

Synthesis of the Novel 3-Benzotriazole-5-yl difluoromethyl-5-trifluoromethyl benzotriazole Nucleosides

Laila M. Break

Correspondence: Laila M. Break, Taif University, Al-Haweiah, P.O.Box 888 Zip Code 21974. Taif, Saudi Arabia
Tel: 966-568480801. E-mail: lailabreak@hotmail.com

Received: July 22, 2015 Accepted: August 6, 2015 Online Published: September 9, 2015

doi:10.5539/ijc.v7n2p99

URL: <http://dx.doi.org/10.5539/ijc.v7n2p99>

Abstract

Triazole ring is a quite important five-membered heterocycle with three nitrogen atoms, possesses aromaticity and is an electron rich system. Triazole derivatives have been frequently becoming clinical drugs or candidates for the treatment of various types of diseases. Synthesis of the novel of 3-Benzotriazole-5-yl difluoromethyl-5-trifluoromethyl benzotriazole compound (3). Synthesis and characterization of two new benzotriazole nucleosides with good yields by silylation method.

Keywords: 1-O-Acetyl-2,3,5-trihydroxy- β -D-ribofuranose, Nucleosides, Trifluoromethyl. Reduction, 3-Benzotriazole-5-yl difluoromethyl-5-trifluoromethyl benzotriazole

1. Introduction

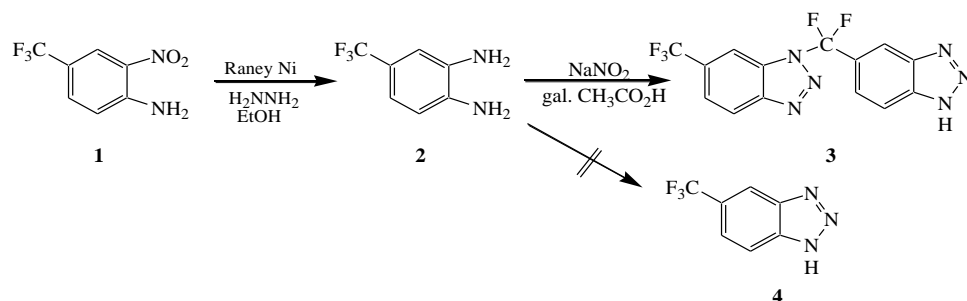
More and more triazole derivatives, with strong pharmacological activity, low toxicity, less adverse effects, fewer multi-drug resistances, high bioavailability, good pharmacokinetics property and drug-targeting, diversity of drug administration, broad spectrum, better curative effect etc., A large number of triazole compounds as clinical drugs or candidates have been frequently employed for the treatment of various types of diseases, Triazole compounds as medicinal drugs, including antifungal, anticancer, antibacterial, antitubercular, antiviral, anti-inflammatory and analgesic, anticonvulsant, antiparasitic, antidiabetic, anti-obesitic, antihistaminic, anti-neuropathic, antihypertensive as well as other biological activities (Zhou & Wang, 2012 and Khabnadideh et al, 2012). In fact several benzotriazoles have shown potential biological activities such as antimycobacterial (Sanna et al, 2002), antitumor (Handratta et al, 2005) and anti-inflammatory (Dawood et al, 2006) activities.

In recent years, the use of antifungal drugs in human medicine has increased, especially with the advent of AIDS epidemic. Efforts have focused on the development of new, less toxic and more efficacious antifungal drugs with novel mechanism of action (Khabnadideh et al, 2012). Biological activities of ribavirin and homo-N-nucleosides, a novel 1,2,4-triazole nucleoside drug derivative as a potential antiviral agent and anticancer activities (Chun et al, 2005; Xia et al, 2010; Konstantinova et al, 2013; Mosselhi & Neidlein, 2009 and Kristinsson et al, 1994).

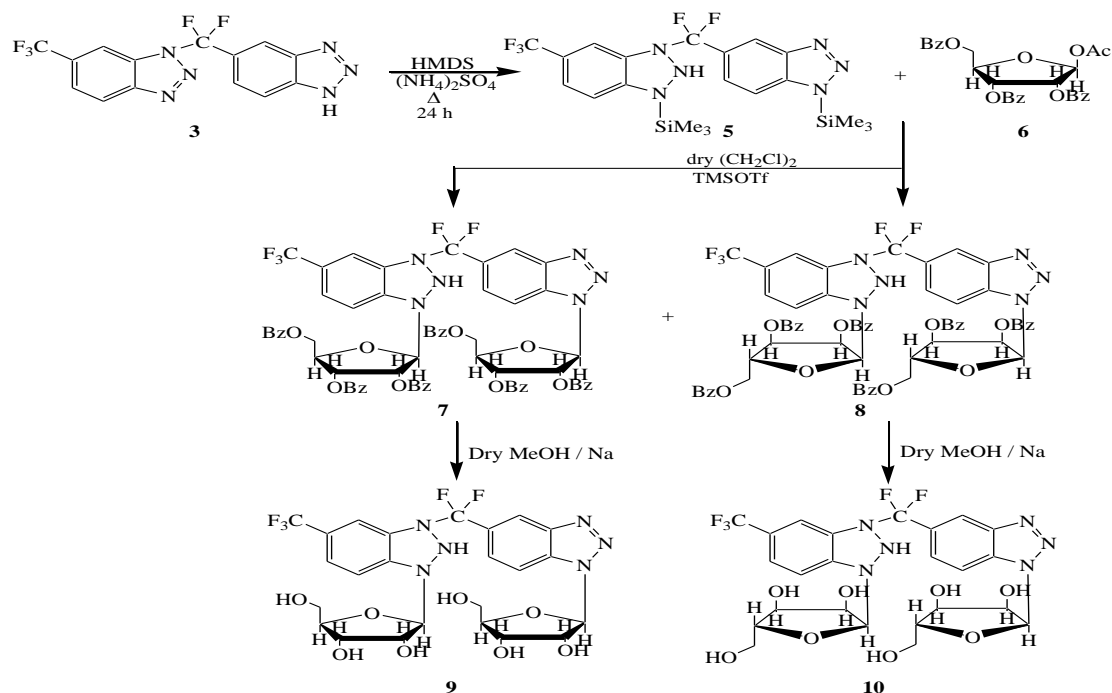
The purpose of this study was to synthesize of some new nucleoside benzotriazole and study their characterization.

2. Material and Methods

All chemicals were supplied by Sigma-Aldrich and Merck (Germany). IR spectra were recorded for KBr discs on Fourier Transform infrared and Pye Unicam SP 300 Infrared Spectrophotometers at Taif University. ¹H NMR spectra were obtained on a Varian (850 MHz) EM 390 USA instrument at King Abdel-Aziz University by using TMS as internal reference. ¹³C NMR spectra were recorded on a JNM-LA spectrometer (850 MHz) at King Abdel-Aziz University, Saudi Arabia. Elemental analyses were obtained on an Elementar Vario EL 1150C analyzer. Mass spectra were recorded on a JEOL-JMS-AX500 at King Abdel-Aziz University, Saudi Arabia. Purity of the compounds was checked by thin layer chromatography (TLC). Thin layer chromatography (TLC) was performed on silica gel sheets F1550 LS 254 of Schleicher & Schull and column chromatography on Merck silica gel 60 (particle size 0.063–0.20. Melting points were measured on Gallenkamp melting point apparatus (UK) and are uncorrected. The starting materials 4-Amino-3-nitrobenzotrifluoride (1) was prepared Pubchem CID: 67865.



Scheme (1): Synthesis of 3-Benzotriazole-5-(yl)- difluoromethyl-5-trifluoromethyl benzotriazole



Scheme (2): Synthesis of 3-Benzotriazole-5-(yl)-difluoro-methyl-5-tri-fluoromethyl benzotriazole Nucleosides

3. Experimental

The method for reduction of aromatic nitro compounds employing NaBH_4 / Raney nickel system or Raney nickel only is simple, inexpensive, easily scaled-up and applicable for large scale preparation of different heterocyclic aryl amines (Chang et al, 1970; Mi et al, 2008; Mi et al, 2007 and Wang & Zhou, 2011).

3,4-Diaminobenzotrifluoride (2)

A mixture of 4.46 g. (0.0216 mole) of 4-Amino-3-nitrobenzotrifluoride (1), 50 ml. of absolute ethanol, and 1g. of Raney nickel and 5 ml of hydrazine hydrate was added. the temperature is rapidly raised to 60–70 °. Owing to the strong exothermic reaction. The reaction mixture was refluxed at 100–120 ° for 3h. After the reaction mixture has cooled, the catalyst is separated from the reaction mixture by filtration through a Büchner funnel. The alcohol and water are removed by distillation, and the product was purified by flash column chromatography (silica gel 100 g, chloroform /ethyl acetate 90: 10) to give 3.3 g (Ayyangar et al, 1984; Dauben et al, 1963 and Allen & James, 1955).

Yield (87%), m.p. 56 oC brown; IR (KBr) ν cm^{-1} : 3400 (NH₂). ¹HNMR (600MHz); (CD₃OD): 4.85 (s, 4H, NH); 6.68 (d, 1H, J = 8.4Hz, H-6); 6.81 (d, 1H, J = 8.4 Hz, H-7); 6.90 (s, 1H, H-4). ¹³CNMR (600MHz) CD₃OD: 140.06, 135.67, 129.27, 125.70, 117.32, 115.79, 113.50. Anal. Calcd. for C₇H₇F₃N₂; M.wt: 176.14; C,47.73; H,4.01; F,23.36; N, 15.90; (%); Found: C, 47.13; H, 3.99; F, 23.04; N,15.10 (%).

3-Benzotriazole-5-(yl)-difluoro-methyl-5-tri-fluoromethyl benzotriazole (3)

To a solution of (2) (3 g, 0.017 mmol) in (1ml) glacial acetic acid and H₂O (10 ml) added 2.3g NaNO₂ in H₂O

(5 ml) at 0°C in ice bath. After stirring for 10 m, the color was changed to brown. A precipitate formed in an icewater collected by vacuum filtration and it washed three times with ice-water. It was purified by flash column chromatography (silica gel 100 g, chloroform/ethyl acetate 90:10) to give 3.3 g (54.62%)(Wang & Zhou, 2011).

Yield (54.62%), m.p. 127°C dark brown; IR (KBr) ν cm⁻¹: 1630 (C=N); ¹HNMR (600MHz); (CD₃OD): 1.90 (s, 1H, NH); 1.87 (s, 1H, NH); 7.61 (d, 1H, J = 8.5 Hz, H-6); 7.76 (d, 1H, J = 8.5 Hz, H-6'); 7.90 (d, 1H, J = 8.5 Hz, H-7); 8.03 (d, 1H, J = 8.5, H-7); 8.17 (s, 1H, H-7'); 8.31 (s, 1H, H-4). ¹³CNMR (600MHz) CD₃OD: 141.14 (CF₂), 140.22, 128.89, 128.68, 128.46, 128.25, 126.62, 124.82, 123.75, 123.53, 123.02, 116.21, (CF₃), 115.32. Anal. Calcd. for C₁₄H₇F₅N₆; M.wt: 354.24; C, 47.47; H, 1.99; F, 26.82; N, 23.72; (%); Found: C, 46.98; H, 1.54; F, 26.25; N, 23.10 (%). M+1 = 355.2 (3)

Synthesis of α and β -1-(2,3,5-Tri-O-benzoyl- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole (7) and (8).

Ribosylation 3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole (3). Synthesis of α and β -1-(2,3,5-tri-O-benzoyl- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole (7) and (8).

General Procedure.

A mixture of 3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole (3) (0.02 mol) and hexamethyl disilazane (20 ml) was heated under reflux for 24h with a catalytic amount of ammonium sulfate (0.01g). After that, the clear solution was cooled and evaporated till dryness to give the silylated derivative (5), which directly was dissolved in 20 ml of dry 1,2-dichloroethane and then 1-O-acetyl-2,3,5-tri-O-benzoyl- β -D-ribofuranose (6) (5.05 g, 0.01 mol) was added. The mixture was added dropwise onto a mixture of (10 ml trimethylsilyl trifluoromethane sulfonate (TMSOTf) in dry 1,2-dichloroethane (50 ml)). All mixture was stirred at room temperature for 24 h, and then washed with a saturated solution of aqueous sodium bicarbonate (3 × 50 ml), washed with water (3 × 50 ml), and dried over anhydrous sodium sulfate. The solvent was removed in vacuum gave an anomeric mixture of α and β -1-(2,3,5-tri-O-benzoyl- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole (7) and (8). This mixture has been separated into the two components by on silica gel with chloroform: acetone (9:1) as eluent to afford a white crystal pure α -anomer (7) and β -anomer (8) respectively.

β -1-(2,3,5-Tri-O-benzoyl- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole (7)

Yield (38.85%), w. 3.02g, m.p. 115°C white; ν (cm⁻¹) (KBr) 1740(C=O), 1670 (C=N); ¹HNMR (850MHz); (CDCl₃): δ 1.57 (s, 1H, NH), 3.49 (m, 1H, H-5'); 3.46 (m, 1H, H-5''); 4.58-4.55 (dd, 1H, H-4'); 4.81-8.79 (dd, 1H, H-4''); 4.89-4.47 (m, 1H, H-3'); 5.00-4.96 (m, 1H, H-3''); 6.31-6.368 (t, 1H, H-2', J = 5.8 Hz); 6.61-6.60 (t, 1H, H-2'', J = 5.8 Hz); 6.67 (d, 1H, H-1', J_{1,2'} = 7.5 Hz); 6.73 (d, 1H, H-1'', J_{1,2''} = 7.5 Hz); 8.30-7.38 (m, 37H, Ar-H).

¹³CNMR (850MHz) (CDCl₃) δ 166.03, 165.97, 165.22, 165.19, 165.17, 164.41 (6C=O), 145.47

(CF₂), 133.96, 133.77, 133.36, 133.29, 132.16, 129.92, 129.86, 129.68, 129.62, 129.07, 129.02,

128.65, 128.58, 128.48, 128.43, 127.34, 127.19, 124.92, 124.50, 124.38, 123.23, 123.10, 121.36,

118.51, 118.49 (CF₃), 110.95, 108.12, 108.10, (Ar.C's), 89.23, 89.02, 81.32, 81.24, 74.70, 74.40, 71.62, 71.40, 63.39, 63.07 (sugar carbons). Anal. Calcd. for C₆₆H₄₉F₅N₆O₁₄; M.wt: 1245.15; C, 63.67; H, 3.97; F, 7.63; N, 6.75; (%); Found: C, 63.24; H, 3.01; F, 7.52; N, 5.31 (%).

α -1-(2,3,5-Tri-O-benzoyl- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole (8)

Yield (20.45%), w. 1.59g, m.p. 98°C yellow; ν (cm⁻¹) (KBr) 1740(C=O), 1670 (C=N); ¹HNMR (850MHz); (CDCl₃): δ 2.04 (s, 1H, NH); 4.59-4.55 (m, 1H, H-5'); 4.80 (m, 1H, H-5''); 4.81 (t, 1H, H-4'); 4.89 (t, 1H, H-4''); 4.87 (d, 1H, H-3', J = 4.5 Hz); 5.00-4.97 (d, 1H, H-3'', J = 4.5 Hz); 6.31 (d, 1H, H-2', J = 5.95 Hz); 6.62 (d, 1H, H-2'', J = 5.95 Hz); 6.74 (d, 1H, H-1', J_{1,2'} = 3.4 Hz); 6.67 (d, 1H, H-1'', J_{1,2''} = 3.4 Hz); 8.29-7.26 (m, 37H, Ar-H). ¹³CNMR (850MHz) (CDCl₃): 166.02, 165.96, 165.22, 165.18, 161.03, 157.41 (6 CO), 145.46 (CF₂), 133.95, 133.75, 133.35, 133.28, 132.17, 130.71, 130.55, 130.40, 130.25, 129.92, 129.86, 129.62, 129.08, 129.02, 128.65, 128.57, 128.51, 128.48, 128.43, 124.91, 124.51, 124.38, 123.23, 123.11, 121.38, 121.35, 118.49, 118.48 (CF₃), 110.98, 108.12, 108.10 (Ar.C's), 89.24, 89.02, 81.31, 81.23, 74.71, 74.59, 71.63, 71.40, 63.40, 63.07 (sugar carbons). Anal. Calcd. for C₆₆H₄₉F₅N₆O₁₄; M.wt: 1245.15; C, 63.67; H, 3.97; F, 7.63; N, 6.75; (%); Found: C, 63.08; H, 3.29; F, 7.12; N, 5.10 (%).

Synthesis of nucleosides free

Deprotection of β - and α -1-(2,3,5-trihydroxy- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoro-methyl-5-trifluoromethyl benzotriazole (**9**) and (**10**) respectively.

General Procedure

A mixture of each protected nucleoside β (**7**), α (**8**) (0.001 mol for each), dry absolute methanol (20 ml) and sodium metal (0.055g, 0.001mol) was stirred at room temperature for 48 h. The solvent was evaporated under vacuum to give a colorless solid, which was dissolved in hot water and neutralized with acetic acid. The precipitate compound was chromatographic on silica gel with chloroform: ethyl acetate (9: 1) as eluent to afford colorless and white crystals of the corresponding nucleosides β (**9**) and α (**10**) respectively.

β -1-(2,3,5-Trihydroxy- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoro-methyl-5-trifluoromethyl benzotriazole (**9**)

Yield (87%), m.p. < 300°C white; (H₂O/EtOH, 1:1); ν (cm⁻¹) (KBr) 3400 (OH); ¹HNMR (850 MHz); (CD₃OH): δ 1.88 (s, 1H, NH); 3.67 (d, 2H, H-5'); 3.68 (d, 68, 2H, H-5''); 3.70 (d, 1H, H-3'); 3.78 (d, 1H, H-3''); 3.79 (t, 1H, H-4'); 3.80 (d, 1H, H-4''); 4.22(m, 1H, H-2'); 4.19 (m, 1H, H-2''); 4.47 (t, 1H, OH-5'); 4.91 (t, 1H, OH-3'); 4.93 (d, 1H, OH-2', J = 5.5 Hz); 6.41 (d, 1H, H-1', J_{1,2'} = 7.5 Hz); 6.46(d, 1H, H-1'', J_{1,2''} = 7.5 Hz); 8.30 (s, 1H). ¹³CNMR (850MHz) (CD₃OH): δ 180.41, 175.58, 148.51(CF₂), 146.42, 139.07, 135.78, 133.47, 131.21, 130.23, 128.68, 128.24, 128.09, 125.43, 122.43, 121.58 (Ar C's), 118.51(CF₃), 113.85, 93.22, 92.94, 87.66, 87.54, 75.39, 75.09, 72.28, 72.22, 63.17, 63.06 (sugar carbons), 24.19. Anal. Calcd. for C₂₄H₂₅F₅N₆O₈; M.wt: 620.848; C,46.46; H,4.06; F,15.31; N, 13.54; (%); Found: C, 46.21; H, 3.89; F,15.12; N,12.95 (%).

α -1-(2,3,5-Trihydroxy- β -D-ribofuranosyl)-3-Benzotriazole-5-(yl)-difluoro-methyl-5-trifluoromethyl benzotriazole (**10**)

Yield (65%), m.p. < 300°C white; (H₂O/EtOH, 1:1); ν (cm⁻¹) (KBr) 3400 (OH); ¹HNMR (850 MHz); (CD₃OH): δ 1.88 (s, 1H, NH); 3.30-3.29 (m, 1H-5'); 3.68 (d,1H, J= 5.1 Hz, H-5''); 3.69 (t, 1H, H-4') ; 3.71 (d, 1H, J= 4.25 Hz, H-4''); 3.78 (d, 1H, J= 3.4 Hz, H-3'); 3.79 (t, 1H, H-3''); 3.80 (d, J= 3.4 Hz, 1H, H-2'); 4.23 (m, 1H, H-2'');4.49 (t, 1H, OH-5'); 4.93 (t, 1H, OH-3'); 4.94 (d, 1H, J= 5.1 Hz, OH-2'); 6.43 (d, 1H, J= 4.25 Hz, H-1'); 6.47 (d, 1H, J= 4.25 Hz, H-1''); 8.55-7.31 (m, 6H, Ar-H).

¹³CNMR (850MHz) (CD₃OH): δ 175.62 (CF₂), 148.45, 146.36, 139.05, 135.75, 133.44, 131.21, 130.230, 128.67, 128.24, 125.41, 122.29, 121.57, 121.55, 118.48 (CF₃), 113.86, 111.25 (Ar. C's), 93.22, 92.96, 87.67, 87.54, 75.40, 72.26, 72.19, 63.17, 63.06 (sugar carbons), 24.23. Anal. Calcd. for C₂₄H₂₅F₅N₆O₈; M.wt: 620.848; C,46.46; H,4.06; F,15.31; N, 13.54; (%); Found: C, 46.21; H, 3.89; F,15.12; N,12.95 (%).

4. Results and Discussion

Reduction of nitro group of compound, 4-Amino-3-nitrobenzotrifluoride (**1**) was used Raney Nickle and hydrazine in ethanol was prepared, 1,2-Diamino-4-benzotrifluoride (**2**) as reported in the literature (Wang & Zhou, 2011). Compound (**2**) was stirred with water glacial acetic acid in ice bath then solution NaNO₂ was dropped to 10 min gave of the dimer 3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoro-methyl benzotriazole in yield of 54.62% compound (**3**) and some starting material, although the product dimerized as unexpected (Scheme 1). The structures of the latter products (**2**), and (**3**) were established and confirmed on the bases of their elemental analyses and spectral data. Thus, their ¹H NMR spectra of compound (**2**) showed two a doublet signals at δ 6.68, 6.81 assigned to the aromatic protons of H-6 and H-7 and a singlet signal at 6.90 of H-4 and a singlet signal at δ 4.85 of NH₂. ¹H NMR spectra of compound (**3**) showed four doublet signals at δ 7.61, 7.76, 7.90 and 8.03 with spin-spin coupling constant (J_{H-6}), (J_{H-6'}), (J_{H-7}) and (J_{H-7'}) equal to 8.5 Hz, assigned to the aromatic protons of H-6, H-6', H-7 and H-7', and two singlet signals at 8.17 and 8.31 of H-4 and H-4'.

The ¹³C NMR of compound (**2**) revealed the seven signals at δ 140.06, 135.67, 129.27, 125.70, 117.32, 115.79, 113.50. The fourteen signals at δ 141.14 (CF₂), 140.22, 128.89, 128.68, 128.46, 128.25, 126.62, 124.82, 123.75, 123.53, 123.02, 116.21 of compound (**3**). HMBC (see Fig(1)). 2D NMR Experiments to characterize dimers of benzotriazole and for studies on the conformation of 3-Benzotriazole-5-(yl)-difluoromethyl-5-trifluoromethyl benzotriazole. The spectrum of sucrose at 850 MHz is shown below. The peak outlined in green shows the two bond correlation between the C-5 carbon and the H-6 proton. The peak outlined in red correlates the CF₂ carbon and H-7 proton separated by 4 bonds. The peak outlined in blue correlates the C-5 carbon and H-4 proton separated by 3 bonds. Note also that the C-7 carbon correlates with the H-7 proton across bond. (Diana et al, 2007; Katritzky et al, 1998; Katritzky et al, 2003 and Shi et al, 2011).

IR spectra of compound (**2**) showed absorptions around 3400 cm⁻¹ for (NH₂). While the (NH₂) group

disappeared of compound (3), showed absorptions 1630 cm^{-1} for (C=N). Mass spectra of compound (3) $M^+ = 355.2$.

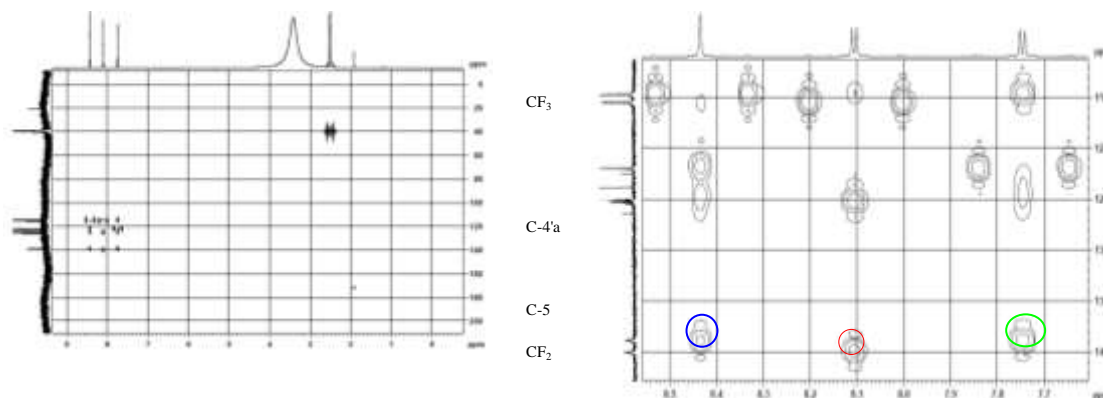
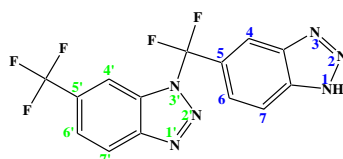


Fig (1). HMBC (Heteronuclear Multiple Bond Correlation) of compound (3)



3-Benzotriazole-5-(yl)-difluoro-methyl-5-trifluoromethyl benzotriazole (3)

Ribosylation of (3) was achieved by refluxing in hexamethyldisilazane (HMDS) to give the silylated derivatives (5). The latter was stirred with 1-*O*-acetyl-2,3,5-*O*-benzoyl- β -D-ribofuranose (6) in the presence of dry 1,2-dichloroethane as solvent using trimethylsilyl trifluoromethane-sulfonate (TMSOTf) as a catalyst for 24 h (followed by TLC), following the silylation method according to to give the corresponding β -anomeric (7) and α -anomeric (8) protected *N*-nucleoside derivatives, respectively, in good yields (Vorbruggen et al, 1981). The mixture was separated by used column chromatography.

Debenzoylation of (7) and (8) were performed by using methanolic sodium methoxide solution following Zemplen et al.'s method (Zemplen et al, 1939) to afford the free nucleosides (9) and (10), respectively (Scheme 2).

The chemical structures of nucleoside derivatives (2)-(10) were established and confirmed on the basis of their elemental analyses and spectral data (IR, ^1H and ^{13}C NMR) (see the Experimental section). ^1H NMR spectra of (7) and (8) showed in each case a doublet signals at δ 6.67, 6.73 for compound (7) and at δ 6.74, 6.67 for compound (8) assigned to the anomeric proton of the ribose moiety with spin-spin coupling constant ($J_{1,2}$) and ($J_{1,2''}$) equal to 7.5 Hz, which confirms the β -anomeric configuration. While confirms the α -anomeric configuration showed spin-spin coupling constant ($J_{1,2}$) and ($J_{1,2''}$) equal to 3.4 Hz, which confirms the α -anomeric configuration for compound (8) (Mosselhi, 1993; Mosselhi, 1999; Mosselhi & Seliger, 2001; Mosselhi & Break, 2011; Break et al, 2010; Metwally et al, 2010; Mosselhi & Neidlein, 2009; Break et al, 2013 and Khalil, 2006). The ^1H NMR spectra of nucleosides free showed a doublet signals at δ 6.41, 6.46 for compound (9) spin-spin coupling constant ($J_{1,2}$) and ($J_{1,2''}$) equal to 7.5 Hz β -anomeric configuration and at δ 6.43, 6.47 for compound (10) assigned to spin-spin coupling constant ($J_{1,2}$) and ($J_{1,2''}$) equal to 4.25 Hz, which confirms the α -anomeric configuration.

The ^{13}C NMR of nucleoside products revealed the signals at δ 166.03, 165.97, 165.22, 165.19, 165.17 and 164.41 are due to the six benzoyl carbonyl groups for compound (7), and 166.02, 165.96, 165.22, 165.18, 161.03 and 157.41 are six carbonyl groups for compound (8). The ten signals at δ 89.23, 89.02, 81.32, 81.24, 74.70, 74.40, 71.62, 71.40, 63.39 and 63.07 for compound (7), and The ten signals at δ 89.24, 89.02, 81.31, 81.23, 74.71, 74.59, 71.63, 71.40, 63.40 and 63.07 for compound (8) were assigned to C-1', C-1'', C-2', C-2'', C-3', C-3'', C-4', C-4'', C-5' and C-5'' of the sugar moiety, respectively. The CF_2 group showed at 145.47 and 145.46 of compound (7) and (8) respectively. The ^1H NMR of (7) and (8) showed the expected base moiety protons in addition to the sugar moiety protons (see the Experimental section). The IR spectrum of compounds (7) and (8) showed the stretching vibration frequencies of the benzoyl carbonyl C=O groups at 1740 cm^{-1} . IR spectra of compounds (9) and (10) showed absorptions around 3400 cm^{-1} for (OH).

5. Conclusion

Benzotriazoles are scientific importance in many biologically active compounds. So synthesis and characterization of 3-Benzotriazole-5-yl difluoro-methyl-5-trifluoromethyl benzotriazole (**3**). Ribosylation of compound (**3**) with 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- β -D-ribofuranose(**6**) afforded mixture β - and α - anomeric of the benzoylated nucleoside derivatives (**7**) and (**8**), respectively.

Deprotection of the latter by using dry absolute methanol and sodium metal gave new free N-nucleosides (**9**) and (**10**), respectively, in moderate yields. Nucleosides obtained have been identified by their spectral analysis.

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