

# Synthesis and Characterization of New 8-trifluoromethyl Quinazolin-2,4-(3*H*)-Dione Nucleosides

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## Abstract

Synthesis of 8-trifluoromethyl quinazolin-2,4-(1*H*,3*H*)-dione **2**, which have been ribosylated by coupling with 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- $\beta$ -D-ribofuranose **4** by using the silylation method, afforded mixture  $\beta$ - and  $\alpha$ -anomeric of the benzoylated nucleoside derivatives **5** and **6**, respectively. Debenzoylation of each of **5** and **6** by sodium metal in dry methanol to afford the corresponding free nucleosides **7** and **8** respectively. The structures of the newly synthesis compounds have been confirmed on the basis of elemental analyses, IR, <sup>1</sup>HNMR, <sup>13</sup>CNMR and Mass spectral data.

**Keywords:** 1-*O*-Acetyl-2,3,5-trihydroxy- $\beta$ -D-ribofuranose, Nucleosides, Quinazolin-2,4-(1*H*,3*H*)-dione, Trifluoromethyl

## 1. Introduction

Quinazolinone is a heterocyclic compound that occupies a distinct and place in the field of medicinal chemistry. Many of them were showed antimicrobial, anti-inflammatory, anticonvulsant, analgesic and anticancer agents (Safinaz E. Abbas et al, 2013; A. Kumar et al, 2011; K.M. Amin et al, 2010; M.M. Aly, 2010; A. Kumar, 2003; S.T. Al-Rashood, 2006 and N. Mulakayala, 2012).

Quinazoline nucleosides were first synthesized by Stout and Robins in 1968 as pyrimidine nucleoside analogs (Stout M. G and Robins R. K., 1968) and consequent synthetic studies were contributed by Dunkel and Pfeleiderer in the 1990s. (Dunkel M and Pfeleiderer W, 1991, 1992 and 1993). More recently, several quinazoline-2,4-dione nucleosides have been incorporated into oligonucleotides as pyrimidine nucleoside substitutes to study the binding affinity and base pairing selectivity (Michel J et al, 1997; Diwan A. R., 1969; T. C. Chien, 2005; F. E. M. El-Baih, 2004; Tun-Cheng CHIEN, 2004).

Many familiar drugs and pharmacological studies contain trifluoromethyl groups. Quinazoline-2,4-diones bearing a trifluoromethyl group derivatives were an inhibitor of human immunodeficiency virus-1 reverse transcriptase, antagonists at ionotropic glutamate receptors (Hao Chen et al, 2003; Vittoria Colotta, 2012) and anticancer compound trifluoromethyl-substituted pyrazole N-nucleoside (Ayman M. Saleh et al, 2016).

In this review, quinazolin-2,4-(3*H*)-dione nucleosides containing trifluoromethyl group were designed as part of our continuing interest in the synthesis of new nucleosides as expected their biological activity.

## 2. Material and Methods

Melting points were measured on Gallenkamp melting point apparatus (UK) and are uncorrected. The purity of the compounds was checked by thin layer chromatography (TLC). Thin layer chromatography (TLC) was performed on silica gel sheets F1550 LS 254 of Schleicher & Schull and column chromatography on Merck silica gel 60 (particle size 0.063–0.20). Elemental analyses were obtained on an Elementary Vario EL 1150C analyzer. IR spectra were recorded on KBr discs on Fourier Transform infrared and Pie Unicom SP 300 Infrared Spectrophotometers at Taif University. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were obtained on a Varian (850 MHz) EM 390 USA instrument at King Abdel-Aziz University by using TMS as the internal reference. Mass spectra were recorded on a JEOL-JMS-AX500 at King Abdel-Aziz University, Saudi Arabia.

## 3. Experimental

### 8-trifluoromethyl quinazolin-2,4-(1*H*,3*H*)-dione **2**

2-Amino-3-trifluoromethyl benzoic acid **1** (Aldrich; 0.019mol, 4g) was added drops of acetic acid in (100 ml) water,

the solution of KNCO (0.049mol, 4g) was dropped to the mixture was stirred in an ice bath at 1h. The reaction mixture was added sodium hydroxide (10 g) and overnight at room temperature and then filtered. The precipitate was neutralized with dilute sulfuric acid (1:1) and washed (3×20 ml) water. The compound purified by column chromatography on silica gel with (Ethylacetate: Acetone 9:1) to afford white crystals.

Yield (63.23%), w. 6.1g, m.p. <300°C white ;  $\nu$  (cm<sup>-1</sup>) (KBr) 3400, 3075,1740, 1640; <sup>1</sup>HNMR (850MHz); (DMSO-D<sub>6</sub>): 11.62 (s, 1H) NH-1, 11.20 (s,1H) NH-3, 8.22 (s, 1H) H<sub>5</sub>, 7.79 (d, 1H, *J* = 7.5 Hz) H<sub>7</sub>, 7.18 (d, 1H, *J* = 7.5 Hz) H<sub>6</sub>. <sup>13</sup>CNMR (850MHz):  $\delta$  167.27, 154.77, 133.08, 130.82,128.39, 125.71, 124.15, 123.09, 109.80; MS m/z: 230 (M<sup>+</sup>, 45%). Anal. Calcd. for C<sub>9</sub>H<sub>5</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub>; M.wt: 230.14; C,46.97; H,2.19; F,24.77; N, 12.17 (%); Found: C, 46.26; H, 2.89; F,24.01; N,11.05 (%).

### Synthesis of protection nucleoside of 8-trifluloromethyl quinazolin-2,4-(1*H*,3*H*)-dione 2

#### General Procedure

Silylation of 8-trifluloromethyl quinazolin-2,4-(1*H*,3*H*)-dione **2** (0.021 mol) with hexamethyldisilazane (HMDS) (20 ml) was refluxed for 3days with a catalytic a few crystals of ammonium sulfate under exclusion of moisture. Excess of HMDS was removed in vacuo by co-evaporation with dry dichloroethane gave the silylated derivative **3**, using the Vorbruggen's silylation method (Vorbruggen et al, 1981). The residue was dissolved in 20 ml of dry 1,2-dichloroethane and then 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- $\beta$ -D-ribofuranose **4** (10.8 g, 0.021 mol) was added. The mixture was added dropwise onto a mixture (4.5ml) of (10 ml trimethylsilyl trifluoromethane sulfonate (TMSOTf) in dry 1,2-dichloroethane (50 ml)). The mixture was stirred at room temperature for 24 h, and then washed with a saturated solution of aqueous sodium bicarbonate (3 × 50 ml), washed with water (3 × 50 ml), and dried over anhydrous sodium sulfate. the organic phase was extracted by CH<sub>2</sub>Cl<sub>2</sub>, dried over MgSO<sub>4</sub> and evaporated. The solvent was removed under vacuum gave an anomeric mixture of  $\beta$  and  $\alpha$ -1-(2,3,5-tri-*O*-benzoyl-D-ribofuranosyl)-8-trifluloromethylquinazolin-2,4-(3*H*)-dione. The protected nucleoside was separated by column chromatography on silica gel with dichloromethane: acetone (9:1) as eluent to afford a white crystal pure  $\beta$ -anomeric **5** and  $\alpha$ -anomeric **6** respectively, in good yields.

#### $\beta$ -1-(2,3,5-Tri-*O*-benzoyl-D-ribofuranosyl)-8-trifluloromethyl quinazolin-2,4-(3*H*)-dione 5

Yield (52.12%), w. 2.7 g, m.p. 119°C white; IR  $\nu$  (cm<sup>-1</sup>) (KBr) 3042, 1725, 1680; <sup>1</sup>HNMR (850MHz); (CDCl<sub>3</sub>):  $\delta$  9.47(s, 1H) H<sub>3Amide</sub>, 8.26 (d, 1H, *J* = 7.5 Hz) H<sub>5</sub>, 8.03 (d, 1H, *J* = 7.2 Hz) H<sub>7</sub>, 7.92 (dd, 1H, *J* = 15.7 Hz) H<sub>6</sub>, 7.51-7.25 (m, 15H) H<sub>(Ar-H)</sub>, 6.48 (d, 1H, *J* = 7.5 Hz) H<sub>1</sub>, 6.16 (dd, 1H, *J* = 8.4 Hz) H<sub>2</sub>, 6.09 (t, 1H, *J* = 13.4 Hz) H<sub>3</sub>, 4.82-4.77 (dd, 1H, *J* = 4.6 Hz) H<sub>5</sub>, 4.69-4.56 (m, 1H) H<sub>4</sub>. <sup>13</sup>CNMR (850MHz) (CDCl<sub>3</sub>):  $\delta$  166.30, 165.46, 164.87,165.53<sub>C=O's groups</sub>, 148.17C<sub>4</sub>, 146.78 C<sub>2</sub>, 133.61-128.33<sub>Ar-carbons</sub>, 119.11 CF<sub>3</sub>, 88.02 C<sub>1</sub>, 79.47 C<sub>2</sub>,73.93 C<sub>3</sub>, 70.98 C<sub>4</sub>, 63.82 C<sub>5</sub><sub>sugar carbons</sub>. Anal. Calcd. for C<sub>35</sub>H<sub>25</sub>F<sub>3</sub>N<sub>2</sub>O<sub>9</sub>; M.wt: 674.58; C,62.32; H,3.74; F,8.45; N, 4.15; (%); Found: C, 62.26; H, 3.89; F,8.21; N,3.95 (%).

#### $\alpha$ -1-(2,3,5-Tri-*O*-benzoyl-D-ribofuranosyl)-8-trifluloromethyl quinazolin-2,4-(3*H*)-dione 6

Yield (47.87%), w. 2.48 g, m.p. 105-107°C white color; IR  $\nu$  (cm<sup>-1</sup>) (KBr) 3020, 1725, 1685; <sup>1</sup>HNMR (850MHz) (CDCl<sub>3</sub>):  $\delta$  9.16 (s, 1H) H<sub>Amide</sub>, 8.06 (d, 1H, *J* = 7.4 Hz) H<sub>5</sub>, 7.98 (d, 1H, *J* = 7.2 Hz) H<sub>7</sub>, 7.89 (d, 1H, *J* = 7.4 Hz) H<sub>6</sub>, 7.55-7.25 (m, 15H) H<sub>(Ar-H)</sub>, 6.73 (d, 1H, *J* = 2.2 Hz) H<sub>1</sub>, 6.14 (dd, 1H, *J* = 8.8 Hz)H<sub>2</sub>, 6.03-5.89 (m, 1H) H<sub>3</sub>, 4.81-4.57 (m, 1H ) H<sub>4</sub>, 3.73 (m, 1H) H<sub>5</sub>. <sup>13</sup>CNMR (850MHz):  $\delta$  166.46,165.82, 165.40<sub>C=O's groups</sub>, 147.95C<sub>4</sub>, 147.26 C<sub>2</sub>, 13.73-128.40<sub>Aromatic carbons</sub>, 118.01 CF<sub>3</sub>, 87.46 C<sub>1</sub>, 79.37 C<sub>2</sub>,73.90 C<sub>3</sub>, 71.69 C<sub>4</sub>, 63.83 C<sub>5</sub>. (CH<sub>3</sub>COOCH<sub>2</sub>CH<sub>3</sub>) (9:1). Anal. Calcd. for C<sub>35</sub>H<sub>25</sub>F<sub>3</sub>N<sub>2</sub>O<sub>9</sub>; M.wt: 674.58; C,62.32; H,3.74; F,8.45; N, 4.15; (%); Found: C, 62.15; H, 3.45; F,8.91; N,4.09 (%).

Deprotection of **5** and **6**. Synthesis of free nucleosides **7** and **8** respectively

#### General Procedure

The pure anomer of each  $\beta$  **5** and  $\alpha$  **6** (0.001 mol for each), dry absolute methanol (20 ml) and sodium metal (0.055 g, 0.001mol) was stirred at room temperature for 48h. The solvent was evaporated under vacuum to give a colorless solid, which was dissolved in hot water and neutralized with few drops acetic acid. Purification of each compound by TLC chromatographic on silica gel with chloroform: ethyl acetate (9: 1) to afford colorless and white crystals of the following Zemplen et al.'s method (Zemplen et al, 1939) to afford the free nucleosides **7** and **8**, respectively.

#### $\beta$ -1-(2,3,5-Trihydroxy-D-ribofuranosyl)-8-trifluloromethyl quinazolin-2,4-(3*H*)-dione 7

Yield (81.67%), w. 0.307g. m.p. 185°C white color; IR  $\nu$  (cm<sup>-1</sup>) (KBr) 3450, 3032, 1715, 1685; <sup>1</sup>HNMR (600MHz)(DMSO-D<sub>6</sub>):  $\delta$  11.59 (s, 1H) H<sub>1Amide</sub>, 8.06 (d, 1H, *J* = 5.5 Hz) H<sub>5</sub>, 7.80-7.79 (d, 1H, *J* = 8.7 Hz) H<sub>7</sub>, 7.77-7.76 (d, 1H, *J* = 5.2 Hz) H<sub>6</sub>, 6.17 (d, 1H, *J* = 7.5 Hz) H<sub>1</sub>, 5.27 (s, 1H) H<sub>2</sub>, 5.07 (m, 1H) H<sub>3</sub>, 4.45 (t, 1H) H<sub>5</sub>, 4.13 (s, 1H) H<sub>4</sub>, 3.80-3.76 (m, 1H) H<sub>2OH</sub>, 3.66-3.61 (m, 1H) H<sub>3OH</sub>, 3.58-3.41 (m, 1H) H<sub>3OH</sub>. <sup>13</sup>C NMR: 155.07 C<sub>4</sub>, 153.63 C<sub>2</sub>, 139.6, 137.59, 133.73, 129.92, 128.86, 120.00, 118.96 CF<sub>3</sub>, 89.89 C<sub>1</sub>, 85.31 C<sub>2</sub>, 69.41 C<sub>3</sub>, 69.29 C<sub>4</sub>, 61.63 C<sub>5</sub>.

(CH<sub>3</sub>COOCH<sub>2</sub>CH<sub>3</sub>: Acetone) (9:1); MS m/z: 362 (M<sup>+</sup>, 21%). Anal. Calcd. for C<sub>14</sub>H<sub>13</sub>F<sub>3</sub>N<sub>2</sub>O<sub>6</sub>; M.wt: 362.26; C,46.42; H,3.62; F,15.73; N, 7.73; (%); Found: C, 46.12; H, 3.75; F,15.91; N,7.24 (%).

#### ***α*-1-(2,3,5-Trihydroxy-D-ribofuranosyl)-8-trifluoromethyl quinazolin-2,4-(3*H*)-dione 8**

Yield (94.73%), w. 0.41 g, m.p. 220°C white color; IR  $\nu$  (cm<sup>-1</sup>) (KBr) 3450, 1714, 1690; <sup>1</sup>H NMR (600MHz)( DMSO-D<sub>6</sub>):  $\delta$  11.85 (s, 1H) H<sub>1Amide</sub>, 8.06 (d,1H, *J* = 1.46 Hz)H<sub>5</sub>, 7.80 (t, 1H, *J* = 11.37 Hz) H<sub>7</sub>, 7.77 (d, 1H, *J* = 1.83 Hz) H<sub>8</sub>, 6.53 (d, 1H, *J* = 4.5 Hz) H<sub>1'</sub>, 5.25 (d, 1H, *J* = 5.87 Hz) H<sub>2'</sub>, 5.05 (t, 1H, *J* = 21.65 Hz) H<sub>3'</sub>, 4.46 (q, 1H) H<sub>5'</sub>, 4.13 (q, 1H) H<sub>4'</sub>, 3.8 (m, 2H) H<sub>4'a</sub>, 3.60 (s, 1H) OH<sub>2</sub>, 3.58 (s, 1H) OH<sub>3</sub>, 3.57 (s, 1H) OH<sub>5</sub>. <sup>13</sup>C NMR: 161.07 C<sub>4</sub>, 150.63 C<sub>2</sub>, 139.67, 137.59, 133.73, 129.93, 125.34, 120.00, 118.82 CF<sub>3</sub>, 89.91 C<sub>1'</sub>, 85.32C<sub>2'</sub>, 69.42 C<sub>3'</sub>, 69.29 C<sub>4'</sub>, 61.63 C<sub>5'</sub>. (CH<sub>3</sub>COOCH<sub>2</sub>CH<sub>3</sub>: Acetone) (9:1); MS m/z: 362 (M<sup>+</sup>, 36%). Anal. Calcd. for C<sub>14</sub>H<sub>13</sub>F<sub>3</sub>N<sub>2</sub>O<sub>6</sub>; M.wt: 362.26; C,46.42; H,3.62; F,15.73; N, 7.73; (%); Found: C, 46.12; H, 3.75; F,15.57; N,7.13 (%).

#### **4. Results and Discussion**

The structures of the products **2-8** were established and confirmed on the bases of their elemental analyses and spectral data (IR, <sup>1</sup>H and <sup>13</sup>C NMR) (see the Experimental section)(Scheme 1). Thus, their <sup>1</sup>H NMR spectra of compound **2** showed doublet signals at assigned to the aromatic protons of H-5 H-6 and H-7 and two a singlet signal of amide NH-3 and NH-1.

<sup>1</sup>H NMR spectra of **5** and **6** showed in each case a doublet signals at  $\delta$  6.48 (d, 1H, *J* = 7.5 Hz) H<sub>1'</sub> for compound **5** and at  $\delta$  6.73 (d, 1H, *J* = 2.2 Hz) H<sub>1'</sub> for compound **6** assigned to the anomeric proton of the ribose moiety with spin-spin coupling constant (*J*<sub>1,2'</sub>) equal to 7.5 Hz, which confirms the  $\beta$ -anomeric configuration. While confirms the -anomeric configuration showed spin-spin coupling constant (*J*<sub>1,2'</sub>) equal to 2.2 Hz, which confirms the -anomeric configuration for compound **8** (Break et al, 2014; Break et al, 2013; Break & Mosselhi, 2012; Mosselhi & Break, 2011 and Break et al, 2010 and Abdullah Hijazi, 1988). The <sup>1</sup>H NMR spectra of nucleosides free showed a doublet signals at  $\delta$  6.17 for compound **7** spin-spin coupling constant (*J*<sub>1,2'</sub>) equal to 7.5 Hz -anomeric configuration and at  $\delta$  6.53 for compound **8** assigned to spin-spin coupling constant (*J*<sub>1,2'</sub>) equal to 4.5 Hz, which confirms the -anomeric configuration. The <sup>1</sup>H NMR of compounds **5** and **6** showed the expected base moiety protons in addition to the sugar moiety protons (see the Experimental section).

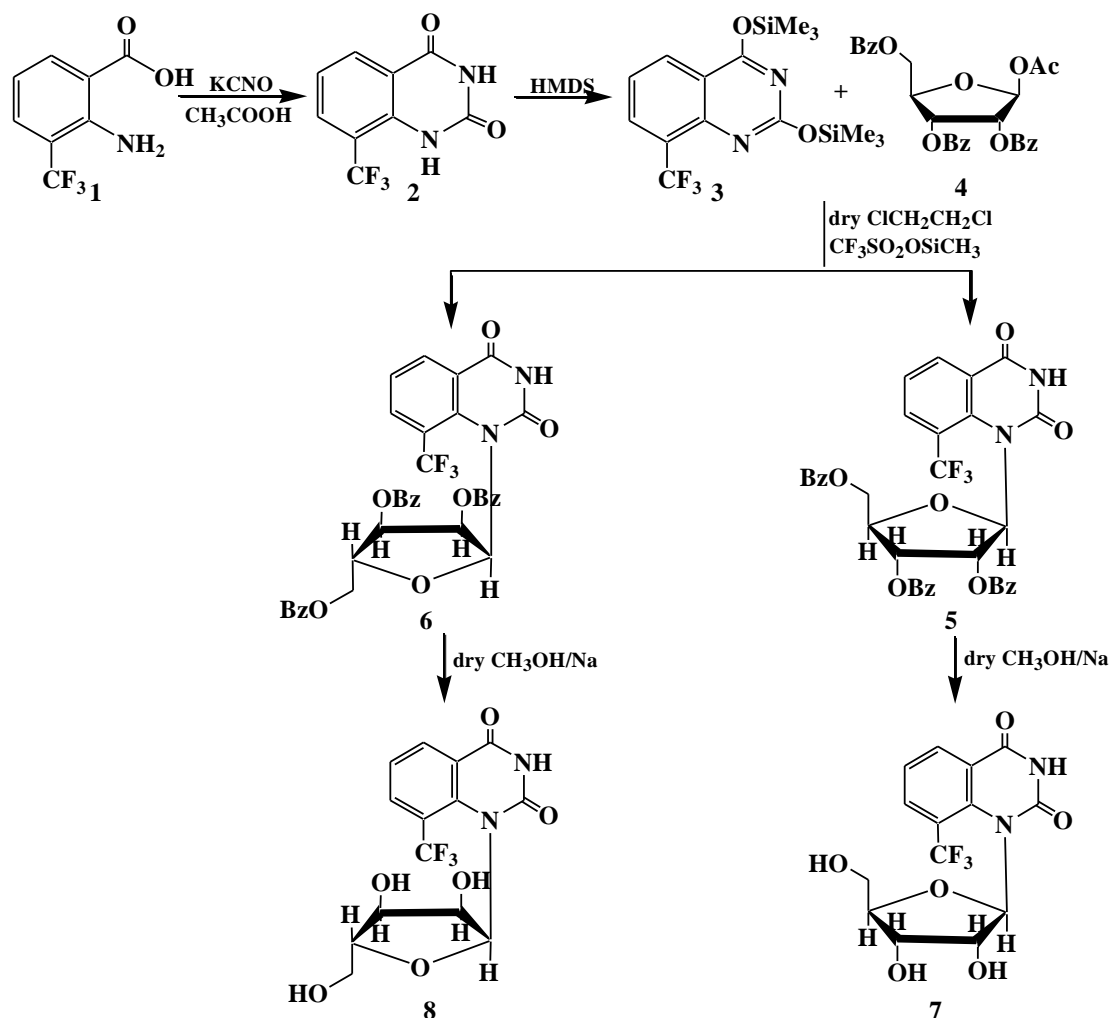
The <sup>13</sup>C NMR of nucleoside products revealed the signals are due to the three benzoyl carbonyl groups at and for compound **5**, and 166.46, 165.82 and 165.40 for compound **6**, while showed the two signals of amide carbons at 148.17, 146.78 for compound **5**, and at 147.95C<sub>4</sub>, 147.26 C<sub>2</sub> for compound **6**, The twenty one signals at 133.61-128.33 and at 133.73-128.40 Aromatic carbons for compound **5** and **6** respectively.

The five signals were assigned to C-1', C-2', C-3', C-4', and C-5' of the sugar moiety, at  $\delta$  88.02 C<sub>1'</sub>, 79.47 C<sub>2'</sub>, 73.93 C<sub>3'</sub>, 70.98 C<sub>4'</sub> and 63.82 C<sub>5'</sub> for compound **5**, at  $\delta$  87.46 C<sub>1'</sub>, 79.37 C<sub>2'</sub>, 73.90 C<sub>3'</sub>, 71.69 C<sub>4'</sub>, 63.83 C<sub>5'</sub> for compound **6**, at  $\delta$  89.89 C<sub>1'</sub>, 85.31 C<sub>2'</sub>, 69.41 C<sub>3'</sub>, 69.29 C<sub>4'</sub>, 61.63 C<sub>5'</sub> for compound **7** and at  $\delta$  89.91 C<sub>1'</sub>, 85.32C<sub>2'</sub>, 69.42 C<sub>3'</sub>, 69.29 C<sub>4'</sub>, 61.63 C<sub>5'</sub> for compound **8**.

The <sup>13</sup>C NMR of CF<sub>3</sub> group showed at  $\delta$  119.11, 118.01, 118.96 and 118.82 of compounds (**5**, **6**, **7** and **8**) respectively Break and Break. The IR spectrum of compounds **5** and **6** showed the stretching vibration frequencies of the carbonyl C=O groups at 1725 cm<sup>-1</sup>. IR spectra of compounds **7** and **8** showed absorptions around 3450 cm<sup>-1</sup> for (OH) and 1715 cm<sup>-1</sup> for (C=O).

#### **5. Conclusion**

Quinazolinone nucleosides are scientific importance in many biologically active compounds. So synthesis and characterization of 8-trifluoromethyl quinazolin-2,4-(1*H*,3*H*)-dione **2**. Ribosylation of compound **3** with 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- $\beta$ -D-ribofuranose **4** afforded mixture  $\beta$ - and  $\alpha$ -anomeric of the benzoylated nucleoside derivatives **5** and **6**, respectively. Debenzoylation of the latter affording the corresponding new free N-nucleosides **7** and **8**, respectively. Nucleosides obtained have been identified by their spectral analysis.



Schem (1). 8-Trifluoroquinazolin-2,4-(3H)-dione Nucleosides

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