# Effect of Hydrogen Peroxide in the Growth of Yellow Passion Fruit Seedlings Under Salinity Stress

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## Abstract

Hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) is a molecule that can flag plants under biotic and abiotic stress conditions. Among the kinds of stress, the salinity stress is the one that most usually affects plants. Consequently, the purpose hereof was to use hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) to mitigate the possible harmful effects of salinity in yellow passion fruit seedlings. We employed a randomized block design, in a 5 × 3 factorial scheme, corresponding to five irrigation water electric conductivity levels (0.3; 1.3; 2.3; 3.3; and 4.3 dS m<sup>-1</sup>) and three hydrogen peroxide concentrations (0; 5; and 15 µmol L<sup>-1</sup>), with four repetitions. The treatments were applied foliarly 7 and 15 days after the seedlings' germination with hand sprayers. Sixty days after sowing, we evaluated the seedlings' growth and quality variables, which finally proved that hydrogen peroxide mitigates the harmful effect of the irrigation water's salinity up to 2 dS m<sup>-1</sup> in the growth of yellow passion fruit seedlings at the concentrations of 5 µmol/L. Nonetheless, excessive concentrations (15 µmol L<sup>-1</sup>) associated with high salt concentrations were proven detrimental to the seedlings' phenological growth and quality.

Keywords: Passiflora edulis, H<sub>2</sub>O<sub>2</sub>, salinity

## 1. Introduction

In Brazil, pomology plays a very important economic and social role in every region of the country: it creates jobs, contributes for the human fixation in the field and for the distribution of the regional income, and it presents promising expectations of internal and external market (Agrianual, 2015).

The yellow passion fruit stands out from all other fruit trees of great expressiveness grown in the country, with the production of around 838,244 tons of fruits in 58,089 ha in the harvest of 2014/2015, which ensured to Brazil the title of the greatest world passion fruit producer. Its significant economic importance to the country is related to the fact that its juice is highly accepted in the international market, and to the fact that the fruit supplies the national market (Faleiro, Junqueira, & Braga, 2008). Nevertheless, the absence of proper handling, lack of cultural cares, and nutritional deficiencies hinder the passion fruit to achieve an excellence level in production for exportation in Brazil (Campos et al., 2013).

In spite of the fact that the Northeast region has proper conditions for the culture's development and the greatest production in the country, it has a yielding of 12.42 ton/ha (IBGE, 2015), which may be mainly related to the region's water deficiency during most of the year, making the producers employ low-quality water for irrigation, which contains high salt levels.

The use of water with high salt concentrations for irrigation reduces the growth and development of sensitive plants, such as the passion fruit tree (Nunes et al., 2016), hindering the development of its maximum potential due to the abiotic stress caused by the excess of salts, which reduces the plant's water absorption level, and causes nutritional unbalance and toxicity due to specific ions (Willadino & Câmara, 2010).

Many are now searching for technologies that can mitigate the salinity's effects to explore fields irrigated with salinity restrictions and/or use of salt water in agriculture to achieve economically viable productions, even in places with elevated ionic contents (Dias, Cavalcante, Leon, Santos, & Albuquerque, 2011; Sá, Mesquita, Bertino, Costa, & Araújo, 2015). Among these technologies, researchers are studying the use of substances that flag the plant's stress and gradually make the vegetable tolerant.

In salinity stress situations, the accumulation of Na<sup>+</sup> and Cl<sup>-</sup> ions in vegetable tissues changes the plant's metabolism, causing, for example, the excessive production of reactive oxygen species (ROS) and consequently the plant's oxidative stress. To mitigate the damages caused by the excess of ROS in the most different cellular compartments, through the balance maintenance between production and elimination, and to maintain the cell's homeostasis, the plants have an enzymatic system composed of superoxide dismutase (SOD), ascorbate peroxidase (APX), and catalase (CAT) enzymes. The SOD, in response to stress, dismutates superoxide ions generating hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), which is dismutated in water (H<sub>2</sub>O) and O<sub>2</sub> through the CAT. Ergo, the exogenous application of H<sub>2</sub>O<sub>2</sub> in a proper amount can cooperate in the response changes of the antioxidant protective system's enzymes, contributing to the increase of the plant's tolerance to salinity stress (Taiz, Zeiger, Moller, & Murphy, 2017; Silveira, S. L. F. Silva, E. N. Silva, & Viégas, 2016).

According to Gondim, Gomes Filho, Marques, and Prisco (2011), the application of  $H_2O_2$  has been proven as an efficient salinity stress mitigator, while Silva, Lacerda, Medeiros, Souza, and Pereira (2016) reported that  $H_2O_2$  is the viable way to acclimatize a plant exposed to an abiotic stress source because it makes the intracellular region activate the plant's defense responses to the stress caused by the excess of salts, resulting in cross-tolerance.

Additionally, according to Petrov and Van Breusegem (2012),  $H_2O_2$  works as a regulator of several biologic process, such as the strengthening of the cell wall, senescence, photosynthesis, stomatal opening, and cell cycle. Nonetheless, the biologic effects of  $H_2O_2$  were proven dependent both of its concentration and of its production site, as well as of the plant's developing stage and previous exposition to other types of stress.

Consequently, purpose hereof was to evaluate the use of hydrogen peroxide as a mitigator of the salinity stress caused by the irrigation water in yellow passion fruit seedlings at their initial stage of growth.

### 2. Material and Methods

The experiment was developed in a greenhouse of the Federal University of Campina Grande's Agri-food Science and Technology Center, Pombal *Campus* (PB), in the period of September to November 2017.

The treatments were arranged in an experimental design of randomized block, in a 5  $\times$  3 factorial scheme, corresponding to five irrigation water electric conductivity levels [(control 0.3); 1.3; 2.3; 3.3; and 4.3 dS m<sup>-1</sup>] and three hydrogen peroxide concentrations [(control 0); 5; and 15 µmol L<sup>-1</sup>] applied through foliar pulverizations, with four repetitions, totalizing 60 experimental units, where each sample unit consisted in one plant. The salinity levels were defined based on the water samples used to irrigate the region.

The sowing was performed in black polyethylene bags with a capacity of 1.2 liter, which were filled with the mixture of soil, sand, and cattle manure in the proportion of 3:1:1, whose chemical characteristics are found in Table 1. Three yellow round passion fruit seeds, of the top seed brand, were sown per bag, at a depth of 1 cm. The thinning process was performed 15 days after the seedlings' germination, remaining only the most vigorous one.

Textural classification	Organic matter	pH(H <sub>2</sub> O)	Р	Sortive complex			
Textular classification				Ca <sup>2+</sup>	Mg <sup>2+</sup>	$Na^+$	$K^+$
	g kg <sup>-1</sup>	mg dm <sup>-3</sup>	$cmol_c dm^{-3}$				
Sandy franc	4.79	6.5	36.6	1.2	0.71	0.07	14.36

Table 1. Chemical characteristics of the substrate used to fill the bags. Pombal, CCTA/UFCG, 2017

The hydrogen peroxide  $(H_2O_2)$  solutions were prepared through the dilution of pure peroxide at 99% in the respective studied concentrations. The application was performed at the end of the afternoon 7 and 15 days after the plants' germination, with the assistance of a hand sprayer, whose solution was applied directly on the leaves until they were completely moist.

The water samples, with their respective conductivity levels, were prepared through the addition of NaCl to the supply water (of 0.3 dS  $m^{-1}$ ) until the desired levels of wEC were obtained, which were measured with the assistance of a portable conductivity meter; said solutions were stored in 30 L, duly protected to avoid evaporation and contamination with materials that might compromise their functionality plastic containers, one for each wEC level evaluated.

The irrigations with salt water begun 16 days after germination (DAG) and were performed every day to keep the soil's humidity close to its maximum retaining capacity, according to the drainage lysimeter method, whose

applied depth receives the addition of a leaching fraction of 15%. The volume applied ( $V_a$ ) by container was obtained by the difference between the previous volume ( $V_{prev}$ ) applied less the mean of drainage (d) divided by the number of containers (n), as indicated in Equation 1:

$$V_a = \frac{V_{prev} - (d/n)}{(1 - FL)} \tag{1}$$

Sixty days after sowing (DAS), the following variables were evaluated: Plant Height (PH): measured with the assistance of a ruler graduated in millimeters from the ground level to the insertion point of the last leaf; Leaf number (LN): by counting the leaves that presented a length above 1 cm; Leaf area (LA cm<sup>2</sup>): calculated through the equation: LA = 0.1525 + 0.8134\*L\*W, where L = Length and W = Width, according to Schmildt, Oliari, Schmildt, Alexandre, Pires et al. (2016); Stem Diameter (SD): with the assistance of a digital caliper (graduated in mm), whose measurement was performed at the plant's base at 1 cm above the soil; Root length (RL); Leaf, stem, and root dry mass (LDM, SDM, and RDM): obtained after washing and drying the different parts of the plants in a muffle with forced ventilation at a temperature of 65 °C, until constant mass was obtained; Total dry mass (TDM): through the sum of the aerial part dry mass and the root dry mass; Dickson quality index (DQI): which was determined according to the variables height, stem diameter, total, aerial part, and root dry biomass, according to Dickson, Leaf, Hosner (1960) through Equation 2:

$$DQI = \frac{TDM}{\left(\frac{PH}{SD}\right) + \left(\frac{DMAP}{RDM}\right)}$$
(2)

Where, TDM = Total dry mass (g); PH = Plant Height (cm); SD = Stem Diameter (mm); DMAP = Dry mass of the aerial part (g); RDM = Root dry mass (g).

Besides these variables, we also determined the water relative level (WRL) and the integrity of the cell membrane through electrolyte extravasation (EE). To determine the WRL, we collected a leaf of each experimental unit, which were promptly weighted in a scale with a precision of 0.001 g (W1), and then deposited for hydration for a period of 24 hours in plastic bags with 100 mL of distiled water. After this period, they were removed from water, weighted (W2), and stored in an air-ventilated muffle at 65 °C for 48 hours to obtain the dry mass (W3). The WRL was calculated through the methodology of Weatherley (1950), according to Equation 3. The leaves were collected by morning to reduce the abiotic effects of radiation and temperature.

$$WRL = \frac{W_1 - W_3}{W_2 - W_3} \times 100 \tag{3}$$

For the electrolyte extravasation, we collected, by experimental unit, eight leaf discs, each one of an area of 2.8 cm<sup>2</sup>, with the assistance of an iron driller, which were then washed and stored in Erlenmeyer vials with 50 mL of distiled water. The Erlenmeyer vials was sealed with foil paper and kept under a temperature of 25 °C for 4 hours. After this period, the initial electric conductivity level in the means was measured ( $X_i$ ) with the assistance of a bench conductivity meter. Then the Erlenmeyer vials were submitted to the temperature of 90 °C for 2 hours in a forced-circulation muffle, and the electric conductivity level was measured again ( $X_j$ ). The electrolyte extravasation was expressed as an initial conductivity percentage for the electric conductivity level after the treatment for 2 hours at 90 °C (Equation 4) (Scotti-Campos et al., 2013).

$$\frac{X_i}{X_c} \times 100$$
 (4)

The data were evaluated through an analysis of variance by the F test at a level of 1 and 5% of probability, and the regression analysis was applied for the irrigation water salt concentrations and  $H_2O_2$  levels with the assistance of the statistical software SISVAR version 5.3 (Ferreira, 2011).

#### 3. Results and Discussion

The interaction between the water's salinity and the peroxide dosage factors significantly affected (p < 0.01) all studied variables, except for stem diameter (SD), which presented a significant interaction at 5% (p < 0.05).

Table 2. Summary of the analysis of variance for the variables plant height (PH), leaf number (LN), stem
diameter (SD), root length (RL), leaf area (LA), root and aerial part ratio (R/AP) of yellow passion fruit
seedlings irrigated with water of different salinity levels for treatments with different H <sub>2</sub> O <sub>2</sub> levels. Pombal,
CCTA/UFCG, 2017

Mean Square							
SV	GL	PH	LN	SD	RL	LA	R/PA
Salinity (S)	4	2025.88**	26.73**	1.92**	129.02**	34719311.9**	0.0175**
Peroxide (P)	2	305.45**	17.50**	$0.40^{ns}$	19.91 <sup>ns</sup>	7018131.6**	0.027**
$\mathbf{S}\times\mathbf{P}$	8	318.22**	17.48**	0.51*	35.78**	8626645.9**	0.017**
Block	3	3.44 <sup>ns</sup>	2.14 <sup>ns</sup>	0.09 <sup>ns</sup>	8.52 <sup>ns</sup>	109471 <sup>ns</sup>	$0.00062^{ns}$
CV(%)		10.72	12.88	12.28	14.32	6.80	18.61
Mean		28.02	12.76	3.52	22.04	4865.37	0.258

*Note.* \*\* significant at 1% of probability (p < 0.01); \* significant at 5% of probability (p < 0.05); <sup>ns</sup> not significant (p > 0.05), Source of Variation (SV), Coefficient of variation (CV).

For plant height, the absence of peroxide and the use of 15  $\mu$ mol caused linear decreases that corresponded to 60.16 and 64.74%, respectively, between the greatest (4.3 dS m<sup>-1</sup>) and the lower (0.3 dS m<sup>-1</sup>) salinity level (Figure 1A). Nevertheless, the concentration of 5  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> presented a quadratic behavior, whose greater plant height value (28.05 cm) was obtained when the plants were irrigated with a water sample of 2.3 dS m<sup>-1</sup>. This adjustment under stress conditions at the concentration of 5  $\mu$ mol is probably caused by the activity increase of the ATPase enzyme, which incremented the relation of K to Na, resulting in adaptation to NaCl stress (Gupta & Huang, 2014).

Regarding stem diameter (Figure 1B), the absence of peroxide and the treatment with 15  $\mu$ mol caused the respective reductions of 5.82 and 7.80% in each unitary increase of the irrigation water's salinity. However, as noticed for plant height, the concentration of 5  $\mu$ mol presented a quadratic behavior over the increase of the irrigation water's salinity, with a maximum point (3.83 mm) at 2.3 dS m<sup>-1</sup>.

Nonetheless, the yellow passion fruit tree is very demanding in water terms and sensitive to salinity stress, with a maximum salinity of 1.3 dS m<sup>-1</sup> (Ayers & Westcot, 1999; Mesquita, Rebequi, Cavalcante, & Souto, 2012). Ergo, this behavior at the concentration of 5  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> can mitigate the depressive effect because, according to Kilic and Kahraman (2016), the H<sub>2</sub>O<sub>2</sub> reduced the salt-induced inhibition at germination, reducing the salinity stress' detrimental effects in the development of barley plants induced to pre-germinative treatment with peroxide at 30  $\mu$ mol.

The leaf sprouting of the yellow passion fruit seedlings was also affected by the interaction (salinity  $\times$  H<sub>2</sub>O<sub>2</sub>), which, at the absence of peroxide, was proven inversely proportional to the salt levels, with a reduction of 18 (0.3 dS m<sup>-1</sup>) to 10 leaves (4.3 dS m<sup>-1</sup>), which corresponds to a decrease of 44.71% (Figure 1C). On the other hand, the use of peroxide at the concentrations of 5 and 15 µmol resulted in a quadratic behavior, with a maximum leaf number point at the concentration of 1.7 and 1.5 dS m<sup>-1</sup>, with 12 and 14 leaves, respectively.

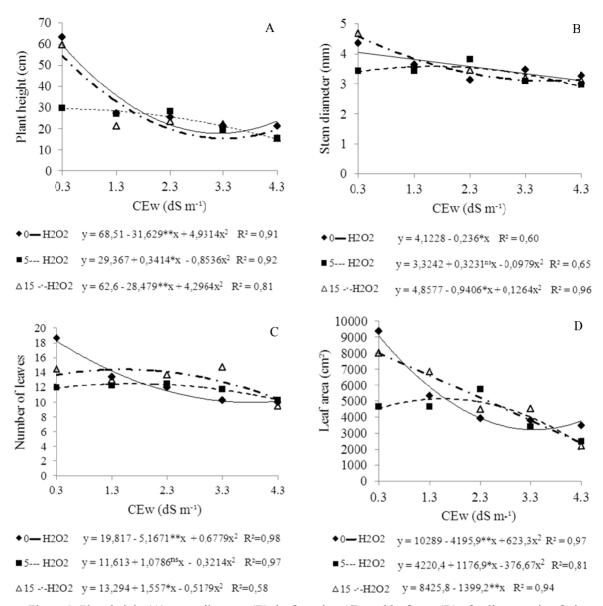


Figure 1. Plant height (A), stem diameter (B), leaf number (C), and leaf area (D) of yellow passion fruit seedlings irrigated with water samples of different electric conductivity levels for treatments with different H<sub>2</sub>O<sub>2</sub> concentrations. Pombal, CCTA/UFCG, 2017

The seedlings' leaf area decreased in the absence of  $H_2O_2$  and in the concentration of 15 µmol, with unitary reductions of 14.62 and 17.47% with the increase of the salinity levels, respectively (Figure 1D). However, at the  $H_2O_2$  concentration of 5 µmol, the leaf area increased from 4539.57 cm<sup>2</sup> under salinity (0.3 dS m<sup>-1</sup>) and reaching its greatest value of 5139.16 cm<sup>2</sup> at the electric conductivity of 1.6 dS m<sup>-1</sup>.

According to Taiz et al. (2017), the salt accumulation in the plants can interfere both with the leaf sprouting and expansion. Therefore, Ribeiro, Seabra Filho, Moreira, Souza, and Menezes (2013), studying the effect of salt water on the initial growth of yellow passion fruit seedlings, also reported an inversely proportional reduction of the number of leaves and of the leaf area to the salt level. We can infer that both the increase of salts and of  $H_2O_2$  caused reductions on the aerial part in response to the osmotic adjustment caused by stress conditions as a way to reduce water loss by transpiration. A similar effect was reported by Silva et al. (2016), who used concentrations of 15 µmol L<sup>-1</sup> upwards and got inferior values for the variables of plants not treated with  $H_2O_2$ . Nonetheless, something remarkable in this work was that the  $H_2O_2$  dosage of 5 µmol caused a tolerance increase for the passion fruit seedlings' leaf area.

Similarly, the root length (Figure 2A) was inversely proportional to the salt levels in the control treatment (0  $\mu$ mol of H<sub>2</sub>O<sub>2</sub>) and at the H<sub>2</sub>O<sub>2</sub> concentration of 15  $\mu$ mol equipotent to 14.53 and 39.86% between 4.3 and 0.3 dS m<sup>-1</sup>, as the H<sub>2</sub>O<sub>2</sub> concentration of 5  $\mu$ mol's maximum point was achieved at the concentration of 1.9 dS m<sup>-1</sup>, with a root length of 24.51 cm. Silva et al. (2016) restated that small concentrations of hydrogen peroxide act intracellularly, activating the vegetable's defense mechanisms and promoting tolerance, a statement that agrees with what was seen herein at the 5  $\mu$ mol peroxide concentration.

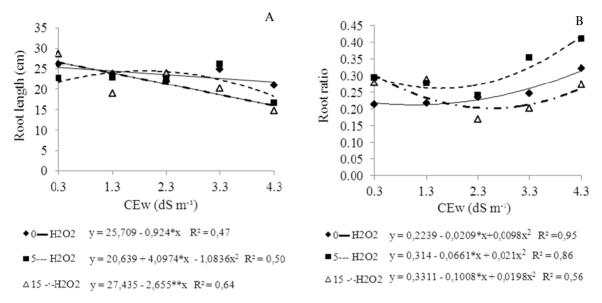


Figure 2. Root length (A) and root/aerial part ratio (B) of yellow passion fruit seedlings irrigated with water samples of different electric conductivity levels for treatments with different H<sub>2</sub>O<sub>2</sub> concentrations. Pombal, CCTA/UFCG, 2017

Consequently, the root/aerial part ratio was significantly afected by the interaction (peroxide and water salinity concentrations). We noticed a directly proportional increase of the root/aerial part ratio to the addition of salts in the water of the control treatment (11%), while the treatments with 5 and 15  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> increased the root/aerial part ratio after 2.3 dS m<sup>-1</sup> up to the greatest salinity level, which was greater in the treatment with 5  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> (53%) in comparison to the 15  $\mu$ mol treatment (29%) (Figure 2B).

Table 3 presents the interactive effect to all listed variables, except for the electrolyte extravasation (EE) variable, which affected the irrigation water salinity factor only when studied separately (p < 0.01).

Table 3. Summary of the analysis of variance for the variables leaf, stem, and root dry mass (LDM, SDM, and RDM), total dry mass (TDM), water relative level (WRL), electrolyte extravasation (EE), and Dickson quality index (DQI) of yellow passion fruit seedlings irrigated with water samples of different salinity levels for treatments with different  $H_2O_2$  concentrations. Pombal, CCTA/UFCG, 2017

Mean Square									
SV	GL	LDM	SDM	RDM	TDM	WRL	EE	DQI	
Salinity (S)	4	2.297**	0.936**	0.290**	8.778**	109.694**	122.21**	0.016**	
Peroxide(P)	2	0.598**	0.507**	0.100**	3.140**	51.630 <sup>ns</sup>	11.271 <sup>ns</sup>	0.0122**	
$\mathbf{S} \times \mathbf{P}$	8	0.469**	0.295**	0.075**	2.032**	65.512**	25.984 <sup>ns</sup>	0.0056**	
Block	3	$0.024^{ns}$	0.019 <sup>ns</sup>	0.012 <sup>ns</sup>	$0.142^{ns}$	17.607 <sup>ns</sup>	47.270 <sup>ns</sup>	0.0013 <sup>ns</sup>	
CV(%)		26.33	33.46	32.48	28.29	6.08	24.49	28.59	
Mean		1.071	0.573	0.403	2.048	74.123	17.398	0.167	

*Note.* \*\* significant at 1% of probability (p < 0.01); \* significant at 5% of probability (p < 0.05); <sup>ns</sup> not significant (p > 0.05), Source of Variation (SV), Coefficient of variation (CV).

The increase of the irrigation water's salinity significantly affected the phytomass build-up of the passion fruit plants, if we consider the leaf and stem dry mass (Figures 3A and 3B, respectively), in the absence of peroxide, as well as in the concentration of 15  $\mu$ mol of H<sub>2</sub>O<sub>2</sub>, which presented inversely proportional decreases to the water's salinity. These decreases were of 63.04 and 80.97% for the leaves, and of 57.61 and 81.97% for the measure of the stems. However, the concentration of 5  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> increased the leaf dry mass in 36.39% up to 2 dS m<sup>-1</sup>, and the stem dry mass increased 12.76% up to the salt concentration of 1.5 dS m<sup>-1</sup>.

We can infer that hydrogen peroxide has some features that allow it to act as a secondary messenger: I) its production is easily ruled by different stimuli, especially through the NADPH-oxidases and peroxides; II) it is a small and relatively mobile molecule, able to transport information between different cell compartments; III) it can modulate flag activities of other components and reaction cascades, with different biological results, including the ones that cause its own synthesis (Petrov & Van Breusegem, 2012), because, through the evaluated variables, we can notice that the application of peroxide up to the dosage of 5  $\mu$ mol promotes greater tolerance in plants irrigated with water samples of up to 2.3 dS m<sup>-1</sup>.

According to Schossler, Machado, Zuffo, Andrade, and Piauilino (2012), higher sodium concentrations also affect the translocation and synthesis of hormones from the roots to the aerial part, which causes leaf area and, consequently, aerial part dry phytomass loss. Similar results are presented by Mesquita et al. (2012) and Ribeiro et al. (2013), who employed greater salt concentrations in the irrigation water of yellow passion fruit seedlings.

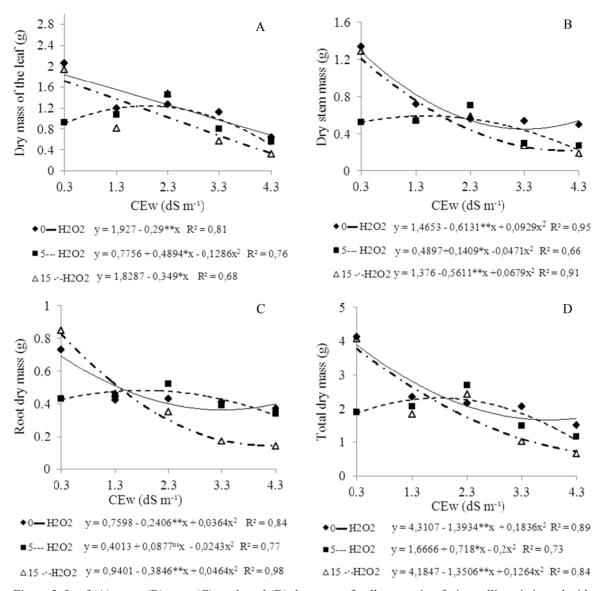
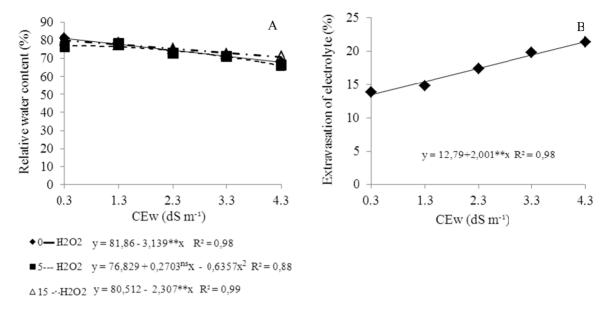


Figure 3. Leaf (A), stem (B), root (C), and total (D) dry mass of yellow passion fruit seedlings irrigated with saline water samples of different electric conductivity levels for treatments with different H<sub>2</sub>O<sub>2</sub> concentrations. Pombal, CCTA/UFCG, 2017

Corroborating, the root (RDM) and total dry mass (TDM) presented a similar behavior with reductions at each saline unitary increase of 10.58 and 20.65% for RDM and 14.03 and 20.18% for TDM at the concentrations of 0 and 15  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> (Figures 3C and 3D, respectively). However, the concentration of 5  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> increased both the root and total phytomass up to the salt concentration of 1.8 dS m<sup>-1</sup>, corresponding to 13 and 24%, respectively.

As for the water relative level in the plant, regardless of the  $H_2O_2$  concentrations, the water relative content (WRC) was proven inversely proportional to salinity (Figure 4A), with reductions of 15.52, 13.80, and 11.56% at the concentrations of 0, 5, and 15 µmol of  $H_2O_2$ , respectively, to the point in which they decreased from 80.9, 76.8, and 79.82% to 68.36, 66.23, and 70.59%, respectively, in relation to the greater (4.3 dS m<sup>-1</sup>) and lower (0.3 dS m<sup>-1</sup>) salinity level.

However, in spite of the inversely proportional decrease of the water content to the amount of salt, regardless of the use and concentrations of peroxide, we noticed that, with the peroxide concentration was inversely proportional to the damages caused by the salinity stress, since the concentration of 15 µmol presented a lower reduction in comparison with all other treatments. Dias and Blanco (2010) state that one of the conditions



offered by salinity is the reduction of water availability to the plants through the osmotic effect, which restrains the water's efficiency. However, the application of peroxide (5 and 15 µmol) favors such efficiency.

Figure 4. Water relative level (A) and electrolyte extravasation (B) for yellow passion fruit seedlings irrigated with saline water samples of different electric conductivity levels for treatments of different H<sub>2</sub>O<sub>2</sub> concentrations. Pombal, CCTA/UFCG, 2017

As for electrolyte extravasation (Figure 4B), the addition of salts in the irrigation water cause a linear increase of the EE, with an increase of 53.55% between the greater and the lower salinity level, which increased from 13.93 at the lower salinity level (0.3 dS m<sup>-1</sup>) to 21.39 in the plants irrigated with the greater salinity level (4.3 dS m<sup>-1</sup>). For Souza (2014), a salinity-tolerant vegetable controls its ion movement through the plasma membrane and through the tonoplast to maintain a low Na<sup>+</sup> concentration in the cytoplasm. We understand that damages caused to the cell membrane, indicated by the increase of electrolyte extravasation, results in hints of cell death caused by the excess of salts therein.

As it happens with the seedlings' phenological growth, the quality thereof was significantly affected by the interaction between the water's salinity and the hydrogen peroxide's foliar application, which resulted, both in the control treatment and with the use of 15  $\mu$ mol of H<sub>2</sub>O<sub>2</sub>, in decreases of the Dickson quality index to 20.67 and 73.32%, respectively, which were registered between the greater (4.3 dS m<sup>-1</sup>) and lower (0.3 dS m<sup>-1</sup>) salinity level (Figure 5). However, although the peroxide concentration of 15  $\mu$ mol does not mitigate the depressive effects of salinity, the concentration of 5  $\mu$ mol of H<sub>2</sub>O<sub>2</sub> increased the seedlings' quality in 36% up to the salinity level of 2.2 dS m<sup>-1</sup>, which declined afterwards.

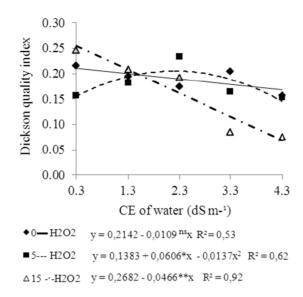


Figure 5. The Dickson quality index of yellow passion fruit seedlings irrigated with saline water samples of different electric conductivity levels for treatments of different H<sub>2</sub>O<sub>2</sub> concentrations. Pombal, CCTA/UFCG, 2017

Furthermore, the increase of Dickson quality index values quantify the seedlings' quality standard (Costa, Durante, Nagel, Ferreira, & Santos, 2011). Similar results related to salinity increase in yellow passion fruit seedlings that negatively affect the quality index were also found by Medeiros et al. (2016). Nonetheless, the use of peroxide at 5  $\mu$ mol in passion fruit seedlings probably made them insensitive to salinity (> 1.3 dS m<sup>-1</sup>), or moderately sensitive thereto (1.3-3 dS m<sup>-1</sup>), according to the results obtained herein (Ayers & Westcot, 1999).

#### 4. Conclusions

The hydrogen peroxide concentration of 5  $\mu$ mol/L mitigated the harmful effect of salts up to 2 dS m<sup>-1</sup>.

Hydrogen peroxide mitigates the harmful effect of salinity. However, excessive concentration (15 µmol) are harmful to yellow passion fruit seedlings.

The irrigation water's salinity affects the phenological growth and the quality of yellow passion fruit seedlings.

#### 5. Referências

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