



## Development of Environmentally Friendly Water-Based Synthetic Metal-Cutting Fluid

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### Abstract

This paper introduces a method for preparation of environmentally friendly water-based synthetic metal-cutting fluid, which is formulated with the self-made triethanolamine ricinoleate as base oil and other additives. This cutting fluid exhibits good cooling, cleaning, anti-rust, anti-corrosive and lubricating properties, is totally free of mineral oil, animal oil, nitrite that is harmful to the human body, phosphate that causes water pollution and etc., and has stable and reliable quality, long service life, easily available raw materials and low production cost. It is quite a perfect “environmentally friendly cutting fluid”.

**Keywords:** Metal-cutting fluid, Triethanolamine ricinoleate, Water-based cutting fluid, Biodegradation, Base oil

### Preface

Metal-cutting fluid, an indispensable additive in metal-cutting process, has functions of lubricating, cooling, cleaning, anti-rust and etc. It has remarkable effects on increasing the durability of cutter and the efficiency of production, improving the product quality, and prolonging the service life of cutter, in turn prolongs the service life of machine and ensures the stability and reliability of working conditions of machine. Therefore, research on cutting fluid technology, as well as improvement of cutting fluid quality play important roles in the modern mechanical processing industry. However, many commercially available cutting fluids contain organic sulfur, chlorine, nitrite and etc that are harmful to the human body and environment, which have severely negative effects on their applications (Feng, Jufen et al, 1995, p. 40-43).

The toxicity of cutting fluid causes harm to the human body. Under the conditions of high speed cutting or high load cutting, the resulted high temperature accelerates the oil-based cutting fluid to release harmful gases and oil mist, the lower the ignition point and viscosity of cutting fluid are, the severer the oil mist is. As for the water-based cutting fluid, the resulted micro-droplets irritate the mucous membrane of the respiratory system, and cause infection (Zeng, Qingliang et al, 2006, p. 113-115).

The environment pollutions resulted from cutting fluids mainly include two aspects as follows: (1) The erratic volatilization, spatter and leakage of cutting fluid during cutting process lead to a drastic environmental deterioration, and severely affect the safety of production; (2) mineral oil-based cutting fluid has a poor biodegradability, and will remain in water and soil for a long time, which causes contamination to lakes, rivers, oceans and underground water. Phosphate, a conventional anti-rust agent, is proved to be responsible for the red tides of lakes and rivers (Li, Chunfeng et al, 2001, p. 15-17).

Ester oil, a substitute for mineral oil, has a biodegradation rate of 90%~100%; however, it is rather expensive. Therefore, the further study on the fully biodegradable cutting fluid of moderate price has become more and more important throughout the world (Asadauskas, S et al, 1996, p. 877-882). Synthetic cutting fluid has advantages of being cheap, fast heat dissipation, being cleanable, excellent visibility in processing, being easy to dilute, high stability and perfect anti-rust ability. In addition, it is free of nutrients that support growth of microorganisms, and prevents the environment from being polluted (Kong Jixia, 2004, p. 49-52). However, the studies on the preparation conditions of base oil for synthetic cutting fluid have been rarely reported. This paper introduces a low-cost method for preparation of environmentally friendly triethanolamine ricinoleate, a base oil that substitutes for mineral oil, investigates the conditions for preparing this base oil, and formulates an environmentally friendly water-based synthetic metal-cutting fluid which is applicable for processing various metals. The above-mentioned base oil has an excellent solubility in



acid to triethanolamine is determined to be 1:3.

### 2.3 Determination of the optimal reaction temperature

Reaction temperature has a certain effect on the performances of the product. Overheat leads to the deterioration of castor acid, whereas overlow temperature results in incomplete reaction. Table 3 exhibits the effect of temperature on the performances of the obtained product.

From table 3, we can see that the reaction is slow and incomplete, and the product is viscous and insoluble when the reaction temperature is lower than 80 °C; whereas the color of the product is darker and the transparency of the product is lower when the reaction temperature is higher than 130 °C. Therefore, the reaction temperature should be in the range of 90~120 °C.

### 2.4 Determination of the optimal reaction period

Reaction period also has a certain effect on the performances of the product. Overshort reaction period leads to incomplete reaction; whereas, overlong reaction results in more byproducts. Table 4 investigates the effects of reaction period on the stability of the ester at reaction temperature of 90~120 °C.

From table 4, we can see that the stability of the synthetic ester increases with the reaction going on at low reaction temperature (90~100 °C); whereas, the stability decreases and large amount of byproducts appear when the reaction temperature is over 110 °C. In view of the production cost, it is determined to carry out the reaction at 90 °C for 150 min.

### 2.5 Selection of catalyst

For further increase of reaction speed and production efficiency, we investigate the effect of different catalysts on the esterification at the optimal molar ratio, reaction period and reaction temperature (see table 5).

From table 5, we can see that the catalysis effect of p-toluene sulphonic acid is better than that of concentrated sulfuric acid. In addition, concentrated sulfuric acid leads to the increase of byproducts with the reaction going on at high temperature. Therefore, p-toluene sulphonic acid is selected as the catalyst in this study, and its amount is determined to be 0.5%.

### 2.6 Selection of polymerization inhibitor

Besides the esterification, intermolecular polymerization and intramolecular polymerization also occur. Polymerization inhibitor is used to prevent molecular polymerization. Table 6 reveals the comparison between two polymerization inhibitors.

From table 6, we can see that these two inhibitors have the same effects. However, the water solution of hydroquinone appears reddish when oxidized in air, and affects the visibility of cutting fluid, meanwhile, it sort of has toxicity. Therefore, we select ferrous ammonium sulfate as the polymerization inhibitor in this study.

### 2.7 Optimal formulation of cutting fluid

Table 7 exhibits the optimal formulation of cutting fluid.

### 2.8 Measurement of the product quality

Measure the performances of the obtained water-based synthetic metal-cutting fluid according to the national standard (GB6144-1985). Table 8 exhibits the measurement results.

### 2.9 Comparison with traditional cutting fluid

Generally, traditional cutting fluids contain organic sulfur, chlorine, nitrite and etc that are harmful to the human body and environment, which have severely negative effects on their applications. In addition, mineral oil or synthetic alkane based cutting fluids have poor biodegradability, which further limits the application of traditional cutting fluid. However, synthetic ester-based cutting fluids have advantages of good biodegradability, being cheap, fast heat dissipation, being cleanable, excellent visibility in processing, being easy to dilute, high stability and perfect anti-rust ability, therefore they have bright future in the market. The only fly in the ointment is that the poor lubricating ability of synthetic ester-based cutting fluids limits their broader applications. In this study, the base oil, triethanolamine ricinoleate, is a nonionic surfactant. As a result, it can lower the friction, and increase the lubricating ability of cutting fluid. In addition, sodium gluconate exhibits an excellent synergistic effect with sodium benzoate, even small amount of sodium gluconate can remarkably increase the anti-rust effect of sodium benzoate.

## 3. Conclusion

A base oil of excellent water solubility, triethanolamine ricinoleate, is formulated at 90 °C for 150 min with castor acid and triethanolamine at the molar ratio of 1 to 3, p-toluene sulphonic acid as the catalyst, and ferrous ammonium sulfate as the polymerization inhibitor. An environmentally friendly water-based synthetic metal-cutting fluid is in turn

formulated with this self-made base oil and other additives. The performances of this cutting fluid meet the requirements specified in the national standard GB 6144-1985. The test results indicate that this cutting fluid has good cooling, cleaning, anti-rust, anti-corrosive and lubricating properties, is totally free of mineral oil, animal oil, nitrite that is harmful to the human body, phosphate that causes water pollution and etc., and has stable and reliable quality, long service life, easily available raw materials and low production cost.

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Table 1. The selection of base oil

Reactant	Triethanolamine + castor acid			Triethanolamine + castor oil			Triethanolamine + soybean oil		
	120	140	160	120	140	160	120	140	160
Reaction temperature (°C)	120	140	160	120	140	160	120	140	160
Solubility*	++	+	+ -	--	--	--	--	--	+ -
Transparency**	100%	50%	20%	—	—	—	—	—	100%

\* Solubility is classified into 5 grades, and expressed as ++, +, +-, - and – in order from fully soluble to insoluble.

\*\* In 2% water solution, transparency is expressed with a percentage from 100% to 1% in order from transparent to milky, and 0 represents the transparency when the synthetic ester is insoluble.

Table 2. The effects of molar ratio on the solubility and stability of the obtained synthetic ester

Molar ratio of castor acid to triethanolamine	1:1	1:1.2	1:1.4	1:1.6	1:1.8	1:2	1:2.5	1:3	1:4
Solubility	--	--	-	+ -	+	+	++	++	++
Stability	—	—	—	30 min	1 h	2 h	4 h	F+	F+

F+ represents that the obtained synthetic ester stays stable for a long time.

Table 3. The effects of reaction temperature on the performances of the obtained synthetic ester

Reaction temperature (°C)	80	90	100	110	120	130	140
Solubility	+	++	++	++	++	+	+
Transparency	100%	100%	100%	100%	100%	80%	50%

Table 4. The effects of reaction period on the stability of the obtained synthetic ester at different reaction temperature

Reaction temperature (°C)	Reaction period (min)							
	30	60	90	120	150	180	210	240
90	2 d	2 d	3 d	4 d	F+	F+	F+	F+
100	2 d	2 d	3 d	4 d	F+	F+	F+	F+
110	2 d	2 d	2 d	2 d	1 d	1 d	1 d	1 d
120	30 min	50 min	60 min					

F+ represents that the obtained synthetic ester stays stable for a long time.

Table 5. The effects of different catalysts on the performances of the obtained synthetic ester

Catalyst	p-toluene sulphonic acid (0.5%)	p-toluene sulphonic acid (1.0%)	Concentrated sulfuric acid (0.5%)	Concentrated sulfuric acid (1.0%)
Solubility	++	++	+-	--
Transparency	100%	100%	30%	0

Table 6. The comparison between two polymerization inhibitors

Polymerization inhibitors	Solubility	Transparency	Stability
Hydroquinone	++	80%	F+
Ferrous ammonium sulfate	++	100%	F+

Table 7. The formulation of cutting fluid

Triethanolamine ricinoleate	20%
OP-10	1%
Extreme pressure agent	15%
Dimethyl silicone oil	1%
Sodium benzoate	0.5%
Sodium gluconate	0.5%
Benzotriazole	1%
EDTA	A little
Deionized water	Margin

Table 8. The measurement of performances of the obtained cutting fluid

Measured performances		Quality index		Test method
Concentrated fluid	Appearance	Transparent solution		GB6144—85
	Storage stability	Qualified		GB6144—85
5% solution of ionized water	Transparency	Transparent		GB6144—85
	pH	8~9		GB6144—85
	Defoaming property	Qualified		GB6144—85
	Surface tension	0.031 dyn/cm		GB6144—85
	Anti-corrosion (55±2 °C)	Grey cast iron	Grade A	GB6144—85
		Red copper	Grade A	GB6144—85
		LY12 Aluminum	Grade A	GB6144—85
	Anti-rust	Single layer	Grade A	GB6144—85
		Laminated layer	Grade A	GB6144—85
	Maximum non-seizure load (PB)	≤150 kg		GB6144—85
Compatibility with the machine paint	Qualified		GB6144—85	

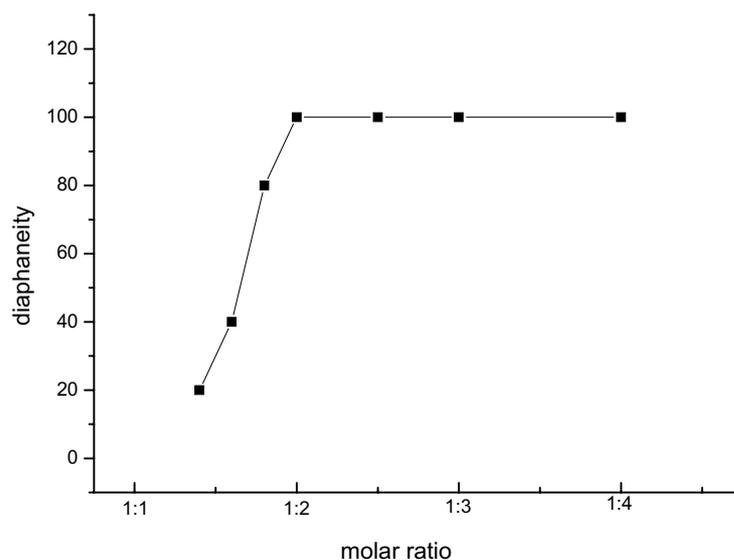


Figure 1. The effect of molar ratio of castor acid to triethanolamine on the transparency of the obtained synthetic ester