



Electrospinning Fabrication of Polycrystalline LaCrO_3 Porous Hollow Nanofibers

Jinxian Wang, Xiangting Dong (Corresponding author) & Qizheng Cui

School of Chemistry and Environmental Engineering

Changchun University of Science and Technology

Jilin 130022, China

Tel: 86-431-8558-2574 E-mail: dongxiangting888@yahoo.com.cn

Guixia Liu & Wensheng Yu

School of Chemistry and Environmental Engineering

Changchun University of Science and Technology

Jilin 130022, China

This work was financially supported by the Science and Technology Development Planning Project of Jilin Province (Grant Nos. 20040125, 20060504, 20070402), the Scientific Research Planning Project of the Education Department of Jilin Province (Under grant Nos. 200224, 2005109, 2007-45)

Abstract

Polyvinyl Pyrrolidone(PVP)/[$\text{La}(\text{NO}_3)_3 + \text{Cr}(\text{NO}_3)_3$] composite nanofibers were fabricated by electrospinning. SEM micrographs indicated that the surface of the prepared composite fibers was smooth, and the diameter of the nanofibers was in the range of 1-3 μm . XRD analysis revealed that the composite nanofibers were amorphous in structure. LaCrO_3 nanofibers were fabricated by calcination of the PVP/[$\text{La}(\text{NO}_3)_3 + \text{Cr}(\text{NO}_3)_3$] composite fibers. The diameters of LaCrO_3 nanofibers were smaller than those of the relevant composite fibers. The surface of the LaCrO_3 nanofibers became coarse with the increase of calcination temperatures. LaCrO_3 hollow-centered and porous nanofibers formed by nanoparticles were acquired when firing temperature was 600-900°C. SEM images indicated that the diameters of the synthesized LaCrO_3 nanofibers ranged from 500 to 800nm, and their lengths were greater than 100 μm . XRD analysis revealed that LaCrO_3 nanofibers were orthorhombic in structure with space group Pbnm. Possible formation mechanism for LaCrO_3 nanofibers was preliminarily proposed.

Keywords: Electrospinning, Nanofibers, La, Cr, LaCrO_3

1. Introduction

The science and technology of nanostructured materials is advancing at a rapid pace (Mohapatra, 2008 & Zhang, 2007). Over the past decade, the preparation and functionalization of one-dimensional nanostructured materials has become one of the most highly energized research fields (Hu, 2008 & Kar, 2006). One-dimensional nanostructured materials, such as nanowires, nanorods, nanowhiskers and nanofibers, have stimulated great interests due to their importance in basic scientific research and potential technological applications (Huynh, 2002 & Duan, 2003). They are expected to play an important role as both interconnects and functional components in the fabrication of nanoscale

electronic and optoelectronic devices. In order to obtain these materials, various preparation methods have been developed including arc discharge, laser ablation, template, precursor thermal decomposition, and other methods (Iijima, 1991, Morales, 1998, Shi, 2001 & Pan, 2001). Electrospinning technique is widely applied to prepare polymers nanofibers (Li, 2004, 1151-1170). Recently, some inorganic compounds nanofibers have been prepared by electrospinning technique using electrospun fibers of polymer/inorganic composite as the precursor (Li, 2004, Zhang, 2008 & Shao, 2004). This processing involved the following three steps: (1) Preparation of a gel with suitable inorganic precursor and proper polymer, and achieving the right rheology for electrospinning process; (2) Electrospinning of the gel to obtain fibers of polymer/inorganic precursors composite; (3) Calcinations of the composite fibers to obtain final oxide fibers. It is important; however, to control all of the above three steps in order to obtain high quality fibers with the desired final properties. LaCrO_3 has attracted much interest recently due to their specific electrical and catalytic properties (Dong, 1994 & Yang, 2003). A few methods on the preparation of LaCrO_3 nanocrystalline materials were reported (Zhang, 2005 & Johnson, 2004). However, to the best of our knowledge, there have been no reports on the preparation of LaCrO_3 porous hollow nanofibers by electrospinning. In this paper, LaCrO_3 nanofibers were fabricated by calcination of the electrospun fibers of PVP/(lanthanum nitrate and chromium nitrate) composite, and some new results were obtained.

2. Experimental section

2.1 Chemicals

Polyvinyl pyrrolidone (PVP) ($M_r \approx 10000$) and chromium nitrate enneahydrate [$\text{Cr}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$] were purchased from Tianjin Kermel Chemical Reagents Development Center. Lanthanum nitrate hexahydrate [$\text{La}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$] was obtained from Tianjin Guangfu Institute of Fine Chemicals. All chemicals were analytically pure and directly used as received without further purification. Distilled water was used as solvent.

2.2 Preparation of PVP/[$\text{La}(\text{NO}_3)_3$ and $\text{Cr}(\text{NO}_3)_3$] composite gel

PVP/[$\text{La}(\text{NO}_3)_3$ and $\text{Cr}(\text{NO}_3)_3$] composite solution was prepared by dissolving 14.2568g of PVP powders, 2.3830g of $\text{La}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ and 2.2021g of $\text{Cr}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ in 12.15g of distilled water, and stirring for 10h, then remaining motionlessly for 2h. Thus, a viscous gel of PVP/[$\text{La}(\text{NO}_3)_3 + \text{Cr}(\text{NO}_3)_3$] composite containing 46%(wt%) PVP, 10%(wt%) metallic nitrate, 44%(wt%) H_2O , and the molar ratio 1:1 of La^{3+} to Cr^{3+} were obtained for electrospinning processing.

2.3 Fabrication of PVP/[$\text{La}(\text{NO}_3)_3$ and $\text{Cr}(\text{NO}_3)_3$] composite fibers and LaCrO_3 nanofibers

The setup used for electrospinning was indicated in Figure 1. The above composite gel of PVP, $\text{La}(\text{NO}_3)_3$, $\text{Cr}(\text{NO}_3)_3$ and H_2O mixture was contained in a plastic syringe with a stainless steel needle on its top. A copper wire connected to a DC high-voltage generator was placed in the gel, and the gel was kept in the syringe by adjusting the angle between syringe and the fixing bar. A grounded aluminum foil served as counter electrode and collector plate. A voltage of 18 kV was applied to the composite gel and a sprayed dense web of fibers was collected on the aluminum foil. The collected fibers were PVP/[$\text{La}(\text{NO}_3)_3 + \text{Cr}(\text{NO}_3)_3$] composite fibers. The prepared composite fibers were dried initially at 70°C for 12h under vacuum, and then calcined at a heating rate of 120°C/h and remained for 10h at 300°C, 600°C and 900°C, respectively. Thus, LaCrO_3 nanofibers were obtained when calcination temperature is 600-900°C.

2.4 Characterization methods

XRD analysis was performed with a Holland Philips Analytical PW1710 BASED X-ray diffractometer using $\text{Cu K}\alpha_1$ radiation, the working current and voltage were 30mA and 40kV, respectively. Scans were made from 10° to 80° at the scanning speed of 3°/min, and step was 0.05°. The morphology and size of the fibers were observed with a S-4200 scanning electron microscope made by Japanese Hitachi company. FTIR spectra of the samples were recorded on BRUKER Vertex 70 Fourier transform infrared spectrophotometer made by Germany Bruker company, and the specimen for the measurement was prepared by mixing the sample with KBr powders and then the mixture was pressed into pellets, the spectrum was acquired in a wave number range from 4000 cm^{-1} to 400 cm^{-1} with a resolution of 4 cm^{-1} .

3. Results and discussion

3.1 XRD patterns

In order to investigate the lowest crystallizing temperature and the variety of phases, the PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers and samples obtained by calcining the composite fibers at different temperatures for 10h were characterized by XRD, as indicated in Figure 2. The results showed that the PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers were amorphous in structure, only a broad peak was located around 22°, it was the typical peak of the amorphous polymer, indicating that the composite fibers were amorphous in structure(Figure 2a). The sample was also amorphous at 300°C, and no obvious diffraction peaks could be observed(Figure 2b). The polycrystalline LaCrO₃ nanofibers with single phase were synthesized when calcination temperature was in the range of 600-900°C(Figure 2c and 2d), the d(spacing between crystallographic plane)values and relative intensities of LaCrO₃ are consistent with those of JCPDS standard card(24-1016), and the crystal structure of the prepared LaCrO₃ was orthorhombic system with space group Pbnm.

3.2 SEM images

In order to study the morphology and size of the as-synthesized fibers, the prepared fibers were investigated by SEM, as shown in Figure 3. As seen from Figure 3, the morphology and size of the fibers varied strongly with the increase of calcination temperatures. The surface of the PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers was very smooth, and the diameter of the composite fibers was in the range of 1-3μm. The morphology and size of the fibers at 300°C were almost the same as those of the composite fibers. The surface morphology of LaCrO₃ nanofibers became coarse with the increase of calcination temperatures. LaCrO₃ porous hollow nanofibers formed by nanoparticles were acquired at 600°C-900°C. SEM analysis indicated that the diameters of the synthesized LaCrO₃ nanofibers were in the range of 500-800nm, and their lengths were greater than 100μm. The diameters of LaCrO₃ nanofibers were smaller than those of the PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers owing to the decomposition and evaporation of PVP and NO₃⁻.

3.3 FTIR spectra analysis

Pure PVP, PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers and LaCrO₃ nanofibers(obtained by calcination of the PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers at 900°C for 10h) were analyzed by FTIR, as shown in Figure 4. As seen from Figure 4, PVP(Figure 4a) and PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers(Figure 4b) had the identical spectra, but absorption peaks intensity of spectrum for PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers was lower than those of spectrum for pure PVP. This resulted from the lower content of PVP in the PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers. All absorption peaks were attributed to PVP at 3438cm⁻¹, 2956cm⁻¹, 1662cm⁻¹, 1424cm⁻¹, and 1290cm⁻¹, corresponding to the stretching vibrations of hydroxyl group(ν_{O-H}), C-H bond(ν_{C-H}), carbonyl group($\nu_{C=O}$), C-H bond(ν_{C-H}), and C-N bond or C-O bond(ν_{C-N} or ν_{C-O}), respectively. It was seen from Figure 4c that all peaks of PVP disappeared, and at low wave number range, new absorption peaks at 598 and 415cm⁻¹ were appeared. The new absorption peaks were ascribed to the vibration of metal-oxygen bond, indicating that LaCrO₃ was formed. The results of FTIR analysis were in good agreement with XRD results.

3.4 Possible formation mechanism of LaCrO₃ porous hollow nanofibers

Possible formation mechanism of LaCrO₃ porous and hollow nanofibers was described as follows(as indicated in Figure 5). La(NO₃)₃·6H₂O, Cr(NO₃)₃·9H₂O and PVP were mixed with distilled water to form sol with certain viscosity. PVP acted as template during the formation processing of LaCrO₃ nanofibers. La³⁺, Cr³⁺ and NO₃⁻ were mixed with or absorbed onto PVP molecules to fabricate PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers under electrospinning. During calcination treatment of the composite fibers, solvent water containing La³⁺, Cr³⁺, and NO₃⁻ ions in the composite fibers would remove to the surface of the PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers and eventually evaporated from the composite fibers. Thus, La³⁺, Cr³⁺, and NO₃⁻ ions were also removed to the surface of the composite fibers brought by removed water. With the increasing in calcination temperature, PVP, and NO₃⁻ would oxidize and volatilize rapidly, La³⁺ and Cr³⁺ were oxidized into LaCrO₃ crystallites, and many crystallites were combined to form small LaCrO₃ nanoparticles, and these nanoparticles were mutually connected to generate hollow-centered and porous LaCrO₃ nanofibers. It was found from experiments that the average molecular weight of PVP and PVP content in the starting mixed sol had important impact on the formation of LaCrO₃ porous hollow

nanofibers. Further work is in progress.

4. Conclusions

4.1 PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers were fabricated by electrospinning. Polycrystalline LaCrO₃ nanofibers were synthesized by calcining the relevant composite fibers at 600-900°C.

4.2 XRD analysis revealed that the composite fibers were amorphous in structure. The crystal structure of LaCrO₃ nanofibers was orthorhombic system with space group Pbnm.

4.3 SEM micrographs indicated that the surface of the prepared composite fibres was smooth, and the diameters of the composite fibres were in the range of 1-3µm. The diameters of LaCrO₃ nanofibers were smaller than those of the composite fibers. The surface of the LaCrO₃ nanofibers became coarse with the increase of calcination temperatures. LaCrO₃ porous and hollow nanofibers formed by nanoparticles were acquired when calcining temperature was 600-900°C. The diameters of LaCrO₃ nanofibers were in the range of 500-800nm, and their lengths were greater than 100µm.

References

- Dong, X. T., Guo, Y. Z., Yu, D. C., et al. (1994). Synthesis and electrical properties of LaCrO₃ nanometer powder. *Chin. J. Mater. Res.*, 8(3), 263-266.
- Duan, X. F., Huang, Y., Agarwal, R., Lieber, C. M. (2003). Single-nanowire electrically driven lasers. *Nature*, 421, 241-245.
- Hu, X. K., Qian, Y. T., Song, Z. T., et al. (2008). Comparative study on MoO₃ and H_xMoO₃ nanobelts: structure and electric transport. *J. Chem Mater*, 20(4), 1527-1533.
- Huynh, W. U., Dittmer, J. J., Alivisatos, A. P. (2002). Hybrid nanorod-polymer solar cells. *Science*, 295, 2425-2427.
- Iijima, S. (1991). Helical microtubules of graphitic carbon. *Nature*, 354, 56-58.
- Johnson C., Gemmen R., Orlovskaya N. (2004). Nano-structured self-assembled LaCrO₃ thin film deposited by RF-magnetron sputtering on a stainless steel interconnect material. *Composites Part B: Engineering*, 35(2), 167-172.
- Kar, S., Chaudhuri, S. (2006). Shape selective growth of CdS one-dimensional nano-structures by a thermal evaporation process. *J. Phys. Chem. B*, 110(10), 4542-4547.
- Li, D., Xia, Y. N. (2004). Direct fabrication of composite and ceramic hollow nanofibers by electrospinning. *Nano Lett.*, 4(5), 933-938.
- Li, D., Xia, Y. N. (2004). Electrospinning of Nanofibers: Reinventing the Wheel. *Adv. Mater.*, 16(14), 1151-1170.
- Mohapatra, S. K., Misra, M., Mahajan, V. K., et al. (2008). Synthesis of Y-branched TiO₂ nanotubes. *Materials Letters*, 62, 1772-1774.
- Morales, A. M., Lieber, C. M. (1998). A laser ablation method for the synthesis of crystalline semiconductor nanowires. *Science*, 279, 208-211.
- Pan, Z. W., Dai, Z. R., Wang, E. L. (2001). Nanobelts of semiconducting oxides. *Science*, 291, 1947-1949.
- Shao, C. L., Guan, H. Y., Liu, Y. C., et al. (2004). A novel method for making ZrO₂ nanofibres via an electrospinning technique. *J. Crystal Growth*, 267, 380-384.
- Shi, W. S., Zheng, Y. F., Wang, N., et al. (2001). A general synthetic route to III-V compound semiconductor nanowires. *Adv. Mater.*, 13, 591-594.
- Yang, Q. H., Fu, X. X. (2003). Analysis of photocatalytic oxidation activity of nano-LaMO₃ (M=Cr, Mn, Fe, Co) compounds. *J. Chin. Ceram. Soc.*, 31(3), 254-256.
- Zhang B. Q., Li S. L., Sun L. C., et al. (2005). Preparation of doped lanthanum chromites nano-powders by aqueous organic-gel method. *J. Chin. Ceram. Soc.*, 33(4), 447-451.
- Zhang, S. H., Dong, X. T., Xu, S. Z., et al. (2007). Preparation and characterization of TiO₂@SiO₂ submicron-scaled

coaxial cables via a static electricity spinning technique. *Acta Chimica Sinica*, 65(23), 2675-2679.

Zhang, S. H., Dong, X. T., Xu, S. Z., et al. (2008). Preparation and characterization of $\text{TiO}_2/\text{SiO}_2$ composite hollow nanofibers via an electrospinning technique. *Acta Materiae Compositae Sinica*, 25(3), 138-143.

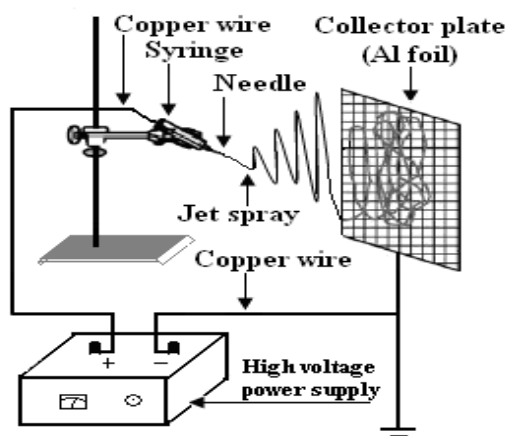


Figure 1. Schematic diagram of electrospinning setup

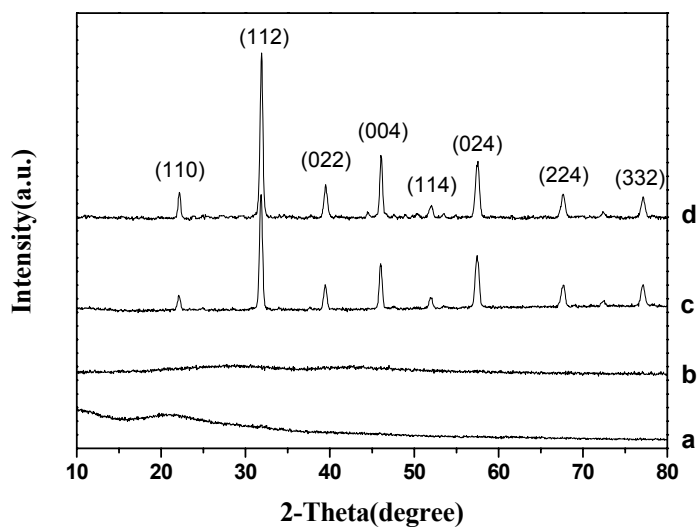


Figure 2. XRD patterns of samples

a. PVP/[$\text{La}(\text{NO}_3)_3 + \text{Cr}(\text{NO}_3)_3$] composite fibers b. 300°C c. 600°C d. 900°C

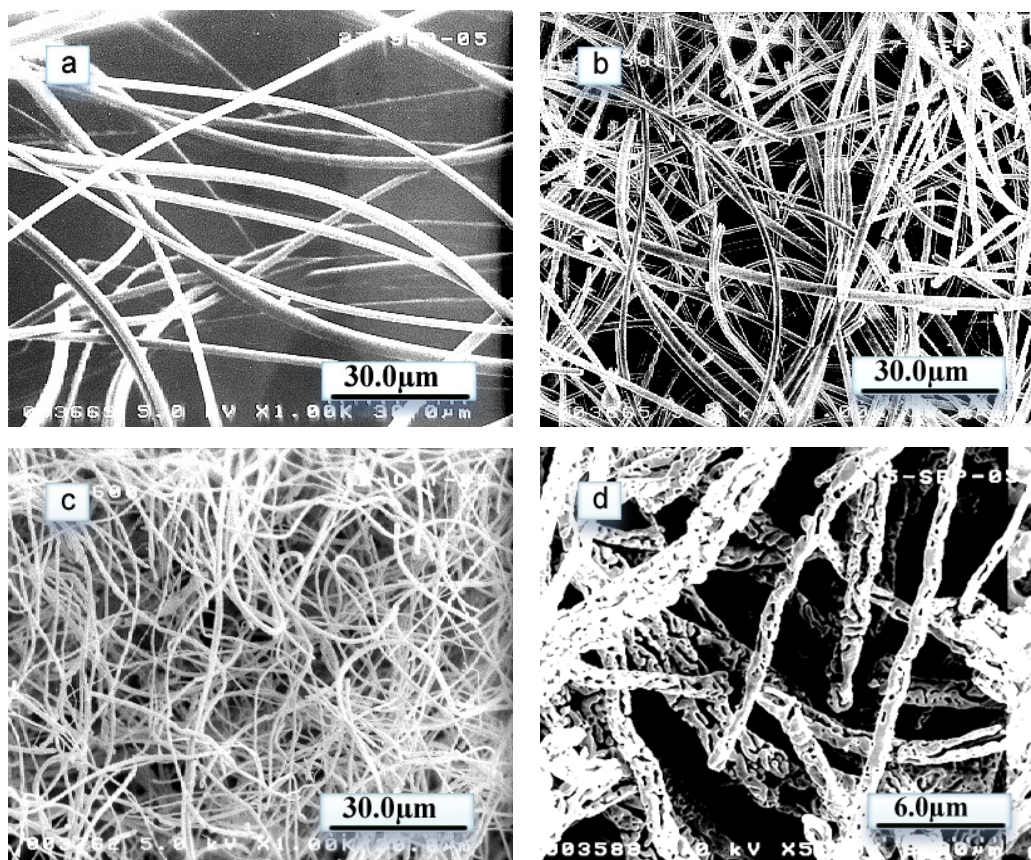


Figure 3. SEM micrographs of the fibers obtained at different temperatures
 a. PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers b. 300°C c. 600°C d. 900°C

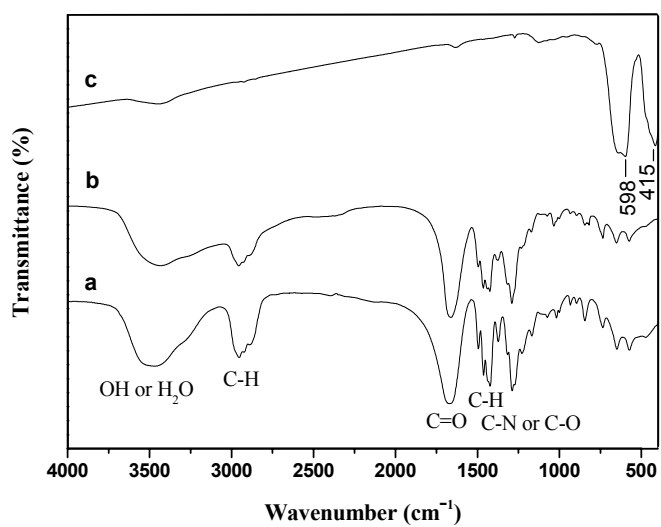


Figure 4. FTIR spectra of the samples
 a. PVP b. PVP/[La(NO₃)₃+Cr(NO₃)₃] composite fibers c. LaCrO₃ nanofibers

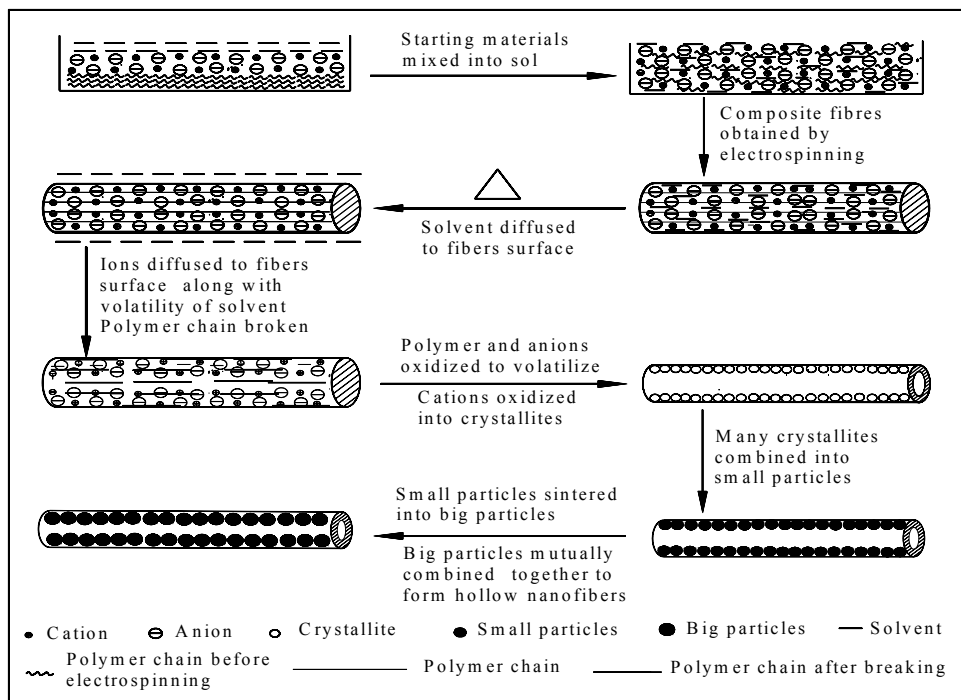


Figure 5. Illustrative diagram of possible formation mechanism of LaCrO_3 porous hollow nanofibers