Study on How to Effectively Extract Bamboo Fibers from Raw Bamboo and Wastewater Treatment

Nguyen Tien Phong Graduate Student, Doshisha University Kyotonabe-City, Kyoto 610-0394, Japan Tel: 81-080-413-38120 E-mail: etk1306@mail14.doshisha.ac.jp

Toru Fujii

Department of Mechanical Engineering and Systems, Doshisha University Kyotonabe-City, Kyoto 610-0394, Japan Tel: 81-090-819-00412 E-mail: tfujii@mail.doshisha.ac.jp

Bui Chuong

Polymer Center, Hanoi University of Technology No 1, Dai Co Viet, Hanoi, Vietnam Tel: 84-090-344-6055 E-mail: buichuong-pc@mail.hut.edu.vn

Kazuya Okubo

Department of Mechanical Engineering and Systems, Doshisha University Kyotonabe-City, Kyoto 610-0394, Japan Tel: 81-774-656-444 E-mail: kokubo@mail.doshisha.ac.jp

Received: August 3, 2011Accepted: August 23, 2011Published: January 1, 2012doi:10.5539/jmsr.v1n1p144URL: http://dx.doi.org/10.5539/jmsr.v1n1p144

Abstract

Bamboo fibers are focused as one of substitution for natural plant fibers having many advantages such as low cost, low density, ecologically friendly, sustainability and biodegradability. The purpose of this study is to develop methods extracting bamboo fibers by using mechanical extracting, steam-explosion and alkaline-treatment techniques. How to control pH of wastewater yielding from alkali treatment is also one of the primary objectives of the present study. Mechanical properties (tensile strength, maximum strain at failure and Young's modulus) of extracted bamboo fibers were measured as well as their microstructure configuration. Interfacial strengths were tested between bamboo fibers and typical resins for polymer matrix composites such as unsaturated polyester and polypropylene. The variation of moisture absorption due to the difference of fiber extraction process was also determined. The experimental results proved that the alkaline-treatment was better than the steam-explosion method to extract bamboo fibers as well for mechanically extracted fibers as reinforcement for green composites, where two years or older raw bamboo (younger than four years old) was immersed in 1% NaOH solution for 10 hours. The present study also revealed that CO_2 gas was applicable to reduce pH from 10.8 to 7.0 within 110 minutes by aeration.

Keywords: Bamboo; Bamboo fibers; Wastewater; Alkaline treatment; Steam explosion; Mechanical extraction, CO₂ gas

1. Introduction

Natural plant fibers such as jute and kenaf are now widely used for automobiles as substitution for glass fibers from viewpoints of sustainability and easy wastability, namely burnability, biodegradability as well as cost. They

are applied for fabricating interior parts; rear partitions, ceilings, door panels, truck liners, parcel shelves, etc. (see Table 1) (C. Alaves et al., 2010; John Summerscales et al., 2010; Keiichiro Sano and Teruo Kimura, 2010, Kazuya Okubo et al, 2010). They are expected to be used more and more with an increase of automobile production. However, less potential exists to increase those fibers production on the earth since there is less land to cultivate for those natural plants in the tropical regions, for instance South East Asia. Cultivation of new farm lands will let disappear the natural habitat of all kind of creatures. Therefore, we have to find an alternative plant from which similar fibers are extracted. The plant must be abundant and grow naturally without changing lands. What is that? The answer is bamboo. Bamboo is not grass neither wood, while it has two of their characteristics. Bamboo, itself is very strong in its longitudinal direction due to strong fiber bundles penetration its body from the bottom to the top although its monofilaments, other word, pulp is shorter than ~2mm. If we could extract string-like fiber bundles, they are very useful as substitution of jute and kenaf fibers.

Bamboo fiber bundles (shortly, bamboo fibers) are attracting more and more attention from researchers, and often called 'natural glass fiber' (Kazuya Okubo et al., 2004). In the automotive industry, natural fibers reinforced polymer composites are generally fabricated by press-molding techniques with thermoplastics (mainly polypropylene, polylactic acid may be used), hand lay-up and resin transfer molding techniques with thermosetting plastics (primarily unsaturated polyester) (Ryoko Tokoro et al., 2008; F. P. La Mantia et al., 2011; Jae Kyoo Lim et al., 2005). These techniques reasonably demand flexible fibers, high aspect ratio of fibers and mechanical properties. Compare to other natural fibers, bamboo fibers (precisely fiber bundles) are often brittle due to their thicker diameter as well as chemical composition containing a high content of lignin and hemicelluloses. These compositions play an important role in fiber swelling and moisture absorption (G. Bogoeva-Gaceva et al., 2007; Maya Jacob John et al., 2008). For those reasons, one of the purposes of the present study is to find out how to effectively extract thin bamboo fiber bundles from raw bamboo to apply to the automotive inner parts. We conduct different techniques, namely, mechanical extraction, steam explosion and alkaline treatment to extract bamboo fiber bundles. Their mechanical properties are examined.

In comparison with the steam-explosion and mechanical-extraction techniques in Table 2, the alkaline-treatment technique predominantly has some advantages such as cheap equipment, a high respect ratio of fiber, relatively low energy consumption and easy control of fiber property. The control of fiber properties in the automotive industry plays an important role, whose parameters much depend on relationship between bamboo age and processing conditions such as alkali concentration, temperature and immersing time (Ryoko Tokoro et al., 2008; Abhijit P. Deshpande et al., 2000; Keisuke Wakasugi et al., 2010; Hongyan Chen et al., 2009; K. Murali Mohan Rao et al., 2007; Moe Moe Thwe et al., 2003; Nguyen Huy Tung et al., 2004). In addition, this technique is especially suitable to apply to rural areas in Asia where bamboo almost only grows in Forest Mountains with a difficult means of transport from harvest fields to factories of fiber extractions as well as place of product fabrications. However, the alkaline-treatment technique evidently exposes a drawback from wastewater yielding a high pH. This is a big problem when applying them in an industrial scale. Therefore, the second goal of this study is how to control the pH of wastewater yielding from the alkaline-treatment technique. There are dozens of ways to treat wastewater such as sulfuric acid, hydrochloric acid or nitric acid. However, these materials usually result in a poisonous environment, salt formation in water and are difficult in process control, maintenance of equipment and chemical cost. To avoid these problems, CO₂ gas was selected for the neutralization. This technique was able to utilize CO₂ emissions from industrial waste and decreasingly contribute to global warming. CO₂ gas offered the following advantages over mineral acids due to low toxicity, it required no special protection equipment to handle and could not reduce pH below 5 even if overdosing occurs. More over, CO₂ gas did not form residual anions such as sulfate, chloride as other acids. Its cost was the same as sulfuric acid and about half the price of hydrochloric acid, so it was totally applicable to replace the conventionally used acid (G. Montees-Henandez et al., 2007; R. Pérez-López et al., 2008; Luc Van Ginneken et al., 2004; A. Gangagni Rao et al., 2007; Ankur Gaur et al., 2009).

2. Materials and Experiments

2.1 Raw bamboo

Raw bamboo was harvested from Moso bamboo of Doshisha University, Kyoto, Japan, and they were extracted by methods as follows:

(1) The mechanical extraction method (Figure 1a)

Raw bamboo (2~8months) was first longitudinal cleaved into small slabs by the roller crusher. The pin-roller looser was used to extract small slabs into coarse fibers before removing fat by the boiler at 90° C for 10 hours. Then, coarse fibers were put into the dehydrator and finally dried in the Rotary dryer.

(2) The alkali-treated method (Figure 1b)

Nodes of raw bamboo (2~4 four years old) were first removed and remaining parts were cleaved in longitudinal direction to thin slabs with 20-30 cm in length and 2-3 mm in thickness by the slicer. These thin slabs were then immersed in NaOH solution at 70° C for 10 hours. The concentrations of NaOH were 1%, 2%, 3%. The 1% concentration of NaOH was selected in this study due to the higher mechanical properties of fibers with 1% than other fibers treated with 2% and 3% of NaOH (see Table (3). The roller looser was used to extract alkali treated slabs into small fibers. Finally, they were washed with fresh water to neutralize, and dried in the oven for 24 hours at 105° C.

(3) The steam-explosion method (Figure 1c)

Raw bamboo (2~4 four years old) was first cut into bamboo culms with 70-80cm in length by saw machine, and put into an autoclave with over-heated steam at 175° C and 0.7-0.8 MPa for 60 minutes. Then, the steam was suddenly released for 5 minutes and the cycles of sudden-steam release were continuously repeated for 9 times to assure the complete facture of cell walls. Finally, they were washed in hot water with addition of soap at 90-95°C for 15 minutes to remove ash and dried in the oven for 24 hours at 105° C.

2.2 Scanning electron microscopy observations

Microstructures of bamboo fiber were examined by scanning electron microscopy (SEM) using JSM-7001FD equipment. With the cross-sectional area at the imaging surface, the bamboo fiber was put in to a plastic mould with shape of cylinder of 3mm in thickness as shown in Fig. 2. Epoxy resin was used as matrix to hold bamboo fiber during the testing process. Epoxy resin was first poured into the mould in which the bamboo fiber was placed, and the mould was then put into heat-oven at 120° C for 4 hours. The fabricated specimens of SEM were polished by the Ecomet 3000 polisher until the specimen surfaces were clear and smooth, and finally cut into the appropriate shape for taking SEM by a slow-speed diamond cutter. Prior to SEM observation, all specimens including cross-section and longitudinal-section were coated with a thin layer of gold to avoid electrical charging.

2.3 Tensile test

Average diameters of bamboo fiber bundle were measured by microscope (the number of observations was about 50). Before the tensile test, bamboo fiber bundle was glued on a sheet of paper with length of 50mm, thickness of 0.5-1mm and gauge length of 25.4 mm as in Figure 3. Then the gauge positions of paper were cut after chucking it on testing machine to apply tensile load. The tensile test was measured under 1mm/min of cross-head speed.

2.4 Chemical composition

Untreated, alkaline treated and steam exploded fiber bundles were chemically analyzed for hemicelluloses, cellulose, lignin contents by using TAPPI standard T2220m-83 and T250um-85.

2.5 Moisture content

Moisture contents of bamboo fiber bundles were tested by Humidity Cabinet LHL-112 equipment using ASTM D2495-07 standard. Moisture content (M%) was calculated from equation following as:

$$M\% = \frac{Wt - Wo}{Wo} x100$$

Wt: weight of the specimen at the time t

Wo: weight of the initial specimen after drying

2.6 Interfacial shear strength

To evaluate interfacial behavior of unsaturated polyester and polypropylene resin with bamboo fiber bundle, pull-out test performed under 1mm/min of cross-head speed with specimens as shown in Figure 4. Interfacial shear strength was calculated by the following equation.

Tinterface
$$= \frac{P}{\pi dl}$$

P: pull-out load, N

d: diameter of bamboo fiber bundle, mm

1: length of embedded bamboo fiber bundle into resin, mm

2.7 Physic chemical characteristics of wastewater and neutralization apparatus

Wastewater in this study was collected from the alkaline-treatment process. The basic quality parameters such as pH, total dissolved solids (TDS), total solid (TS), total suspended solids (TSS), volatile suspended solids (VSS), volatile dissolved solids (VDS) were analyzed according to APHA standard (P.R Sreemahadevan Pillai, 2010). Table 4 resulted in the characteristics of wastewater.

The used apparatus in neutralization were described as shown in Figure 5. Alkaline wastewater was first put into a chamber with volume of 1000 ml, and then CO_2 gas was dispersed by a stirring system. Temperature of the chamber was controlled by an automatic heat system. pH value was continuously determined at a series time interval of 10 minutes until pH reached 7.0.

3. Results and Discussion

3.1 Microstructure configurations, chemical compositions and mechanical properties of bamboo fiber

Observations of SEM pictures in Figure 1a-c showed single fibers in the extracting methods were longitudinally aligned, and they were together bonded by lignin and hemicelluloses with the different bonding levels. In the case of untreated and alkaline treated fibers, the single-fiber bonding was strong by rich lignin areas. In contrast, the single-fiber bonding of steam exploded fibers relatively weakened. Most lignin content was extracted outside the fiber bundle, and significantly condensed on fiber surfaces under the repeating cycles of high temperature and pressure (see 2.1). Figure 1c also showed the formation of holes and round cracks inside fiber bundles (Maha M. Ibrahim et al., 2011). Determination of chemical compositions and mechanical properties of bamboo fibers as in Table 5 evidently exhibited relationship between chemical compositions (lignin and hemicelluloses content) and mechanical properties. The removal of lignin and hemicelluloses in fiber structure led to a decrease in their mechanical properties as well as in diameters. Table 5 also indicated that fiber bundles more softened with the alkaline-treatment and steam-explosion technique. Their Young's modulus values were respectively 26.1 ± 14.5 and 25.7 ± 14.0 GPa, and lower than the untreated fibers of 38.2 ± 16.0 GPa. These results were completely suitable to those of previous researchers, which asserted lignin and hemicelluloses playing a role as a matrix in natural fibers (G. Bogoeva-Gaceva et al., 2007; Maya Jacob John et al., 2008; Nguyen Huy Tung et al., 2004).

3.2 Moisture absorption

In fabrication of green-composite materials, moisture content plays an important role, which directly affects manufacturing process, product properties as well as their quality (G. Bogoeva-Gaceva et al., 2007; A. Athijayamani et al., 2009; F. Gouanvé et al., 2007). To investigate the characteristics of moisture absorptions in this study, bamboo fibers were determined at 20° C and 50%, 60%, 70%, 80% and 90% relative humidity. Bamboo fibers were weighted at a series of time intervals until moisture absorption reached its equilibrium level, and the value of moisture content increased in the square root of time (see Figure 6a-e). This completely corresponded to behavior of natural fibers as well as green-composite materials (A. Athijayamani et al., 2009; F. Gouanvé et al., 2007). In Figure 6a-e, the result showed that when bamboo fibers were put into a chamber with different relative humidity levels, their moisture contents together increased in elevation of relative humidity level. The curve of equilibrium was formed from moisture content and relative humidity with difference of fiber types as shown in Figure 6f. The result also presented that extracted fibers from the steam-explosion and alkaline-treatment technique offered lower moisture contents than untreated fibers at all relative humidity levels, and the difference of moisture content between steam exploded and alkaline treated fibers was not considerably. The result also revealed the relationship between lignin, hemicelluloses and moisture content (see Table 5 and Figure 6f).

3.3 Interfacial shear strength

Figure 7a-b shows the investigated relation between bamboo fibers and resin for interfacial shear strength (IFSS). The results revealed that the IFSS of alkaline treated fibers was higher than that of steam exploded and untreated fibers with unsaturated polyester and polypropylene. Observation of SEM imagine in Figure 1b indicated that the surface of treated fibers was smooth and clear and its interfacial bonding with respect to polypropylene and unsaturated polyester strengthened. In the case of untreated fibers in Figure 1a, its surface obtained a lot of hemicelluloses, lignin, soft cell, and some impureness which led to incompatibility between untreated fibers and resins. Hence, the interfacial bonding weakened. With steam exploded fibers under effect on high temperature and pressure, its surface was covered a considerable amount of lignin as shown in Figure 1c. That led to a decrease of the adhesion between steam explosion fibers and resins.

3.4 Effect of flow rate on pH of wastewater

Wastewater yielding from the alkaline-treatment technique is often a drawback to apply alkali treated fibers on an industrial scale. In the present study, CO_2 gas was particularly selected to control pH of wastewater. In Table 4, wastewater had a high pH 10.8. This was a difficult problem to discharge it from manufactory to lakes, rivers and namely public water. In order to overcome this problem, CO_2 gas was used as a neutralization agent. Its reactions with sodium hydroxide included steps from (1) to (5) as below:

$CO_2(g)$	+	H_2O	\leftrightarrow	CO ₂ (aq)	+	H_2O	(1)
CO ₂ (aq)	+	H_2O	\leftrightarrow	H_2CO_3			(2)
H_2CO_3	+	OH ⁻ (aq)	\rightarrow	HCO ₃ (aq)	+	H_2O	(3)
HCO ₃ ⁻ (aq)	+	OH	\rightarrow	$CO_3^{2-}(aq)$	+	H_2O	(4)
$CO_3^{2-}(aq)$	+	$2Na^+$	\rightarrow	Na ₂ CO ₃			(5)

 CO_2 gas was initially dissolved in solution and reacted with water to form H₂CO₃. Once equilibrium was established between CO₂ gas and H₂CO₃, H₂CO₃ further dissociate into HCO₃⁻ and CO₃⁻ ions. The buffer solution acted a mean of keeping pH at a nearly constant value due to existence HCO₃⁻. CO₃²⁻ ions were the dominant species occurring in solution and these ions reacted with Na⁺ ions to form Na₂CO₃. Figure 8 showed that CO₂ gas was put into a chamber with the stirring speed of 400 rpm and flow rate of 5ml/minutes at room temperature. At that time, pH of wastewater decreased from pH 10.8 to pH 7.0 for 130 minutes. Then CO₂ gas was continuously put into the chamber, and pH did not much decrease from pH 7.0 to pH 6.48 for 130 minutes. This could be due to the wastewater that was completely saturated by CO₂ gas. Figure 8 also shows the comparison with various flow rates of 3, 7 and 9 ml/minutes for 260, 125 and 110 minutes. They all reached pH 7.0. In the case of the flow rate of 3 ml/minutes, pH slowly decreased in a long time. In the other hand, with the flow rate of 7 and 9 ml/minutes, the reaction of CO₂ gas with wastewater quickly happened for a short time, and consumptions of CO₂ gas was 650 ml.

3.5 Effect of stirring speed and temperature on pH

Stirring speed was one of the important parameters, which directly affected the pH value of wastewater. With a suitable stirring speed, CO_2 gas was quickly dispersed into wastewater with an optimal content.

Figure 9a indicated that with a stirring speed of 500 rpm the reaction time to obtain a pH of 7.0 was 110 minutes. This value was better when compared with the 300 and 400 rpm stirring speeds. This was due to increase of dissolution of CO_2 gas in wastewater when the stirring speed changed. However, observation of stirring speeds of 500 and 600 rpm showed that these curves nearly overlapped, and that their values of pH reached saturation after 220 minutes. Figure 9b also presented that there was no effect of temperature on pH of wastewater, and most of the curves nearly overlapped at room temperature, 30, 50 and 70^oC.

4. Conclusions

In this study, the steam-explosion, mechanical-extraction and alkaline-treatment technique were applied to extract bamboo fiber bundles from raw bamboo with different age, and the effect of CO_2 gas on the pH value of alkaline wastewater was also determined.

Some conclusions obtained from this study are:

1. Microstructures of bamboo fiber were determined by using scanning electron microscope. The results showed the single fibers were longitudinally aligned in the flat-wise direction. They were together bonded by hemicelluloses and lignin with different bonding levels.

2. Mechanical properties of fiber bundle and their interfacial shear strength (IFSS) with typical resins such as unsaturated polyester and polypropylene are related to lignin and hemicelluloses content in chemical composition. The experimental results also proved that the IFSS of alkaline treated fibers were better than that of steam exploded and untreated fibers.

3. Moisture absorptions of bamboo fibers at 50, 60, 70, 80 and 90% relative humidity at 20° C was also determined. The results showed that the difference of moisture content of the alkaline treated and the steam exploded fibers was much lower than that of the untreated fibers at all relative humidity levels.

4. Wastewater of the alkaline-treatment technique was treated by CO_2 gas, and the ratio of wastewater to CO_2 gas was 1000 ml to 650 ml. pH of wastewater was decreased from 10.8 to 7 within 110 minutes.

References

A. Athijayamani, M. Thiruchitrambalam, U. Natarajan, & B. Pazhanivel. (2009). Effect of moisture absorption on the mechanical properties of randomly oriented natural fibers/polyester hybrid composite. *Materials Science and Engineering A*, *517*, 344-353. http://dx.doi.org/10.1016/j.msea.2009.04.027

A. Gangagni Rao, T. Sasi Kanth Reddy, S. Surya Prakash, J. Vanajakshi, Johny Joseph, & P. N. Sarma. (2007). pH regulation of alkaline wastewater with carbon dioxide: A case study of treatment of brewery wastewater in UASB reactor coupled with absorber. *Bioresource Technology*, *98*, 2131-2136. http://dx.doi.org/10.1016/j.biortech.2006.08.011

Abhijit P. Deshpande, M. BhaskarRao, & C. Lakshmana Rao. (2000). Extraction of bamboo fibers and their use as reinforcement in polymeric composites. *Journal of Applied Polymer Science*, *76*, 83-92. http://dx.doi.org/10.1002/(SICI)1097-4628(20000404)76:1<83::AID-APP11>3.0.CO;2-L.

Ankur Gaur, Jin-Won Park, Jung-Hwa Jang, Sanjeev Maken, JooJae Lee, & Ho-Jun Song. (2009). Characteristics of alkaline wastewater neutralization for CO₂ capture landfill gas(LFG). *Energy Fuels, 23,* 5467-5473. http://dx.doi.org/10.1021/ef900615h

C. Alves, P. M. C. Ferrão, A. J. Silva, L. G. Reis, M. Freitas, L. B. Rodrigues, & D. E. Alves. (2010). Ecodesign of automotive components making use of natural jute fiber composites. *Journal of Cleaner Production*, *18*, 313-327. http://dx.doi.org/ 10.1016/j.jclepro.2009.10.022

F. Gouanvé, S. Marais, A. Bessadok, D. Langevin, & M. Métayer. (2007). Kinetics of water sorption in flax and PET fibers. *European Polymer Journal, 43*, 586-598. http://dx.doi.org/ 10.1016/j.eurpolymj.2006.10.023

F. P. La Mantia, & M. Morreale. (2011). Green composites: A brief review. *Composites: Part A: applied science and manufacturing*, 42, 579-588. http://dx.doi.org/10.1016/j.compositesa.2011.01.017

G. Bogoeva-Gaceva, M. Avella, M. Malinconico, A. Buzarovska, A. Grozdanov, G. Gentile., & M. E. Errico. (2007). Natural fiber eco-composites. *Polymer Composites*, 98-107. http://dx.doi.org/10.1002/pc.20270

G. Montees-Hernandez, F. Renard, N. Geoffroy, L. Charlet, & J. Pironon. (2007). Calcite precipitation from CO₂-H₂O-Ca(OH)₂ slurry under high pressure of CO₂. *Journal of Crystal Growth, 308,* 228-236. http://dx.doi.org/10.1016/j.jcrysgro.2007.08.005

Hongyan Chen, Menghe Miao, & Xin Ding. (2009). Influence of moisture absorption on the interfacial strength of bamboo/vinyl ester composites. *Composite: Part A, 40, 2013-2019.* http://dx.doi.org/10.1016/j.compositesa.2009.09.003

Jae Kyoo Lim, Jun Yong Choi, & Hyo Jin Kim. (2005). Effects of matrix on mechanical property test bamboo fiber composite material. *Key engineering Materials*, 297-300, 1529-1533. http://dx.doi.org/10.4028/www.scientific.net/KEM.297-300.1529

Jonh Summerscales, Nilmini Dissanayake, Amandeep Virk, & Waynehall. (2010). A review of bast fibres and their composites. *Composite: Part A, 41,* 1336-1344. http://dx.doi.org/10.1016/j.compositesa.2010.05.020

K. Murali Mohan Rao, & K. Mohana Rao. (2007). Extraction and tensile properties of natural fibers: Vakka, date and bamboo. *Composite Structures*, 77, 288-295. http://dx.doi.org/10.1016/j.compstruct.2005.07.023

Kazuya Okubo, Toru Fujii, & Xiaochen Fu. (2010). An application of bamboo fibers in Japan. Effect of needle punching and spring-back molding on stiffness of stampable bamboo fiber sheet processed with pressed mat. *VCCS-5*, November, Hanoi, Vietnam

Kazuya Okubo, Toru Fujii, & Yuzo Yamamoto. (2004). Development of bamboo-based polymer composites and their mechanical properties. *Composite: Part A, 35,* 377-383. http://dx.doi.org/10.1016/j.compositesa.2003.09.017

Keiichiro Sano, & Teruo Kimura. (2010). Trend of natural fiber reinforced plastic for automobile industry in Germany and Japan. Proceeding of The Sixth international Workshop on Green Composites (IWGC-6)-109-112

Keisuke Wakasugi, Kazuya Okubo, & Toru Fujii. (2010). Improvement of strength of bamboo fiber paper by addition of MFC (Micro Fibrillated Cellulose). *Proceeding of The Sixth International Workshop on Green Composites (IWGC-6)*, 178-181.

Luc Van Ginneken, Veronika Dutré, Walter Adriansens, & Herman Weyten. (2004). Effect of liquid and supercritical carbon dioxide treatments on the leaching performance of a cement-stabilised waste form. *Journal of Supercritical Fluid*, *30*, 175-188. http://dx.doi.org/10.1016/j.supflu.2003.07.004

Maha M. Ibrahim, Waleed K. El-Zawawy, Yasser R. Abdel-Fattah, Nadia A. Soliman, & Foster A. Agblevor. (2011). Comparison of alkaline pulping with steam explosion for glucose production from rice straw, *Carbohydrate Polymers*, *83*, 720-726. http://dx.doi.org/10.1016/j.carbpol.2010.08.046

Maya Jacob John, & Rajesh D. Anandjiwala. (2008). Recent developments in chemical modification and characterization of natural fiber-reinforced composites. *Polymer composites*, 187-207. doi: 10.1002/pc.20461, http://dx.doi.org/10.1002/pc.20461

Moe Moe Thwe, & Kin Liao. (2003). Durability of bamboo-glass fiber reinforced polymer matrix hybrid composites. *Composite Science and Technology*, *63*, 375-387. http://dx.doi.org/10.1016/S0266-3538(02)00225-7

Nguyen Huy Tung, Hiroshi Yamamoto, Takashi Matsuoka, & Toru Fujii. (2004). Effect of surface treatment on interfacial strength bamboo fiber and PP resin. *JSME International Journal Series A*, 47, 561-565. http://dx.doi.org/10.1299/jsmea

P. R Sreemahadevan Pillai. (2010). New age international publishers, A comprehensive laboratory manual for environmental science and engineering.

R. Pérez-López, G.Montes-Hernandez, J.M. Nieto, F. Renard, & L.Charlet. (2008). Carbonation of alkaline paper mill waste to reduce CO₂ greenhouse gas emssision into the atmosphere. *Applied Geochemistry*, 23, 2292-2300. http://dx.doi.org/10.1016/S0048-9697(99)00025-X

RyokoTokoro, Duc Minh Vu, Kazuaya Okubo, Tatsuya Takanaka, Toru Fujii, & Takayasu Fujiura. (2008). How to improve mechanical properties of polylactic acid with bamboo fibers. *Journal of Material Science, 43*, 775-787. http://dx.doi.org/10.1007/s10853-007-1994-y

Manufacturer	Fiber	Resin	Parts		
Toyota	Kenaf, sisal, ramie	Door trim, inner board, roof lining, cover board of the spare tire			
Mercedes	Flax, hemp	Polypropylene	Inner board of instrumental panel, door, pillar inners, head liner and trunk components.		
Mitsubishi	Bamboo	Inner board of trunk door panel.			
BMW	Flax	polypropylene	Inner board of door panel.		

 Table 1. Automakers using green-composites (Keiichiro Sano & Teruo Kimura, 2010)

Method	Cost of equipment	Energy consumption	Installing area	Environmental impact	Aspect ratio of fiber	Control of fiber property	Bamboo age and species
Mechanical extraction	Reasonably expensive	Relatively low	Small	No	Low	Acceptable	Young (2~8months)
Alkali treatment	Cheap as long as excluding waste water treatment	Medium	Huge	Yes	High	Good	Thin F.: ~12months Medium F.: 1~2 years. Coarse F.: 3years.~
Steam explosion	Expensive	High	Large	No	High	Good	Preferable: younger than 2 years. Thick F.: ~5 years.

Table 2. Comparison of different extraction methods of bamboo fiber

Table 3. Mechanical properties of bamboo fiber bundle with respect to NaOH concentration

	Fiber diameter,	Tensile strength,	Young's modulus,	Maximum strain at	
	(µm)	(MPa)	(GPa)	failure, (%)	
NaOH 1%	230±180	395±155	26.1±145	2.82±1.3	
NaOH 2%	218±175	360 ± 170	25.2±13.3	2.62±1.2	
NaOH 3%	209±180	324±170	23.7±12.5	2.54±1.2	

Table 4. Characteristics of wastewater

Parameters	PH	TS, mg/l	TSS, mg/l	TDS, mg/l	VSS, mg/l	VDS, mg/l
Value	10.8	44.040	41.380	2660	840	22.140

Table 5. Chemical compositions and mechanical properties of bamboo fiber bundle

	Fiber diameter, (µm)	Tensile strength, (MPa)	Young's modulus, (GPa)	Maximum strain at failure, (%)	Lignin, (%)	Hemi celluloses, (%)	Cellulose, (%)
Untreated	262 ± 160	420 ± 170	38.2±16.0	9.8±2.5	24.4	25.2	48.5
Alkaline treated	230 ± 180	395±155	26.1±14.5	2.82 ± 1.3	22.6	16.4	48.8
Steam exploded	195 <u>+</u> 150	308±185	25.7±14.0	2.51 ± 1.2	23.3	7.4	47.4

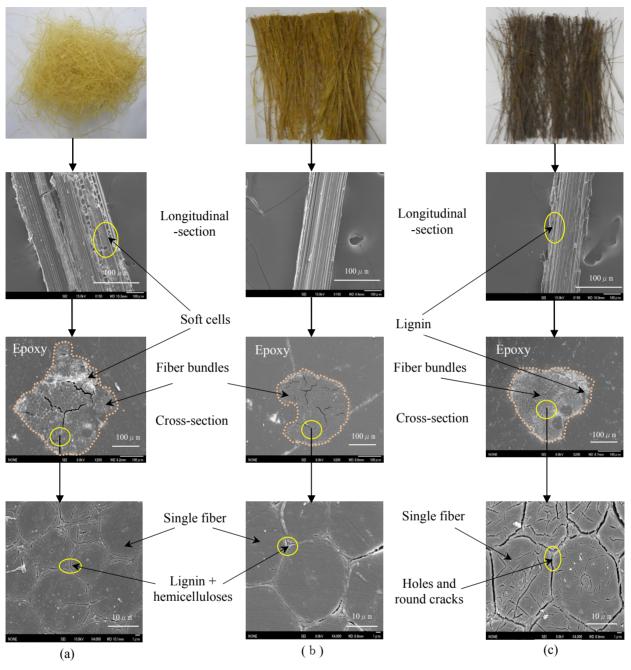


Figure 1. SEM imagines for side-view and cross-section surfaces of bamboo fiber (a) Untreated fiber, (b) Alkaline treated fiber, (c) Steam exploded fiber

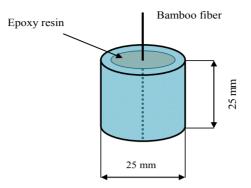


Figure 2. Preparation of cross-section surface

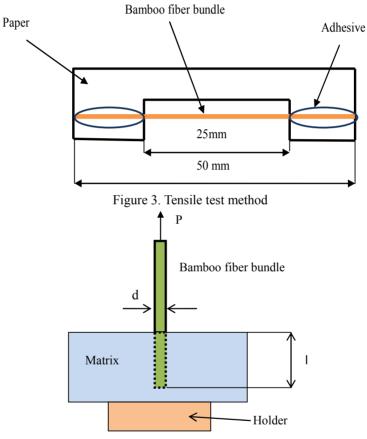


Figure 4. Pull-out test method of bamboo fiber bundle

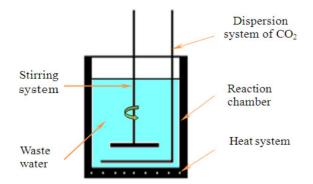


Figure 5. Reaction apparatus of CO₂ gas with alkaline wastewater

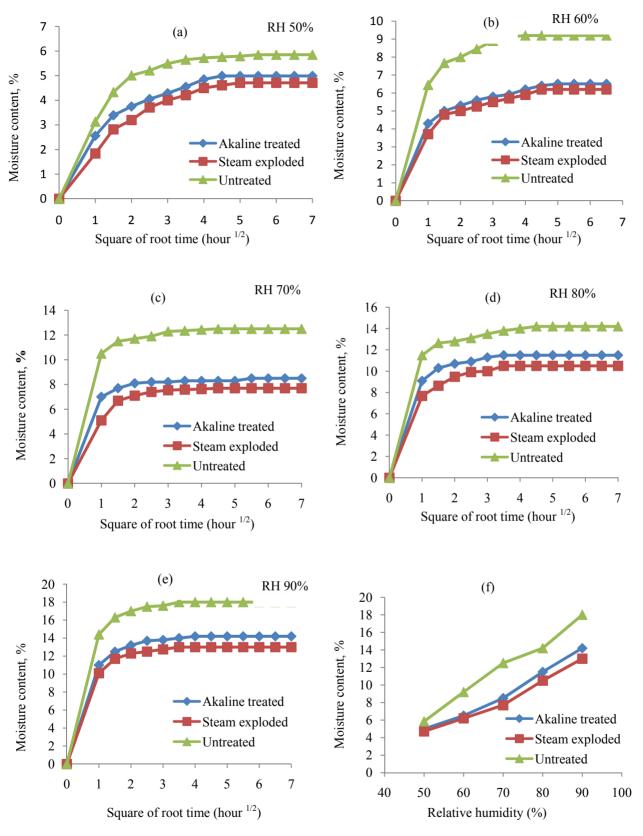


Figure 6. Moisture content curves of bamboo fiber at different relative humidity: 50%(a), 60%(b), 70%(c), 80%(d), 90%(e) and relationship between equilibrium moisture content and relative humidity(f)

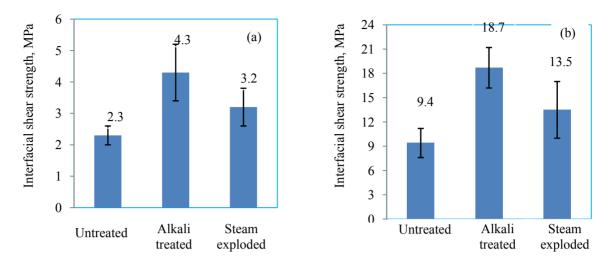


Figure 7. Interfacial shear strength of bamboo fiber bundles and polypropylene (a), unsaturated polyester resin (b)

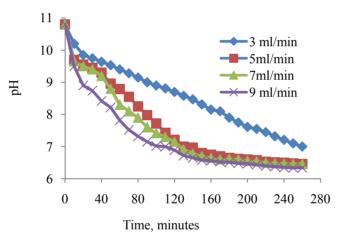


Figure 8. Change in pH with various flow rate

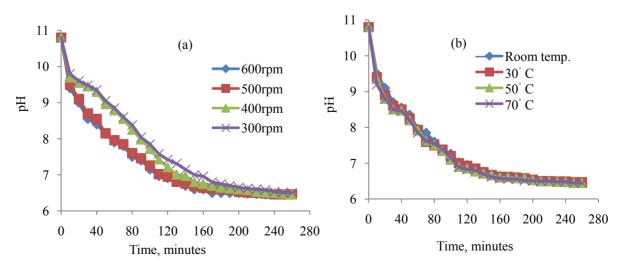


Figure 9. Change in pH with various stirring speed (a) and reaction temperature (b)