatives. Its mass spectrum exhibited [M - H + HCOOH]- adduct ion at m/z 491, [M - H]- at m/z 411 and base peak at m/z 279 (deprotonated butyl sinapate), produced by the loss of arabinose (132 Da). Based on literature data (Teixeira et al., 2013; Nićiforović et al., 2017; Martinović et al., 2019), compound 29 was tentatively identified as 4-*O*-arabinosyl butyl sinapate. For compound 31 (Rt - 13.5 min) the mass spectrum exhibited [M - H + HCOOH]- adduct ion at m/z 461 and [M - H]- at m/z 415. Based on literature data (Teixeira et al., 2017; Martinović et al., 2019), compound 31 was tentatively identified as 4-*O*-glucosyl ethyl dihydrosinapate.

Compound 35 (25.1 min) exhibited UV maximum absorption at 266 and 355 nm, characteristic of flavonol. Band I (330 – 380 nm) is attributed to the B-ring cinnamoyl system, and Band II (240 – 280 nm) is attributed to the A-ring benzoyl system. Mass spectrum of 35 exhibited [M – H + HCOOH]– adduct ion at m/z 493 and [M –H]- ion at m/z 447. Based on UV and MS data compared with literature (Soares et al., 2019; Vukics et al., 2010; Pitura and Arntfield, 2019, Loi et al., 2019), compound 35 was identified as kaempferol-3-*O*-glucoside. Compound 36 (Rt - 26.5) exhibited UV maximum absorption at 280 nm and the mass spectrum gave [M – H]- at m/z 421. Its MS/MS fragmentation yielded base peak at m/z 289 (deprotonated catechin), produced by the loss of arabinose (132 Da). Based on literature data (Kang et al., 2016; Zerbib et al., 2018; Karar and Kuhnert, 2015), compound 36 was tentatively identified as catechin arabinoside.

3.5 Compounds detected by RPHPLC-ESI-MS/MS Analyses of Stem Ethanolic Extract

Compounds 21, 23, 24, 26, 32, 33 and 34 were detected only in ethanolic extract from stems (Table 2). Compound 21 (Rt - 6.1 min, Table 2) exhibited UV maximum absorption at 254 nm. Mass spectrum exhibited $[M - H]^-$ at m/z 315 and base peak at m/z 153 ($[M - H - 162]^-$, corresponding to the loss of glucose moiety. Compound 21 was identified as 4-hydroxy-3-methoxybenzyl alcohol glucoside, based on literature data (Karar and Kuhnert, 2015; Kumar, 2017; Veličković et al., 2012). The mass spectra of compounds 24 (Rt - 10.3 min) and 26 (Rt - 11.0 min) displayed the same $[M - H]^-$ at m/z 387. The mass spectrum of compound 24 (Rt - 10.3 min) exhibited base peak at m/z 163, that resulted to the loss of 225 Da from the molecular ion at m/z 387, which was formed by cleavage of the ester bond (Aziz et al., 2014; Shen et al., 2018; Wicha et al., 2015). Compound 24 was tentatively identified as ethyl rosmarinate. The mass spectrum of 26 exhibited base peak at m/z 207, corresponding to dihydrosinapoyl moiety, which was formed by cleavage of the ester bond that resulted in loss of glucose (180 Da) (Tian et al., 2013; El-Sayed et al., 2017). Compound 26 was tentatively assigned as dihydrosinapic acid-*O*-glucoside.

Compound 34 (Rt - 16.6 min, Table 2) exhibited UV maximum absorption at 328 nm, characteristic of caffeic acid derivatives. The mass spectrum exhibited $[M - H]^-$ at m/z 521 and base peak at m/z 179 (deprotonated caffeic acid) (Liang et al., 2017; Yang et al., 2018, Borr \pm -Linares et al., 2014; Lee et al., 2017). This fragment ion was not observed in mass spectra of rosmarinic acid-3-*O*-glucoside (Yang et al., 2018, Borr \pm -Linares et al., 2014, Lee et al., 2017), but it was detected in rosmarinic acid, the second most common ester of caffeic acid in the plant kingdom (Lee et al., 2017; Hashem et al., 2019). The base peak at m/z 179 was formed by cleavage of the ester bond, which produced the loss of 342 Da, corresponding to the 3,4-dihydroxyphenyl-lactic moiety plus a glucose, probably attached at the 3'-*O* position. Compound 34 was tentatively identified as rosmarinic acid-3'-*O*-glucoside.

Compound 23 (Rt - 8.5 min) and 33 (Rt - 15.9 min) exhibited UV maximum absorption at 266 and 355 nm, characteristic of flavonols. The mass spectrum of compound 23 exhibited $[M - H + HCOOH]^-$ adduct ion at m/z 523 and $[M - H]^-$ at m/z 477 and was identified as quercetin glucuronide (23) (Soares et al., 2019). The mass spectrum of compound 33 exhibited $[M - H]^-$ at m/z 579 and base peak at m/z 417, produced by the loss of glucose moiety (162 Da) (Soares et al., 2019, Vukics and Guttman, 2010) and was identified as kaempferol-7-*O*-arabinosyl-3-*O*-glucoside. Compound 32 (Rt - 14.0 min) exhibited UV maximum absorption at 290 nm, characteristic of flavanonol. Its mass spectrum exhibited $[M - H]^-$ at m/z 581, a fragment ion at m/z 419, resulting from loss of glucose moiety (162 Da) and base peak at m/z 389 (Pitura and Arntfield, 2019). Compound 32 was assigned as dihydrokaempferol-7-*O*-arabinosyl-3-*O*-glucoside.

3.6 Compounds Detected by RPHPLC-ESI-MS/MS Analyses of Leaf and Stem Ethanolic Extracts

Compounds 19, 25, 27 and 28 were detected in ethanolic extracts from leaves and stems (Table 2). The mass spectrum of 19, (Rt - 2.7 min) exhibited [M - H]⁻ at m/z 341, and base peak at m/z 179 (deprotonated caffeic acid), produced by the loss of glucose moiety (162 Da) (Karar and Kuhnert, 2015; Liang et al., 2017). Compared with its elution order and MS/MS fragmentation pattern, compound 19 was assigned as caffeic acid glucoside. For compound 25 (Rt - 10.6 min), the mass spectrum exhibited [M - H]⁻ at m/z 401 (Figure 2), and base peak at m/z 269 (C₁₃H₁₇O₆), produced by the loss of arabinose moiety (132 Da). Benzyl alcohol-7-*O*-arabinosyl glucoside 25 was detected in *Sanchezia nobilis* (Ellah et al., 2014) and was identified based on literature data (Karar and Kuhnert, 2015).

Compounds 27 (Rt - 11.4 min) and 28 (Rt - 11.8 min) exhibited maximum UV absorption at 310 nm, which is characteristic of coumaric acid derivatives. Mass spectrum of 27 exhibited $[M - H + HCOOH]^-$ adduct ion at m/z 491, $[M - H]^-$ at m/z 445 and base peak at m/z 293, which correspond to sinapyl alcohol diacetate after the loss of galloyl moiety (152 Da). Compound 27 was tentatively identified as 4-*O*-galloyl-sinapyl alcohol diacetate based on literature data (Nićiforović et al. 2017, Tian et al., 2013). The mass spectrum of 28 exhibited $[M - H + HCOOH]^-$ adduct ion at m/z 431 and $[M - H]^-$ at m/z 385. Based on literature data (Nićiforović et al., 2017; El-Sayed et al., 2017, Yang et al., 2018), compound 28 was identified as sinapic acid-*O*-glucoside. The mass spectrum of compounds 20, 22, 24, 25 and 34 are shown on Figure 2. The proposed structures of compounds 20, 22, 24, 30 and 34 are shown on Figure 3. Further studies should be performed through the isolation of these compounds in order to confirm the identification by NMR methods.



Figure 2. HPLC-ESI-MS/MS spectra in negative ion mode of compounds 20 (1-*O*-coumaroyl-2-hydroxy propanal; **A**), 22 (*1-O*-coumaroyl-2-*O*-arabinosyl propanal; **B**), 24 (ethyl rosmarinate; **C**), 25 (benzyl alcohol-7-O-arabinosyl glucoside; **D**) and 34 (rosmarinic acid-3'-*O*-glucoside; **E**)

Cp.N ^a	Rt ^b	UV	RPHPLC/(-)ESI-	Proposed	Refer. ^d	Stems ^e	Leaves ^f
	(min)	(nm)	MS/MS mz (%)	Compound ^c			
19	2.7	320	[M–H] ⁻ : 341	caffeic acid	32,43	+	+
			MS/MS:	glucoside			
			179 (100),				
			161 (40), 143 (40)				
20	3.6	310	[M–H] ⁻ : 235	1-O-coumaroyl-	19-24	-	+
			MS/MS: 217 (100),	2-hydroxy propanal			
			199 (50), 155 (20)				
21	6.1	254	[M–H] ⁻ 315	4-Hydroxy-	34,35	+	-
			MS/MS: 153	3-methoxybenzyl			
				alcohol glucoside			
22	8.3	310	[M-H+HCOOH]	1-O-coumaroyl-	19-24	-	+
			413; [M–H] ⁻ : 367	2- <i>O</i> -arabinosyl			
			MS/MS: 235	propanal			
23	8.5	260,	[M-H+HCOOH]	Quercetin	28	+	-
		355	523; [M–H] ⁻ : 477	glucuronide			
24	10.3	320	[M–H] ⁻ : 387	Ethyl rosmarinate	36-39	+	-
25	10.6	254	[M–H] ⁻ : 401	benzyl alcohol-7- <i>O</i> -	3, 32	+	+
			MS/MS: 269	arabinosyl glucoside			
26	11.0	-	[M–H] ⁻ : 387	dihydrosinapic acid-	41,42	+	-
			MS/MS: 207 (100)	<i>O</i> -glucoside			
27	11.4	325	[M-H+HCOOH]	4-O-galloyl-sinapyl	26,41	+	+
			491; [M–H] ⁻ : 445	alcohol diacetate			
			MS/MS: 293				
28	11.8	330	[M-H+HCOOH]	sinapic acid-O-	26,42,44	+	+
			431; [M–H] ⁻ : 385	glucoside			
29	12.8	318	[M-H+HCOOH]	4- <i>O</i> -arabinosyl		-	+
			457; [M–H] ⁻ : 411	butyl sinapate	25-27		
			MS/MS: 279				
30	13.3	315	[M-H+HCOOH]	1-O-coumaroyl-2-O-rhamnosyl		-	+
			427; [M–H] ⁻ : 381	propanal	19-24		
			MS/MS: 235				
31	13.5	-	[M-H+HCOOH]	4-O-glucosyl-ethyl-	25-27	-	+
			461; [M–H] ⁻ : 415	dihydrosinapate			
32	14.0	290,	[M-H]: 581	dihydrokaempferol-		+	-
		330 sh	MS/MS: 419 (50),	7- <i>O</i> -arabinosyl-3- <i>O</i> -	30		
			389 (100)	glucoside			
33	15.9	266,	[M–H] ⁻ : 579	kaempferol-7- <i>O</i> -		+	-
		348	MS/MS: 417	arabinosyl-3-O-	28,29		
				glucoside			
34	16.6	328	[M-H+HCOOH]	rosmarinic acid-	43-46	+	-
			567; [M–H] : 521	3'-O-glucoside			
			MS/MS: 179	č			
35	25.1	266.	[M-H+HCOOH]	kaempferol-7-0-glucoside	28-30	-	+
		355	493; [M–H] ⁻ : 447				
36	26.5	280	[M–H]: 421	catechin-O-arabinoside	21,31-33	-	+
			MS/MS: 289				

Table 2. Pseudomolecular ions and MS/MS fragment ions obtained for compounds 19-36 by RPHPLC-SPD-ESI-MS/MS analyses in negative ionization modes

^aCp.N: compound number.

^bRt: retention time.

cidentified based on MS/MS and UV data and their comparison with MS/MS and other data from reference sources.

^dreferences used in identification (see text)



Figure 3. Proposed structures for compounds 20, 22, 24, 29, 30 and 34

4. Discussion

Differences were observed comparing the distribution of constituents of leaves and stems extracts by GC-EI-MS (Table 1) and RPHPLC-DAD-ESI-MS/MS analyses (Table 2). In the analyses by GC-EI-MS (Table 1), the steroids 14-18 were found only in extract from stems. Other unique constituents of stem extracts were methyl salicylate (2) and 9,12-octadecadienal (8). Stigmasterol (15) and sitosterol (16) were detected in higher amounts (5.8% and 4.8%, respectively) than campesterol (14, 1.3%), stigmasta-4,22-dien-3-one (17, 0.6%) and stigmasta-4-en-3-one (18, 0.9%). 5-Methyl-4-hepten-3-one (3) was the main constituent of the ethanolic extract from stems (31.1%) and was also detected with high content in leaf ethanolic extract (21.4%, Table 1).

Regarding the RPHPLC-ESI-MS/MS analyses (Table 2), the new compounds, tentatively identified as phenylpropane glycerides derivatives (20, 22 and 30), sinapic acid glycoside esters (29 and 31), kaempferol-7-*O*-glucoside (35) and catechin-*O*-arabinoside (36) were detected only in ethanolic extract from leaves. Caffeic acid glucoside (19), benzyl alcohol-7-*O*-arabinosyl glucoside (25), 4-*O*-galloyl-sinapyl alcohol diacetate (27) and sinapic acid-*O*-glucoside (28) were detected in the two ethanolic extracts. On the other hand, 4-hydroxy-3-methoxybenzyl alcohol glucoside (21), ethyl rosmarinate (24), dihydrosinapic acid-*O*-glucoside (36) were detected only in ethanolic extract from stems (Table 2). It is noteworthy that ethyl rosmarinate (24) and rosmarinic acid-3'-*O*-glucoside (34) were unique in the ethanolic extract of stems. Differences in gene expression between plant parts is well known. For example, plant polyphenol profile varied widely ontogenetically and among organs of *Geranium sylvaticum* (Tuominem and Salminem, 2017).

There have been few chemical studies about genus *Sanchezia*. In this study a hyphenated chromatographic analysis of ethanolic extracts of leaves and stems of *S. oblonga* was performed. Although the isolation of constituents is necessary for their identification by NMR methods, it is possible to observe that the results are different from those previously obtained for this species (Ellah et al., 2014). Table 2 lists various synapic acids and their esters, phenylpropane glycerides derivatives and rosmarinic acid derivatives, which had not been reported previously for this species. In this work, phytosterols and kaempferol glycosides were detected only in stems. However, a recent work reported that phytosterols were isolated from *n*-hexane and ethylacetate extract of the leaves (Xuan et al., 2019a,b, Loi et al., 2019a), while kaempferol glycosides were isolated from ethylacetate and aqueous extracts of leaves of *S. nobilis* (syn. *S. oblonga*) (Xuan et al., 2019b, Loi et al., 2019b).

The differences in chemical composition between the extracts analyzed in the present work and the previously published papers regarding *S. nobilis* can be explained by two possibilities: a) they may correspond to the same species but to different varieties or chemotypes; b) differences may be due to climate, season or growing

conditions (Liu et al., 2015).

S. oblonga may turn out an interesting source of bioactive substances. Phytosterols exhibited antibacterial and antifungal activity (Aldini et al., 2014), are active against leukemic cell lines (Suttiarporn et al., 2015) and useful in the treatment of gastrointestinal inflammatory diseases by association with systemic and local metabolic anti-inflammatory drugs (Burčová et al., 2018). Stigmasterol inhibited tumor endothelial cells and cholangiocarcinoma (Saeidnia et al., 2014, Kangsamaksin et al., 2017). Sterols inhibited cholesterol absorption and exhibited trypanocidal and mosquito larvicidal activity (Ghosh et al., 2013). Hydroxycinnamic acids and their derivatives promote a variety of health benefits, e.g. reducing obesity and adverse health complications (Alam et al., 2016), as well as exhibiting antibacterial, antifungal (Guzman, 2014), antityrosinase, UV protection, anti-aging and anti-inflammatory effects (Alam et al., 2016).

Synapic acid derivatives exhibited antioxidant, antimicrobial, anti-inflammatory, anti-cancer, anti-anxiety (Nićiforović et al., 2017), anti-glycemic, neuroprotective, antibacterial (Chen, 2016) and analgesic activities (Hameed et al., 2016). Rosmarinic acid esters exhibited a broad spectrum of biological effects (Shen et al., 2018; Liang et al., 2017; Thammason et al., 2018). Glycerol phenylpropanoid glucosides act as hepatic glucose production inhibitors; structure–activity relationships indicated that the hydroxylation pattern of the hydroxycinnamic acid moiety and acetylation accounted for differences in activity (Murray et al., 2019).

5. Conclusion

Leaves and stems of *Sanchezia oblonga* contain a variety of biologically active compounds, such as phytosterols, flavonol glycosides, benzyl alcohol glycosides, sinapic acid glycoside esters, phenylpropane glycerides and rosmaric acid derivatives, which were detected using GC-EI-MS and RPHPLC-DAD-ESI-MS/MS. These techniques revealed that ethanolic extracts of leaves and stems contain distinct chemical profiles. The leaves contain phenylpropane glycerides and can be used in cooking as supplement. Comparison of the current results with previous studies suggests the possibility of chemical polymorphisms within species. Further studies are needed to explore this possibility. Compounds should be isolated and identified by NMR methods. Statistical methods, principal component analysis and hierarchical cluster analysis can be applied to evaluate intrinsic quality and identify chemical markers, which is useful for the chemical standardization of this species.

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Conflict of Interests

We confirm that there is no conflict of interest associated with this work.

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